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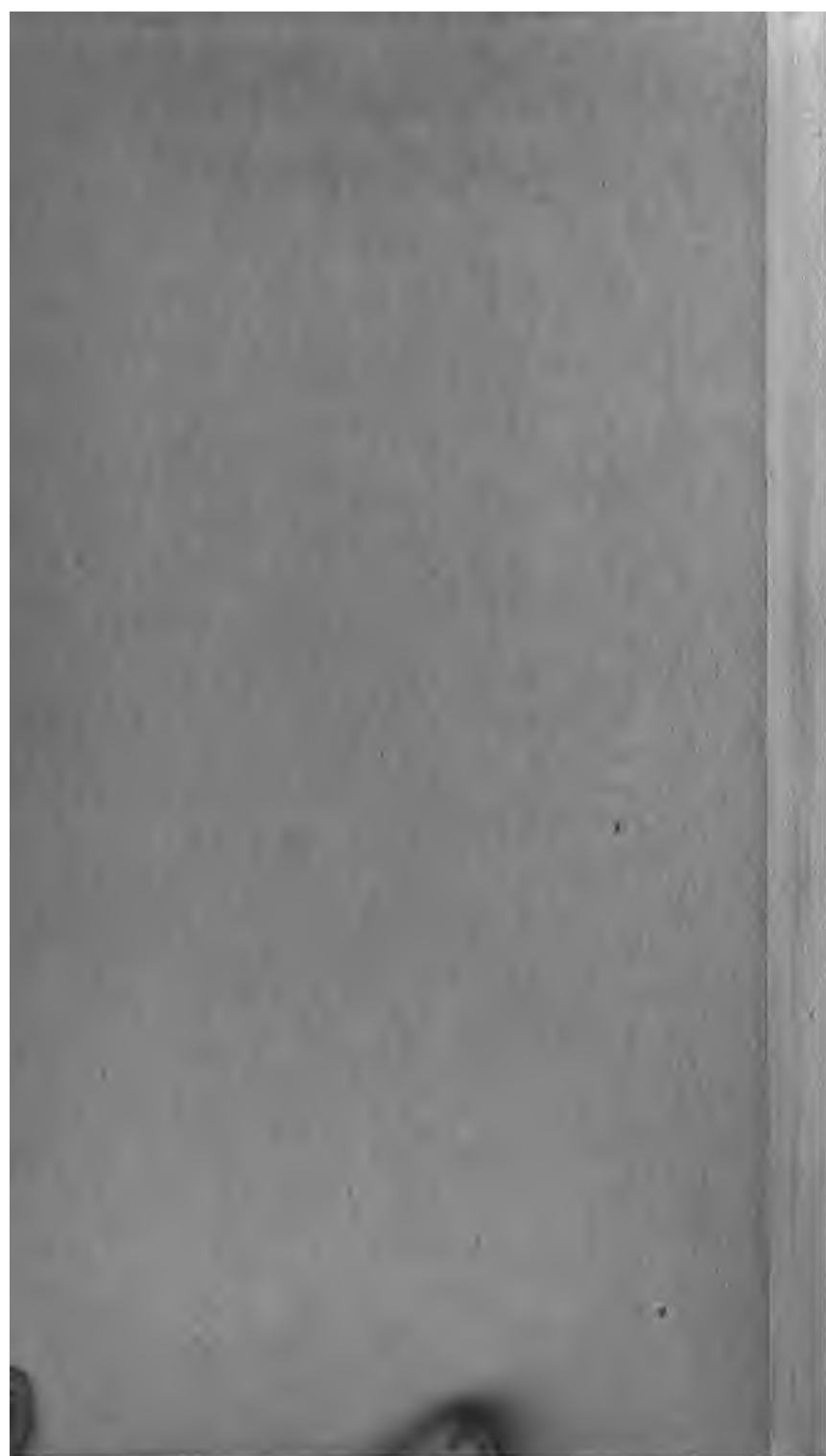
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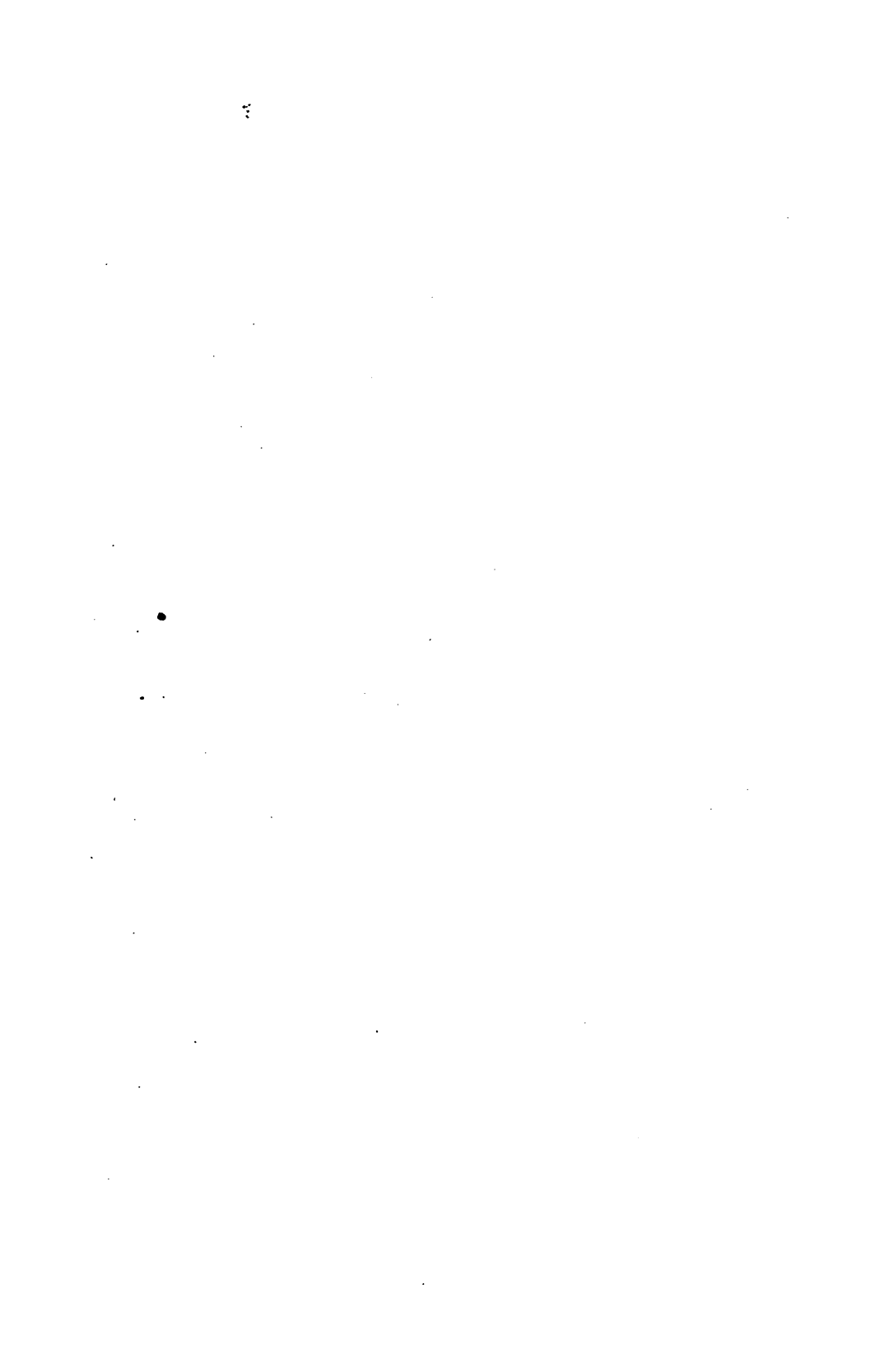
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ELECTRIC LAMPS

— AND —

ELECTRIC LIGHTING.

A Course of Four Lectures on Electric Illumination

DELIVERED AT THE

Royal Institution of Great Britain.

BY

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PREFACE.

THE following pages are a transcript, with some additions, of a course of Four Lectures on "Electric Illumination," delivered by the Author to afternoon audiences at the Royal Institution, London. They are, however, in no sense presented as an adequate treatment of a subject which has long since outgrown the limits of a lecture, or even of a course of lectures.

The original aim was merely to offer to a general audience such non-technical explanations of the physical effects and problems concerned in the modern applications of electricity for illuminating purposes as might serve to further an intelligent interest in the subject, and perhaps pave the way for a more serious study of it. The growth of electric lighting is assisted by the diffusion of all accurate (even if simple) information, on

the mode of production and employment of electric currents for illuminating and other purposes.

Many of those who were present at the Lectures expressed a desire to be again able to refer to the descriptions and illustrations then given, and at their request, and in the hope that they may be of use to others; the lecture notes have been revised and published. The aim throughout has been rather to deal with principles than with details, and to give such general and guiding explanations of terms and processes as are essential for a grasp of the outlines of the subject. For this purpose simple drawings have been given of as many of the experiments and apparatus used as possible. For permission to reproduce a set of photographs of views of the interior of rooms artistically illuminated with incandescent lamps, and which are to be found in the second Lecture, the author is indebted to Mr. A. F. Davies, under whose guidance these particular instances of electric lighting were carried out.

J. A. F.

University College, London,
October, 1894.





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ERRATA.

The reader is requested to make the following corrections:—

Page 15, line 10 from top—

instead of “movement of fixed plates,”

read “movement of suspended plates.”

Page 69, line 4 from bottom—

instead of “If a small piece of wire,”

read “If a small piece of iron wire.”

Page 206, line 8 from bottom—

instead of “90lb. on the square inch,”

read “60lb. on the square inch.”

Page 207, bottom line—

instead of “one of these larger machines,”

read “one of these smaller machines.”



LECTURE I.

ELECTRIC LIGHTING in Great Britain.—A Glance Backward over Fourteen Years.—Present Condition of Electric Lighting.—An Electric Current.—Its Chief Properties.—Heating Power.—Electrical Resistance.—Names of Electrical Units.—Chemical Power of an Electric Current.—Hydraulic Analogies.—Electric Pressure.—Fall of Electric Pressure down a Conductor.—Ohm's Law.—Joule's Law.—Units of Work and Power.—The Watt as a Unit of Power.—Incandescence of a Platinum Wire.—Spectroscopic Examination of a Heated Wire.—Visible and Invisible Radiation.—Luminous Efficiency.—Radiation from Bodies at Various Temperatures.—Efficiency of Various Sources of Light.—The Glow Lamp and Arc Lamp as Illuminants.—Colours and Wave-Lengths of Rays of Light.—Similar and Dissimilar Sources of Light.—Colour-distinguishing Power.—Causes of Colour.—Comparison of Brightness and Colour.—Principles of Photometry.—Limitations due to the Eye.—Luminosity and Candle-power.—Standards of Light.—Standards of Illumination.—The Candle-foot.—Comparison of Sun-light and Moonlight.—Comparison of Lights.—Ritchie's Wedge.—Rumford and Bunsen Photometers.—Comparison of Lights of Different Colours.—Spectro-photometers.—Results of Investigations.



HOEVER undertakes to describe the remarkable progress during the last two decades of the art of electric illumination must certainly direct attention to three important dates which, in England at any rate, formed turning points in the history of this application of scientific knowledge.

In the year 1879 the Government of our country had its attention directed for the first time to electric lighting as a possible subject of legislation, and referred the whole matter to a Select Committee of the House of Commons, of which Lord Playfair was appointed Chairman. This Committee met, and took voluminous evidence from numerous and

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various experts, but the broad conclusion reached in the report which it finally presented was generally the expression of opinion that there was no reasonable scientific ground at that date for supposing that domestic electric lighting had obtained a sufficient footing to entitle it to be described as a practical success, and that, therefore, there seemed no useful result to be attained by interfering at once, by special legislation, with electric lighting. Passing over a gap of three years, we find in the year 1882 the whole aspect of affairs entirely changed by the completed invention of the electric glow lamp. At the beginning of that year the first Crystal Palace Electrical Exhibition enabled the public to properly appreciate the extent to which the then perfected electric incandescence lamp had revolutionised artificial lighting, and also the charm and beauty of that illuminant. Schemes were rapidly put forward for the supply of electric current from public generating stations for the purpose of electric lighting, and the enthusiasm which was thus awakened in the public mind led inventors and promoters to prognosticate a very speedy and entire revolution in the art of illumination. As a result, immediate legislative action was taken to control the public supply of electric current. In 1882 a Bill was introduced into the House of Commons entitled "An Act for Facilitating Electric Lighting." We need not pause to enumerate the various causes which interfered with the immediate fulfilment of the sanguine expectations which were formed in 1882 as to the immediate future of electric lighting. Practical difficulties presented themselves the moment that many of the immature schemes which were then launched were attempted to be put into practice. The dynamo machine in some of its forms had hardly emerged from the condition of being a large laboratory or workshop instrument, and mechanical engineers had not yet brought to bear upon it that knowledge which subsequently enabled them to convert it into a most efficient and trustworthy machine for generating

electric current for public electric supply. Countless details remained to be perfected in the many arrangements for distributing, using, and measuring electric current so produced. From the commercial point of view much information had to be slowly accumulated before even approximately correct opinions could be formed as to the revenue to be gained from the sale of electric energy for domestic purposes, and there was at that date but little information available for enabling an accurate forecast to be made concerning the probable average annual consumption of electrical energy by incandescent lamps when used instead of gas jets in different classes of buildings for illuminating purposes. There is no doubt, however, that the Act of 1882, though much abused at the time, performed the important function of preventing the survival of the unfit.

Between 1882 and 1888 exceedingly important improvements (some of which it will be necessary later on to examine) were made, and in 1888 the time seemed ripe for a fresh forward movement. This was effected by the pressure brought upon the Legislature to repeal one of the clauses in the Act of 1882, by which revision much more favourable conditions were created for inviting the support of capital ; and as soon as the Electric Lighting Amendment Act of 1888 was an accomplished fact, a very important inquiry was held by the Board of Trade in May, 1889, in the Westminster Town Hall, London, under the chairmanship of Major Marindin. At this inquiry, which lasted for eighteen days, the whole subject of electric lighting by public supply, especially with reference to the needs of London, was carefully debated by many of the leading scientific and legal experts, and, as a result, the Metropolis was divided up into certain areas of electric supply, and conditions were laid down under which the distribution of current might be undertaken either by Public Companies or by the Local Authorities. Up to the date of that inquiry the total amount of electric lighting in London

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or in the Provinces had been, comparatively speaking, very small. From and after that date it has advanced by leaps and bounds. The progress made in the use of the incandescent lamp as a means of artificial illumination is shown by the following figures, and graphically indicated in Fig. 1.

At the end of 1890 there were probably in use about 180,000 incandescent lamps in London, but at the end of 1892 some 500,000 were being employed, and at the end of 1893

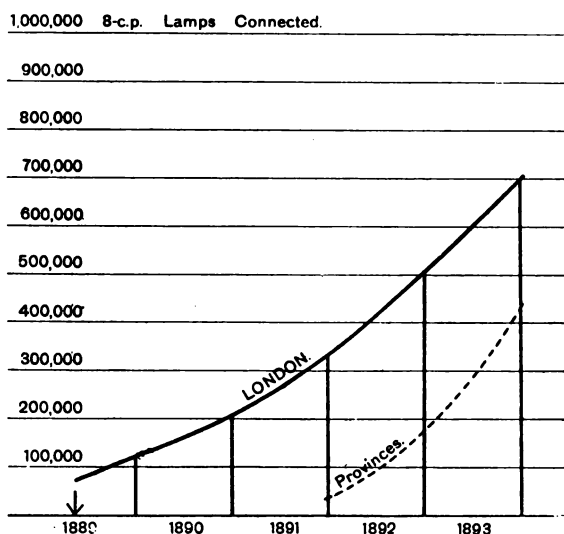


FIG. 1.—Diagram showing the Progress of Electric Lighting in London and the Provinces in Five Years. The altitude of the vertical black lines represents to scale the growth, from year to year, in the total number of lamps supplied.

700,000. In the provinces the grand totals at the end of 1892 and 1893 were probably 147,000 and 425,000. Hence, while in 1888 the total number of incandescent lamps in use in the United Kingdom probably hardly exceeded 100,000, even if so many, we find that at the end of 1893 there was a grand total of rather more than 1,125 000 electric glow lamps in use in the United Kingdom. There are at the

present time (1894) some thirteen or fourteen Companies and Corporations supplying electric current for lighting purposes in London, and these have a total length of more than 250 miles of copper mains laid down under the streets. In the Provinces there are between 60 and 70 towns in which a similar public electric supply is given, either by the Local Authority or by a Public Company, and these "undertakers," as they are called in the Act, have a total length of probably more than 400 miles of mains in use at present. Some idea may be formed of the extent to which electric lighting has proceeded in the United Kingdom in the five years which have elapsed since the Board of Trade inquiry was held at the Westminster Town Hall in 1889, by picturing to ourselves an underground electric main or pair of conductors of copper stretching across Great Britain from Penzance to Perth, with an 8-candle-power incandescent lamp placed every yard along this distance. This would roughly represent the total usage of electric lamps and length of electric mains at the present date (1894).

The 700,000 incandescent electric lamps in London have probably already begun to make their presence felt by the Gas Companies. If these 700,000 glow lamps were replaced by gas jets of equal illuminating power, their annual consumption of gas might at least amount to about 1,000 million cubic feet. The average rate of increase of the metropolitan gas consumption between 1886 and 1891 was 947 million cubic feet, the increase for 1891 alone being 1,313 millions. In 1892, however, the year when the London Electric Lighting Companies got fairly to work, this increase fell to $35\frac{1}{2}$ millions; and in 1893 the consumption was less by 1,007 million cubic feet. The year 1893 was unusually bright and sunny in the summer and free from fog in winter in London. Hence this cause amongst others may have operated to arrest the growth of gas, but it is clear that a stage has now been reached in which the

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older illuminant will begin to feel the competition of the younger. An industry which has progressed with such rapid strides, and which in the short space of half a decade has made electric lighting in large towns one of the necessary luxuries of life, is not likely to be arrested in its present stage; and as it has, in its various aspects, not only much scientific and artistic, but also considerable practical interest for every householder, your attention in this and the three succeeding lectures will be directed to the elucidation of facts which ought to be known by every user of the electric light, and the knowledge of which will enable him to understand something of the principles which underlie the art of electric illumination, and to comprehend as well the aid which it can render, when properly applied, to beautify and please.

It will be necessary to open the whole of our discussion by some simple illustrations of the meaning of fundamental terms. Every science as well as every art has its necessary technical terms, and even if these words at first sound strangely, they are not therefore necessarily difficult to understand. We are all familiar with the fact that electric illumination depends upon the utilisation of something which we call the electric current. Little by little scientific research may open up a pathway towards a fuller understanding of the true nature of an electric current, but at the present moment all that we are able to say is that we know of what it can do, how it is produced, and the manner in which it can properly be measured. Two principal facts connected with it are, that when a conductor, such as a metallic wire or a carbon filament, or any other material which is capable of being employed as a conductor, is traversed by an electric current, heat is generated in the conductor, and the space round the conductor becomes capable of influencing a magnetic needle. These facts can be simply illustrated by passing an electric current through an iron wire thus (Fig. 2). You will notice that as the current is gradually-

increased the iron wire is brought up from a condition in which it is only slightly warm to one in which it becomes visibly red hot in the dark, and finally brilliantly incandescent. At the same time if I explore the region round about the wire with a suspended compass needle, we find that at every point in the neighbourhood of the wire the magnetic needle places itself, or tries to place itself, in a position perpendicular to the wire. This fact, of capital importance, was discovered by

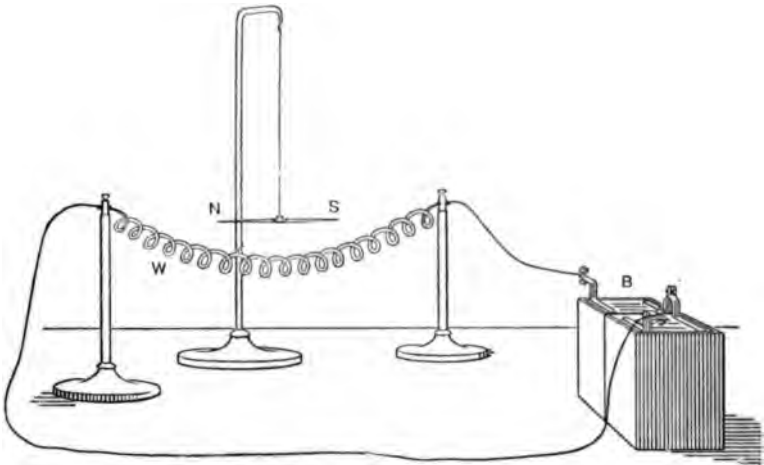


FIG. 2.—An iron wire **W** is rendered incandescent by an electric current sent through it from a battery **B**. The magnetic needle **NS** held near it sets itself across the wire.

H. C. Oersted in 1820, and in the Latin memoir in which he describes this epoch-making discovery he employs the following striking phrase to express the behaviour of a magnetic needle to the wire conveying the current. He says, "The electric conflict performs circles round the wire;" and that which he called the electric conflict round the wire, we now in more modern language call the *magnetic field* embracing the conductor. We shall return in a later lecture to this last fact.

Meanwhile I wish at present to fasten your attention on the heating qualities of an electric current, and the laws of that heat production and radiation. The same electric current produces heat at very different rates in different conductors, and the quality of a body in virtue of which the electric current produces heat in passing through it is called its electric resistance. If the same electric current is passed through conducting wires of similar dimensions, but of different materials, it produces in them different quantities of heat in the same time. Before you (Fig. 3) is a chain composed of spirals of iron and copper wire. These wires are each of the same length and of the same diameter. Sending through this com-

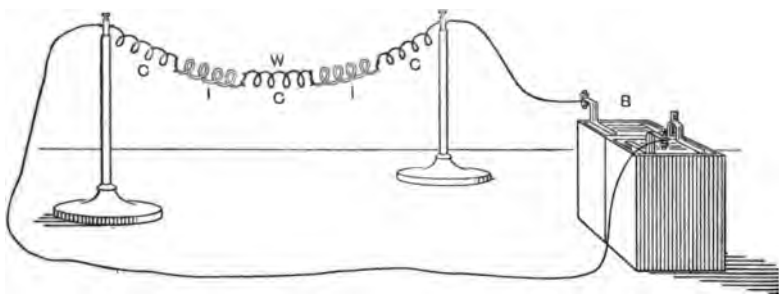


FIG. 3.—A chain *W* composed of alternate spirals of copper *C* and iron *I* is traversed by an electric current sent through it by a battery *B*. The iron links become red hot, the copper links only slightly warm.

pound chain an electric current, we notice that the iron wire links are very soon brought up to a bright red heat, whilst the copper links, though slightly warm, are not visibly hot. We have, therefore, before us an illustration of the fact that a current heats the conductor, but that each conductor has a specific quality called its electrical resistance, in virtue of which the same strength of current produces heat in it at a rate depending on the nature of the material. Other things being equal, the bodies which are most heated are said to have the highest resistance.

It is now necessary to notice the units in which these two quantities, namely, electric current and electric resistance, are measured. For the sake of distinction, units of electric quantities are named after distinguished men. We follow a similar custom in some respects in common life, as when we speak of a "Gladstone" bag or a "Hansom" cab, and abbreviate these terms into a gladstone and a hansom. Primarily the distinctive words here used are the names of persons, but by application and abbreviation they become the names of things. An electric current is measured in terms of a unit current which is called an *ampere*, and electrical resistance is measured in terms of a unit which is called an *ohm*, these being respectively named after two great investi-

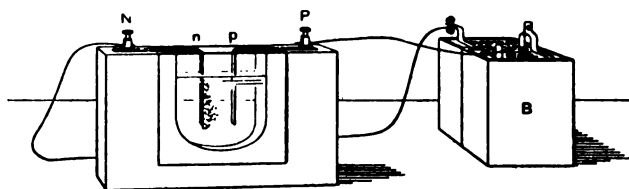


FIG. 4.—Glass lantern trough containing two lead plates *n p*, and a solution of sugar of lead. When a current from a battery *B* is sent through the cell it deposits the lead in tufts on the negative plate *n*.

gators, André Marie Ampère and Georg Simon Ohm. In order to understand the mode in which an electric current can thus be defined, we must direct attention to another property of electric currents, namely, their power of decomposing solutions of metallic salts. You are all familiar with the substance which is called sugar of lead, or, in chemical language, acetate of lead. Placing in a small glass trough a solution of acetate of lead and two lead plates (*see* Fig. 4), I place the cell in the electric lantern and project the image upon the screen. If an electric current is passed through the solution from one lead plate to the other it decomposes

the solution of acetate of lead, extricating from the solution molecules of lead and depositing them on one of the lead plates, and you see the tufted crystals of lead being built up in frond-like form on the negative pole in the cell. We might employ, in preference to a solution of acetate of lead, a solution of nitrate of silver, which is the basis of most marking inks, and the same effect would be seen. It was definitely proved by Faraday that we might define the strength of an electric current by the amount of metal which it extricates from the solution of a metallic salt in one second, minute, or hour. The Board of Trade Committee on Electrical Standards have now given a definition of what is to be understood by an electric current of *one ampere* in the following terms: An electric current of one ampere is a current which will in one hour extricate from a solution of nitrate of silver 4.025 grammes of silver.* Otherwise we might put it in this manner: A current of electricity is said to have a strength of one ampere if, when passed through a solution of nitrate of silver, it decomposes it and deposits on the negative plate one ounce of silver in very nearly seven hours. We are acquainted in the laboratory with currents of electricity so small that they would take 100,000 years of continuous action to deposit one ounce of silver, and we are familiar in electric lighting practice with currents great enough to deposit one hundred-weight of silver in thirty minutes. The simple experiment just shown is the basis of the whole art of electro-plating. Hence, when we speak later of a current of one ampere, or ten amperes you will be able to realise in thought precisely what such a current is able to achieve in chemical decomposition. It may be convenient at this stage to bring to your notice the fact that an 8 candle-power incandescent lamp working at 100 volts usually takes a current of about one-third of an ampere, a current which would deposit by electro-plating action one ounce of silver in about twenty-one hours.

* 28.3495 grammes = 1 ounce avoirdupois.

We pass next to consider another important matter, viz., that of electric pressure or potential; and we shall be helped in grasping this idea by considering the corresponding conception in the case of the flow of fluids. When a fluid such as water flows along a pipe it does so in virtue of the fact that there is a difference of pressure between different points in the pipe, and the water flows in the pipe from the place where the pressure is greatest to the place where the pressure is least. On the table before you is a horizontal pipe (Fig. 5) which is connected with a cistern of water, and which delivers

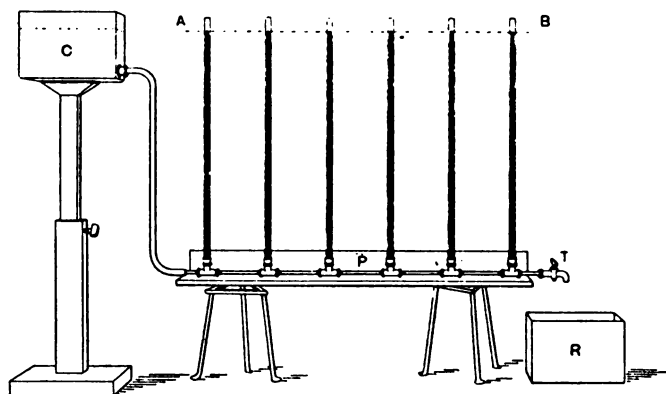


FIG. 5.—Horizontal pipe *P*, having six vertical gauge tubes attached to it. The pipe *P* is in connection with a water cistern *C*, and when the tap *T* is shut the water stands up at the same level, *A B*, in all the tubes.

water to another receptacle at a lower level. In that pipe are placed a number of vertical glass tubes to enable us to measure the pressure in the pipe at any instant. The pressure at the foot of each gauge glass is exactly measured by the *head* or elevation of the water in the vertical gauge glass, and at the present moment, when the outlet from the horizontal pipe is closed, you will notice that the water in all the gauge glasses stands up to the same height as the water in the cistern. In other words, the pressure in the pipe is everywhere the same.

Opening the outlet tap we allow the water to flow out from the pipe, and you will then observe that the water sinks (see Fig. 6) in each gauge glass, and, so far from being now uniform in height, there is seen to be a regular fall in pressure along the pipe, the gauge glass nearest the cistern showing the greatest pressure, the next one less, the next one less still, and so on, the pressure in the horizontal pipe gradually diminishing as we proceed along towards the tap by which the water is flowing out. This fall in pressure along the pipe takes place in every gas and water pipe, and is called the

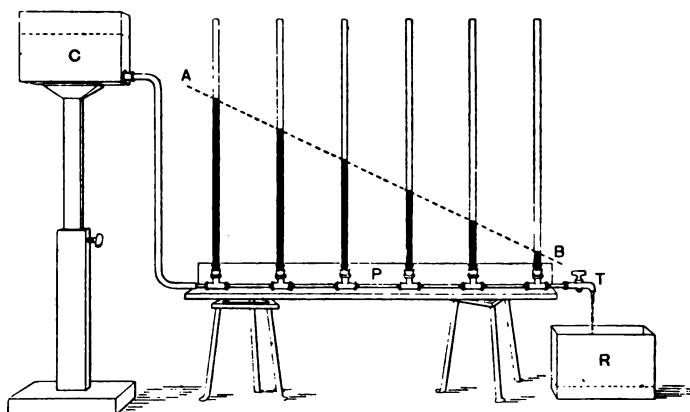


FIG. 6.—Horizontal pipe **P**, through which water is flowing from a cistern **C** to a reservoir **R**. When the tap **T** is open the water stands at gradually decreasing heights in the pressure tubes. The dotted line **A B** shows the hydraulic gradient.

hydraulic gradient in the pipe. The flow of water takes place in virtue of this gradient of pressure. It will be next necessary to explain to you that there is an exactly similar phenomenon in the case of an electric current in a wire, and that there is a quantity which we may call the *electric pressure*, which diminishes in amount as we proceed along the wire when the current is flowing in it. In order to understand the manner in which this electric pressure can be

measured, a few preliminary experiments will be essential. Every body which is charged with electricity has, in virtue of that charge, a certain electrical potential, or pressure, as it is called, and electricity always tends to flow from places of

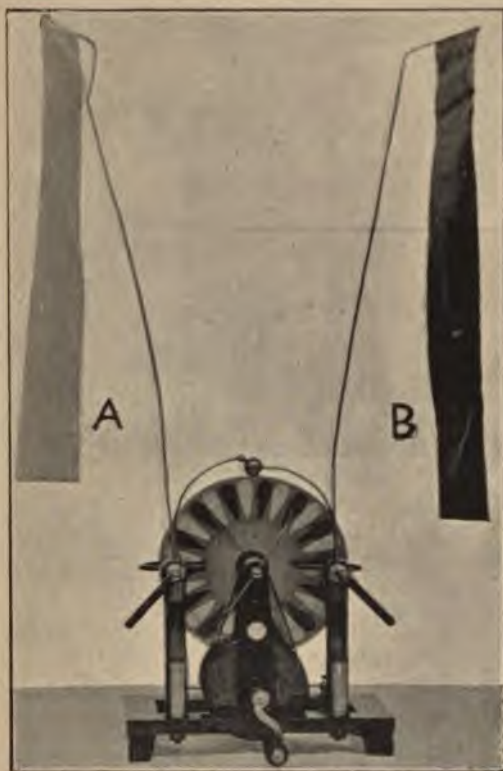


FIG. 7.--A small Wimshurst electrical machine, having two strips of paper A and B attached to its terminals. When the machine is worked the paper strips are drawn together.

higher to lower potential, just as water or other fluids tend to flow from places of greater to less pressure. When two bodies are at different electric pressures, or potentials, it is found that there is an attraction or stress existing between

them, and a tendency for them to move, if possible, nearer together. If I attach to the terminals of a small electrical machine two paper strips, and then charge those paper strips to different electric pressures, we find the strips

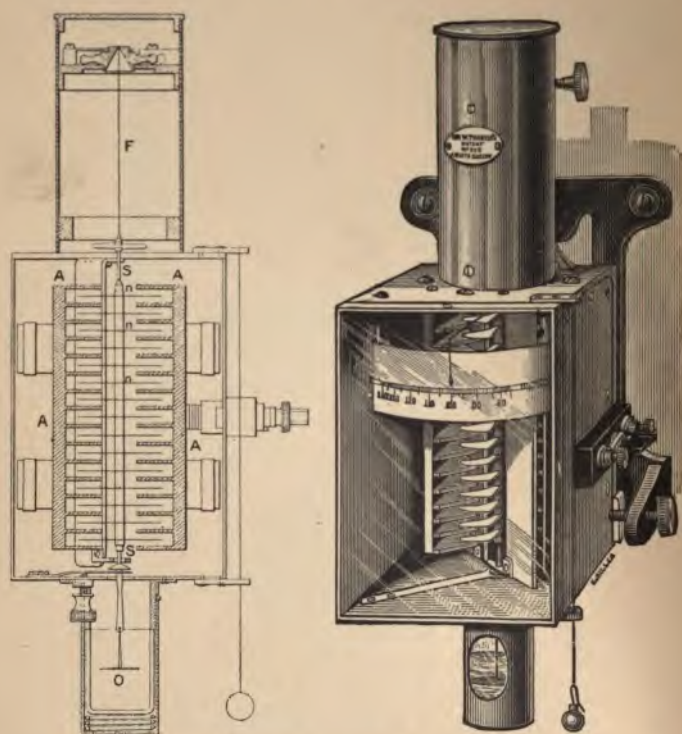


FIG. 8.—Lord Kelvin's Multicellular Electrostatic Voltmeter.
The fixed plates or cells are marked **A** in the sectional drawing. The movable plates **n** are attached to a suspended axis **S**.

are drawn together (*see* Fig. 7). The difference of pressure between these two bodies can be exactly measured by the mechanical force with which they attract one another, or by the force required to keep them apart by a certain distance. This fact is taken advantage of to construct many instruments

which are called electric pressure-measuring instruments, or voltmeters. One of the most valuable of these is the electrostatic voltmeter, invented by Lord Kelvin. It consists (*see* Fig. 8) of a series of fixed plates, which are called cells, and suspended between these are a number of movable plates, all attached to a common axis, this axis being suspended by a very fine wire. The suspended plates are so arranged that, when they are at a different electric pressure or potential from the fixed plates, they are attracted in between them, and the movement of the fixed plates is resisted by the torsional force

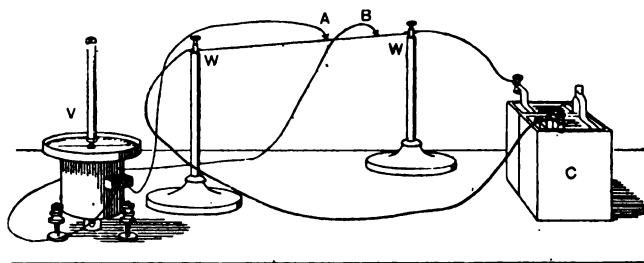


FIG. 9.—A battery *C*, which in the actual experiment consists of 50 cells, sends a current through a wire *W W*. By means of a pair of contact pieces *A B*, the terminals of an electrostatic voltmeter *V* are connected to various points on the wire *W W*, and the fall in electric pressure along it is thus measured.

of the suspending wire. The extent to which they have so moved can be measured by an indicating needle fastened to the movable plates. If the fixed and movable plates in this instrument are brought to different electric pressures, or different electric potentials, they will be attracted towards one another, and we then have an instrument which can be converted by proper graduation into an electric pressure-measuring instrument. Furnished with such an appliance we can now explore the change in pressure down a wire through which the electric current is flowing. Through this manganese-steel wire we are now passing a current of electricity. One terminal of the voltmeter, namely, that connected to the fixed

plates, is kept connected to one end of that manganese-steel wire, whilst the wire connected to the movable plates of the voltmeter can be slid along the manganese-steel wire to different points (*see* Fig. 9). You will see from the indications of the voltmeter that as I slide the movable contact along the conductor conveying the current the indications of the voltmeter increase, and it is possible thus to show that there is a difference of electric pressure between two points on the wire conveying the current, precisely as there is a difference of fluid pressure between two points in a pipe along which water or gas is flowing. The water flows in the pipe from places where water pressure is greatest to places where water pressure is least. The electric current, whatever it may really be, likewise flows from places where the electric pressure is greatest to places where it is least. A unit of electric pressure has been selected which has been called a *volt*, after Volta, who, in 1801, gave us the first galvanic battery. It may be remarked in passing that the pressures at which it is most usual to work incandescent lamps are either 50 or 100 volts between the terminals of the lamp, and that the pressure between the terminals of a single galvanic cell, such as is used for working electric bells, is from 1 to $1\frac{1}{2}$ volts.

We are then able to connect these two ideas of electric current and electric pressure when measured in the units above defined, and to give a definition of the unit in which electric resistance is measured. If we assume that a wire is taken through which a uniform current of one ampere is passed always in one direction, and two points on this wire are found such that the difference of electric pressure between those points is one volt, then such wire is said to have a resistance of one ohm between those chosen points. The resistance of any conductor can be measured by comparing it by certain methods with that of a conductor whose resistance is one ohm, and the electrical resistance of

any wire or conductor can thereby be expressed in units, each of which is called one ohm. The great service which Dr. G. S. Ohm rendered in 1827 to electrical science was that he gave the first clear definition of the manner in which electric current, electric pressure, and electric resistance are related to one another. This is now embodied in a statement which is called Ohm's law, and which is stated as follows:—The strength of the electric current which flows in any conductor when that current has a uniform flow in one direction being measured in amperes, and the difference of the electric pressures between the ends of that conductor being given in volts, the electrical resistance of that conductor in ohms is obtained by dividing the last number by the first. Ohm's law is not only a definition of the mode of measuring electrical resistance, but is also a statement of a physical law. It has been experimentally proved that the resistance of a conductor, as measured above, does not depend upon the value of the current, and is the same for large currents as for small ones if a correction is applied for the change of temperature of the conductor which is produced by the current. The resistance of a conductor is generally affected considerably by change of temperature; for some bodies, such as pure metals, it is increased, and for other bodies, such as carbon and certain metallic alloys, it is decreased.

Apart, however, from changes of temperature, it is an experimental fact that the strength of the current produced in any given conductor is exactly proportional to the fall in electric pressure down the conductor, and the numerical ratio of the numbers representing the fall of pressure in volts and the strength of the current in amperes is the value of the electrical resistance of that conductor in ohms. In order that you may see the application of this in electric lighting, let us consider the above in connection with an incandescent electric lamp. Such a glow lamp, as at present constructed, consists, as

we shall see in the next lecture, of a fine carbon thread or filament, which, as usually made, is traversed by a current of about two-thirds of an ampere, when that conductor is one which is suitable for a 16-candle-power lamp worked at the usual pressure of 100 volts. The electric supply companies bring into our houses two wires, between which they are constantly engaged in keeping an electric pressure difference of 100 volts or 110 volts, or thereabouts. If, therefore, the terminals of a lamp are connected to these two supply wires, the ends of the carbon filament are exposed to an electric pressure of 100 volts. The electric resistance of that carbon, when incandescent, is therefore expressed in ohms by dividing the number expressing the pressure difference in volts by the number defining the current in amperes; hence it is the quotient of 100 by two-thirds, or 150 ohms.

Another fundamental law in connection with the flow of an electric current in conductors was enunciated by Mr. Joule in 1841, and is called Joule's law. It is thus stated: If a current flows through an electric conductor, the heat produced in that conductor per second is proportional to the product of the square of the current strength as measured in amperes and the resistance of the conductor measured in ohms. Joule deduced this law from elaborately careful experiments made on the quantity of heat produced in a certain wire when traversed by an electric current, that wire being immersed in water. His experimental procedure was as follows:—He immersed a wire, formed into a spiral, in a vessel of water so protected as not to be able to lose heat from the outside. He then passed measured currents of electricity through the spiral, and observed with delicate thermometers the rise of temperature of the water in a stated time. The whole of the energy which is thus being spent in the wire is converted into heat, and that heat is employed in raising the temperature of the water. If by suitable means we prevent the loss of heat

from the containing vessel, or otherwise take it into account, and if we try this experiment with currents of two different strengths, say of one ampere and two amperes, it will be found that, if the resistance of the conductor remains the same, the heat generated in a given time by the current of two amperes will be four times as great as the heat generated in the same time by a current of one ampere, and in like manner a current of three amperes would generate nine times as much heat as a current of one ampere, always provided that the resistance of the wire is not sensibly changed when the current is altered. A little consideration of the law of Joule and the law of Ohm, when taken together, will show you that, since the total amount of heat produced per second by a current of a given magnitude is proportional to the products of the numbers representing the resistance of the circuit in ohms, and the square of the strength of the current measured in amperes flowing through it; and, since the product of the value of the resistance of the circuit in ohms and the current strength in amperes is numerically equal to the difference of pressure between the two ends of the conductor, it follows that the total rate at which energy is being expended in any conductor to produce heat when a current of electricity is flowing through it is measured by the numerical product of the strength of that current in amperes, and the pressure difference between the terminals of that conductor measured in volts. If we apply this rule to the case of an electric lamp we find that, in order to measure the total rate at which energy is being transformed into heat and light in an incandescent electric lamp, we have to measure, in the first place, the current passing through it in amperes and the pressure difference between the terminals of the lamp in volts, and the product then gives us, in certain units, which are called *watts*, the rate at which energy is being dissipated or converted into heat in the carbon filament. Thus, for example, if a lamp which takes two-

thirds of an ampere is placed upon a circuit having a pressure difference of 100 volts, the product of 100 and two-thirds being 66, the lamp would be taking 66 *watts*, and this is the measure of the rate at which energy is being supplied to the lamp, and converted by it into light and heat.

It is important that you should possess a very clear conception of the exact meaning attached to the *watt* as a unit of power. If a weight of one pound is lifted one foot high, the amount of exertion or work required to raise this weight is, in engineering language, called *one foot pound* of work. If 550 pounds, or nearly one-quarter of a ton, are so lifted one foot high, the work done is called 550 foot-pounds. Imagine this last exertion made in one second, and repeated every second; the *rate at which work is being done*, or exertion made, will be equal to that which is called *one horse-power*. An ordinary man might for some time keep on lifting 55 pounds, say the weight of a good-sized full portmanteau, one foot high per second, and he would then be doing one-tenth of a horse-power. The work done in lifting one pound about nine inches high, or more nearly 0.7373 of a foot, has been selected as a unit of work, and is called one *joule*, and, if this work is repeated every second, this *rate of doing work*, or making exertion, is called *one watt*. It is necessary to notice carefully that a *watt* is a unit rate of doing work and not a unit of work. If the pound weight is lifted against gravity nine inches high slowly or quickly the work done is the same, viz., one *joule*. If it is lifted nine inches high in one second this rate of doing work is called *one watt*. If it is lifted nine inches high in half a second, the rate of doing work is doubled, and is then *two watts*. Hence it will be seen that one watt is a unit of power, which is the $\frac{1}{746}$ th part of a horse-power. The reason for choosing such a small unit is that it gives convenient numbers in which to measure the rate at which work is done in such energy

transforming agents as incandescent lamps. When a larger unit is necessary, it is usual to employ the *kilowatt*, which is equal to 1,000 watts, as a unit of power, and it is obvious that a kilowatt is equal to about one and one-third of a horse-power. Accordingly, such a lamp as we have above assumed dissipates energy at a rate which would require one horse-power to maintain eleven or twelve lamps, and the rate at which physical energy has to be supplied to a 16-candle-power glow lamp to keep it at full candle-power is almost as great as the maximum rate at which a strong man can do work.

Having prepared the way for further explanations by these elementary definitions, we must now proceed to examine more carefully what happens when an electric current is sent through a conductor, and the temperature of that conductor is allowed to rise. Coming back to our initial experiment with the iron wire heated by an electric current, we must note that there are three ways in which the energy supplied to that wire is being dissipated. In the first place, it is dissipated by contact with the cold air around it. The air molecules are continually beating against the hot wire, coming up to it cool, taking energy from it to heat themselves, and going away warm, and this process of carrying off of heat by the air molecules is called *convection*. In the next place, the wire loses heat by conduction out at the ends; the warm wire is held in cool metallic supports, which are so large and such good conductors that they are not sensibly heated by the current. Hence, part of the heat of the wire is carried away by them, and if you examine the red-hot wire you will find that it is cooler at the ends than it is in the middle, being not quite so brilliantly incandescent close up to the clamps as it is in the centre. Then, furthermore, the wire is losing energy by a process called radiation. It is imparting energy to the *ether*, and sending out waves into this ether which represent an energy transformation, some of these waves being of such a

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character that they can affect the retina of the eye, and constitute what we call light; but by far the larger quantity of that wave energy is not capable of affecting the retina of the eye, and is called the non-luminous radiation. The luminous radiation, as we shall see later, in this case is only about one or two per cent. of the total radiation. Let us follow, then, the changes that take place as any conductor, say a wire or thin carbon rod, is gradually heated to incandescence. Prof. Draper, as far back as 1847, carried out such a series of experiments with a platinum wire, which he heated up gradually by an electric current, measuring at each stage the total amount of energy which was thrown out from the wire in the form of heat and the amount of energy thrown out from the wire in the form of luminous radiation or light. Before you is a table of Draper's results. His figures are in certain arbitrary units.

Table of Draper's Experimental Results (in 1847) on the Incandescence of Platinum Wire.

Temp'rature of the Wire in Centigrade degrees.	The Heat given out by the Wire.	The Light radiated by the Wire.	Remarks.
525°C	...	practically nil	Just visible in the dark.
527°C	0.87 units	"
590°C	1.10 "	"	Spectrum visible to line E.
653°C	1.50 "	"	" " " F.
718°C	1.80 "	"	" " " F-G.
782°C	2.80 "	"	" " " G.
910°C	3.70 "	"	Full " spectrum visible.
1038°C	6.80 "	0.39 units
1100°C	8.60 "	0.62 "
1166°C	10.0 "	1.73 "	Spectrum to H.
1230°C	12.5 "	2.92 "
1293°C	15.5 "	4.40 "
1367°C	...	7.24 "
1421°C	...	12.34 "

The above amounts of heat and light are given in arbitrary units. The letters E, F, G, refer to the fixed lines in the solar spectrum.

Let us, however, follow the whole progress of the phenomena experimentally by the employment of an incandescent lamp. Before me on the table is a carbon glow lamp, having a straight filament of carbon as the conductor to be rendered incandescent. By means of a lens we can project an image of the glowing carbon upon a screen, and by means of a prism we can expand that linear optical image into a prismatic spectrum, or rainbow strip, and notice the gradual changes which occur as the carbon is heated up from its lowest temperature to the highest incandescence it will safely bear. By passing a carefully graduated current through the lamp, we find out that at a certain point the carbon just begins to be visible in the dark. It has sometimes been inferred, as a deduction from Draper's experiments, and was, indeed, stated by him, that all bodies begin to be visibly red hot in the dark at the same temperature, namely, at about 525°C . This, however, is certainly not the fact. It has been shown by Weber, Bottomley, and others, as the result of careful experiments on carbon filaments and platinum wires, that bodies with a black, sooty surface have to be brought up to a higher temperature than bodies with a bright metallic surface before they begin to be visible in complete darkness. The crucial experiment has been made by J. T. Bottomley of taking two platinum wires, one of which is made to have a dull, sooty surface by coating it with the finest possible coating of lampblack, the other being highly polished. If these are placed in closed glass tubes from which the air is exhausted, and electric currents passed through the wires so as to heat them, it will be found, on carefully increasing the current and examining the wires in complete darkness, that the wire which has a bright metallic surface becomes visible in the dark at a lower temperature than the wire which has a dull, sooty surface. In Mr. Bottomley's experiments the temperature of the two wires was obtained from their electrical resistance; this last having been carefully measured at various tempera-

tures previously. This and other experiments show that it is not strictly true that all bodies become luminous in the dark at the same temperature. But approximately we may say that most bodies, such as metals and carbons, when heated to 600° Centigrade, begin to give out radiation which is capable of affecting the eye as dull red light. Returning to our lamp, let us gradually increase the current through the carbon conductor of this Bernstein lamp, whilst at the same time we project the image of the straight carbon upon the screen, and examine it with a prism. As the current is gradually increased the carbon gives off radiation which is first entirely non-luminous, and which, though not capable of affecting our eyes, can be detected by a very delicate thermometer or other suitable instrumental means. As the temperature of the wire is increased to a higher point, radiation makes its appearance which is capable of affecting the retina of the eye. It is sometimes stated that the first colour which makes its appearance is red, but careful observation shows that the light which impresses the eye first, and which most easily stimulates the retina, is a *greyish-green*, which occupies in the spectrum the position of maximum brightness. If the temperature of the conductor is still further increased, we see that the prism shows us that red and yellow rays have made their appearance also, and a short spectrum is projected upon the screen in which the red, yellow, and green rays are clearly visible. Increasing still more the temperature of the conductor by passing a stronger current through it, we notice that the spectrum lengthens, greenish-blue, blue, and violet rays successively making their appearance, and are added to the others, whilst the previously existing green and red rays, and especially the green and yellow, are much strengthened. The spectrum, therefore, exhibits a growth, which growth consists in the addition of more and more refrangible rays, or rays towards the violet end of the spectrum, whilst at the same time a gradual increase in the intensity or luminosity of all

the rays takes place, so that, when finally we have the complete spectrum exhibited on the screen, the conductor is found on examination to be in a state of brilliant incandescence, throwing out white light. We can, therefore, by the assistance of the prism, watch the gradual progress in the emission of radiation from the carbon conductor, which is capable of affecting the eyes; but experiments can be made which, at the same time, show that non-luminous or invisible radiation is being sent out from the conductor, and that this, at all stages of the increase in luminosity of the visible spectrum, is also increased in a certain proportion. Broadly speaking, therefore, as we increase the temperature of any body, it first begins by emitting rays which are not capable of affecting the eye, but finally, as the temperature is carried up to a higher and higher point, it emits successively rays which are capable of affecting the optic nerve; and in the end, when a temperature of 1,600 or 1,700°C. is reached, we have a total radiation from the conductor, which affects the eye as white light. The annexed table shows approximately the character of the light emitted from bodies when raised to different temperatures:—

Radiation from Bodies at different Temperatures.

	Centigrade.
Bodies begin to be just visible in the dark at about ...	500°
Dull red heat at	700°
Dull cherry-red heat at	800°
Full cherry-red at	900°
Melting point of silver at	945°
Clear red heat at	1,000°
Melting point of gold at... ..	1,045°
White cast-iron melts at... ..	1,040°
Orange-red heat at	1,100°
Bright orange heat at	1,200°
White heat at	1,300°
Steel melts at	1,300°
Bright white heat at	1,400°
Dazzling white heat at	1,500°
Palladium melts at	1,500°
Wrought-iron melts at	1,600°
Platinum melts at	1,775°
Iridium melts at	1,950°

Returning again to our incandescent wire, it must be noted that, of the total radiation sent out by such a hot body, only a small fraction of that radiant energy is capable of affecting the eye, and making its impression upon us as light. By far the larger proportion of radiation from most incandescent bodies is non-luminous radiation, and makes no impression at all upon the organ of sight. The proportion of luminous to total radiation from various sources has been measured by different observers, with the following results :—

Proportion of Luminous to Non-Luminous Radiation in Various Sources of Light.

Source.	Luminous Radiation.	Non-Luminous Radiation.
Red-hot wire	practically nil.	100 per cent.
Hydrogen flame	ditto.	100 "
Oil flame	3 per cent.	97 "
Gas flame	4 "	96 "
White-hot wire	4·5 "	95·4 "
Electric glow lamp	3 to 7 "	95 "
Arc lamp	5 to 15 "	90 "
Sunlight	34 "	66 "

The above table shows us, then, the percentage of luminous to non-luminous radiation in these various sources of light. The *luminous efficiency* of any illuminant is defined as the fraction, expressed as a percentage, which the luminous radiation is of the total radiation. We may also express, in the unit previously defined, called the watt, the rate at which energy is being expended in any illuminant to produce an illuminating power equal to that of one candle, and coupling this with the known luminous efficiency of the illuminant, we obtain two numbers which precisely express the energy consumption of the agent, and that which may be called its efficiency as a translating device for converting energy from one form, whether electrical or chemical, into another form—namely, eye-affecting wave-motion in the ether. Below are

collected together these numbers for various sources of light :—

Efficiency of Various Sources of Light.

Source of Light.	Total Consumption of Energy in Watts required to produce a light of one Candle.	Ratio of Luminous to total Radiation, or Luminous Efficiency.
Candle	86 watts	2 to 3 per cent.
Oil lamp	57 ..	3 ..
Petroleum lamp	42·8 ..	3 ..
Argand gas lamp	68·8 ..	4 ..
Electric glow lamp	3½ ..	3 to 7 ..
Electric arc	0·8 ..	5 to 15 ..
Magnesium wire	15 ..
Electric discharge in rarefied gases	33 ..

One great problem awaiting solution in the future is the discovery of a source of artificial illumination which is, relatively speaking, much more efficient than those which are at present available; that is to say, of a method which will convert a greater proportion of the energy which is transformed into radiation strictly limited to that which can affect the eye; whether that energy be electrical energy in the form of an electric current, or whether it be chemical potential energy associated with materials to be combined. It appears probable that the solution of this problem rests in the development of phosphorescence into a practical method for yielding light. By the employment of apparatus of extraordinary delicacy, Prof. S. P. Langley and Mr. Very, in America, have succeeded in determining the luminous efficiency of the light emitted by the Cuban fire-fly. They find that in this natural light the whole energy of radiation is comprised within the limits of the visible spectrum, and, what is more important, chiefly within the limits of the green and greenish-yellow rays, which are especially important for the purposes of vision. These experiments show that the light emitted by this insect is,

indeed, light without heat, and that its luminous efficiency is not far below 100 per cent. In our artificial production of light we have much lee-way to make up before we can rival the efficiency of the glow-worm and fire-fly.

Before, however, we can discuss this matter further, it will be necessary to turn our attention a little more at length to the subject of photometry, or the comparison of different sources of illumination in regard to their light-giving qualities. This is a part of our subject which, we may say at once, is in a much less satisfactory condition, as regards exact measurement, than the other practical electrical measurements to which reference has been made. The various light-giving bodies which we know and use, such as the sun, candle, gas lamp, incandescent electric lamp, electric arc lamp, &c., are bodies which are at very different temperatures, and they emit light, therefore, of very different composition. Every ray of light which has a pure and simple ray is characterised by two qualities—first, its *wave length*, which determines the *colour* impression it makes on the organ of vision; and, secondly, the *amplitude* of that wave motion, which determines its luminous *intensity*.

It may fairly be assumed that my audience is familiar with the fact that all optical research has indicated the fact that what we call light is a disturbance of some kind taking place in a universal medium called the ether. We do not know precisely the nature of that medium or that disturbance, but crucial experiment shows that, when a ray of light is passing through space, some change is being repeated very rapidly at every point in its path, and that this disturbance travels out from any point with a velocity which is, approximately, 186,000 miles a second; that is, it moves nearly one foot in the thousand-millionth part of a second. This disturbance is a wave motion—that is to say, at points in

spectra that they become exactly equal in brightness or luminosity for any particular ray, say the yellow—that is to say, we can so weaken the light of the arc lamp that its spectrum in the yellow is exactly equal in brightness to the spectrum of the incandescent lamp in the yellow. When this is done, we find that the arc lamp spectrum is very much brighter in the violet, green, and blue than that of the incandescent lamp spectrum, but that it is not so bright in the red. It is clear therefore, that these two lights are dissimilar in quality, and before we can speak of the comparison of two lights, we must understand distinctly what we are going to compare.

Broadly speaking, the two great purposes for which we require artificial light are to discriminate form and to discriminate colour difference. With regard to the discrimination of form, it is not necessary that the light which is employed should possess rays of more than one colour or wave length. Such a light is called mono-chromatic. If I illuminate this room with light which is wholly yellow or wholly red, by burning metallic sodium in the flame of a spirit lamp, or, in a similar manner, by burning some nitrate of strontium, which is the basis of red fire, we find that we are able to discriminate the form of surrounding objects, but that their normal colour differences have vanished. I need not spend much time in reminding you that what we call the colour of different objects is only a physical difference in their surface or substance which enables them to select certain rays out of a compound bundle of rays falling upon them for reflection, and to absorb the rest. Thus, a ripe cherry is red because, out of the whole collection of rays of light falling upon it, it sends back to our eyes by preference a large proportion of the red, yellow, and green rays, but absorbs some violet and blue; and the

leaves of the cherry tree are green for the reason that they absorb a large proportion of rays other than green, but reflect more copiously these last. This selective reflection is, however, due to the absorption or diminution in intensity of certain rays in the compound light falling on the body, and which, passing *through* the surface of the reflecting body, is internally reflected, and then is returned back through the surface, having in the double transit suffered a deprivation or weakening of certain constituent rays. Hence, the colour of a body is dependent upon the character of the radiation which falls upon it, and if we illuminate a group of coloured bodies, first by light which is purely red, and then by light which is purely green, we find that in these two cases the apparent colour of the bodies is totally different.

These facts we are able to illustrate before you in a very simple manner by throwing some red light, produced by passing the light from the electric lantern through a pure ruby-red glass, upon a group of coloured ribbons or pieces of coloured paper. We note that the red paper still looks red in the red light, but that the green and blue paper or ribbons look almost perfectly black. On the other hand, when plunged into the green light, the red paper or ribbons now present an almost perfectly black appearance, whilst the green and greenish blue retain their normal colour. In order to understand in what sense sources of light can be compared together, it is necessary to keep in view the fact that every ray of light has three qualities: firstly, what we call its colour; secondly, its luminosity or brightness; and thirdly, its purity, or the degree to which that ray is mixed with white or other light.

When white light, the standard of comparison being daylight, falls upon a coloured surface, be it leaf or flower, some

part of the light is reflected back unaltered in quality. Some portion passes through the surface, and is reflected back by internal reflection, and in performing the double journey through the surface layer it has certain of its constituent rays weakened or destroyed. The apparent hue or colour of the object depends upon the nature of this selective absorption. The brightness or luminosity of the surface depends upon the intensity of the light which falls upon it, and upon the amount by which that light which is reflected from it, whether superficially or internally, is weakened. The eye is sensitive to both these qualities of colour and brightness, and can pronounce judgment upon either of them, or on both. Moreover, we appreciate the extent to which the altered light is mixed with the unaltered light. In the case of coloured surfaces which we call *pale* or *light* in tint, such as a light pink, or light or Cambridge blue, there is a considerable admixture of white light when these surfaces are seen by normal daylight. Hence, when white light falls upon a surface, some of it is reflected unaltered, and some part is returned to the eye after having suffered a weakening or destruction of some, or all, of the rays present in it. A surface we call white reflects a large fraction of the light which falls upon it unaltered in quality; a surface we call black reflects a very small fraction, mostly unaltered, of the light which falls upon it; and a surface we call coloured sends back to our eyes a fraction of the light which falls upon it, but the quality of the light is altered by the diminution in brightness of some, or all, of the various rays present in it.

Suppose that there are two white surfaces, on each of which pure red light from different sources is being thrown. They may differ in brightness or luminosity, but there is generally no difficulty in deciding which of these two surfaces is the brighter. If, however, we have two surfaces, one of which is reflecting pure blue light and the other pure red

light, then we have two differences to appreciate, namely, a real difference in colour due to the difference in the wave lengths of the lights reflected by these surfaces, and a difference in brightness or luminosity, which may, or may not, exist, due to the different intensities of the lights reflected from the two surfaces. A certain training of the eye is necessary in order to distinguish a difference in the luminosity of two surfaces, altogether apart from the question whether they have, or have not, a difference of tint. An inexperienced eye cannot do this with the exactitude of a trained eye. It is comparatively easy in extreme cases. If, for instance, I place before you, in a white light, a piece of dark blue paper and a piece of light yellow paper, you have no difficulty, quite apart from the colour difference, in deciding at once that the blue is darker than the yellow, or less bright; and if, on the other hand, I present to you similar pieces of a light Cambridge blue and a very dark red, you have also no difficulty in deciding as to the relative brightness of those surfaces. The artistic eye is especially trained in this sort of discrimination of brightness or luminosity, as distinguished from colour. An artist, for instance, does not admit that any objects are correctly depicted in which the proper differences of luminosity, or brightness, or of light and shade, as he would say, are not properly expressed or suggested. Very often it is quite out of the power of the artist to give the different parts of the surface of his paper or canvas the same actual relative luminosity or brightness as exists in the case of the objects he is depicting. He cannot, for instance, depict on his canvas dark green leaves when seen against a bright background of white cloud or snow in a manner which shall give them the proper relative brightness or luminosity as seen in Nature, especially when that paper or canvas has to be viewed in the interior of a room. And a great part of the art of painting consists in suggesting differences of luminosity which cannot be exactly obtained. But when an artist makes a sketch with

monochrome, or crayon, or pencil, he endeavours in some sense to so alter the surface of the paper in its various parts that, without regard to colour differences, these patches of the paper imitate more or less nearly the relative luminosity or brightness of the various parts of the surfaces of the objects to be depicted.

When, therefore, objects having what we call colour are viewed by the aid of two different illuminants, each sending out light of different qualities, as far as regards *the colour distinguishing powers* of those two lights there is no sense in which they can be compared. As regards the power of distinguishing colours, we cannot compare an electric arc lamp with a candle, because no number of candles, however great, are, or can be, the equivalent of any arc lamp in their power of revealing or producing colour differences. We can, however, within certain limits, compare lights in regard to their different powers of producing on a white surface equal luminosity or brightness, whether these lights be monochromatic—that is to say, emit rays of only one wave length—or whether they are emitting a compound light composed of rays of many wave lengths. Taking any surface, such as white paper, which is capable of reflecting rays of all wave lengths, at least as far as those which affect the eye are concerned, we may illuminate part of this paper from one source of light and part from another source, and we may adjust the intensity of these two sources of light in such a way that a discriminating eye can assert that the two parts of the white surface are of equal brightness, whether they are apparently of the same colour or not. There are, however, certain great difficulties in doing this which must not be ignored when comparing together the illumination produced by two lights of different character.

It is necessary to bear in mind that in making these photometric comparisons, our eye is the only instrument

which we can employ, and we are limited in our operations by the physiological properties of that organ. The eye is wonderfully susceptible of education, but there are certain inherent properties of it which no education can overcome or displace. It is often taken for granted that the physiological effect—namely, the amount of visual sensation produced by a given light—is proportional to the luminous or intrinsic brilliancy of that light, but this is not really the case. If two white surfaces are illuminated respectively by pure red and pure blue light, and the lights are adjusted so that the surfaces are of apparently equal brilliancy, then, if both the sources of light are doubled or trebled in intensity, the relative illuminations on those surfaces will no longer be equal. In other words, the visual sensation does not increase proportionately to the luminous intensity of the source of light. The visual impression produced by violet or blue light increases more slowly than that of red light when the absolute or intrinsic intensities of the two lights are steadily increased by equal steps. As long, however, as we are dealing with lights of fairly similar character, we have a practical basis for comparison in the experimental fact that the illumination produced on a white surface, due to any such source of light, say a standard lamp, placed at any distance from the surface, can be exactly imitated by placing four such equal sources of light at double the distance from the surface; provided that in both cases the rays of light fall in a similar manner on the surface.

It is owing to the above-mentioned different physiological action of different coloured lights (a phenomenon discovered by Purkinje) that high authorities such as Helmholtz (*"Physiological Optics,"* p. 420) have stated that any comparison of lights of different colours is impossible. One fact is perfectly certain, and that is, that there is no way in which we can by any simple number, such as the candle-power, express

the relative colour-distinguishing powers of different lights, and any attempt to do so is pure nonsense. Taking daylight from a bright northern sky as our standard of normal light, we cannot express the degree in which the light from an electric arc or glow lamp can reveal colour differences as seen by such normal daylight by stating the candle-power of those lights, and hence all such expressions as that an electric arc lamp has so many candle-power are insufficient and inaccurate methods of assigning a visual value to the light. We can, however, compare within certain limits the relative brightness of white surfaces when exposed to the two lights which are to be compared. For many purposes for which we require light, such as reading or writing, this gives us a measure of the value of the light for visual purposes. At present, the only scientific basis for photometry is to be found in the approximately correct fact that *brightnesses* can be added together, and that the illumination or brightness produced on a white surface by two separate lights is the sum of the brightness produced by the individual illuminations. The limitations set upon photometry by the physiological properties of the human eye seem, in some aspects, to be insurmountable; and, since *seeing* means discriminating colour and brightness difference, no substitute for the eye which is not an eye can be found. Our chief resource is the power we have of training the eye to increased visual sensibility, but a person with a naturally "bad eye" for colour or luminosity difference will never make a good photometrist.

In order to effect such a comparison of lights we must start, first, with a standard illuminant, and, second, with a standard of illumination. Unfortunately, the legal standard illuminant is agreed on all hands to be an exceedingly unsatisfactory standard. This standard is, by the Metropolis Gas Act of 1860, defined to be a standard sperm candle $\frac{3}{4}$ in. in diameter, burning 120 grains in the hour, and it is called a *standard or*

parliamentary candle. We will return in a moment to consider the various other and much better standards of illumination which have been suggested, but, for the present, let us assume that by the candle we mean a normal standard candle, or, at any rate, the mean of a large number of such standard candles in illuminating power. The illumination which such a candle produces on a white surface, say a sheet of paper, held at a distance of one foot from it, is called *one candle-foot*, and the candle-foot is the unit of illumination, just as the candle is the unit of illuminating power. The principle on which practical photometry is based is the experimental fact that the brightness of the surface of white paper produced by two candles at a distance of one foot from the surface is twice that produced by one candle at a distance of one foot, and, moreover, that the illumination of such a surface produced by four candles at a distance of two feet is equal to the illumination produced by one candle at a distance of one foot. Similarly, nine candles at three feet and sixteen candles at four feet distance from the white surface produce the same illumination as does one candle at a distance of one foot. The candle-foot is found to be an exceedingly convenient illumination for the purposes of reading. Most persons can hardly read comfortably if the illumination on their book or paper is less than one candle-foot. The illumination in a well-lighted room may vary from two candle-feet on the tables to half a candle-foot on the walls and a quarter of a candle-foot on the floor, while a brilliant illumination, such as that required on a theatre stage, is from three to four candle-feet. The illumination in a street due to the street gas lamps may be from one-third to one-fourth of a candle-foot; in a picture gallery, from one to three candle-feet. The illumination due to full, high moonlight in London is only $\frac{1}{80}$ th to $\frac{1}{100}$ th of a candle-foot, depending on the clearness of the sky and altitude of the moon. These fractional figures must be understood as follows :—An illumination of a quarter

of a candle-foot means illumination which would be produced by a light of one candle held at two feet distance from the white surface, and an illumination of $\frac{1}{16}$ th of a candle-foot is the illumination produced by a candle held at a distance of ten feet from the white surface. The illumination of full sunlight may be from 7,000 to 10,000 candle-feet when the sun is high in a bright clear climate. The mean daylight in the interior of a well-lighted room may be from ten to forty candle-feet. Attempts have been made at various times to estimate the relative brightness of a white surface when held in full sunlight and when held in full moonlight. Various numbers* have been obtained by different observers. Bouguer, in 1725, and Bond, in America, in 1851, made estimates of the relative brightness of sunlight and moonlight. Unless the altitudes of the sun and moon and atmospheric conditions are stated, these relative values do not mean very much. Roughly speaking, full sunlight produces an illumination from 400,000 to 700,000 times as great as that of full moonlight. It is difficult to read anything but fairly large type in full moonlight.*

* Bouguer dispersed the rays of the sun falling on a small hole by a concave lens and compared the diffused sunlight with the light of a candle. Bond formed the image of the sun on a sphere of silvered glass, and compared it when so weakened by reflection with a standard light. Bouguer found that the intensity of full sunlight was 300,000 times greater than that of full moonlight, and Bond found the ratio to be 470,980 to 1. More recently Prof. Young, in America, has made an estimate of the sun's mean candle-power. After correcting for atmospheric absorption, he finds the candle-power of the sun to be $1,575 \times 10^{24}$. Prof. L. Weber, of Breslau, finds that the red rays in sunlight are $1,110 \times 10^{24}$ times more intense than the corresponding rays in candle light, and the green rays in sunlight $2,294 \times 10^{24}$ times as intense as green candle-light rays. Since the sun's mean distance from the earth is 48×10^{10} feet, the sun's illumination at the earth is $\frac{1,575 \times 10^{24}}{48 \times 48 \times 10^{20}} = 7,000$ candle feet. Average moonlight is $\frac{1}{16}$ th of a candle-foot, and hence full sunlight illumination, on a white surface, is to full moonlight in the ratio of 490,000 to 1. (10^{24} stands for a billion times a billion.)

We turn, then, to consider some of the methods of photometric comparison which depend on the principle that one source of light is said to have an illuminating power equal to that of another when the brightness of a white surface held at the same distance from both these sources is the same. Taking, for instance, the normal candle as a standard, any other source of light is said to have four candle-power if it produces the same apparent brightness on a white surface held two feet from it as does a candle if held one foot from the same surface ; such comparison being made wholly with regard to luminosity or brightness of the white surface, and no attention being given to the difference in colour in any

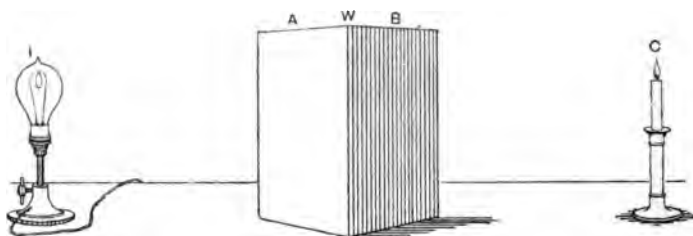


FIG. 10.—Comparison of Illuminating Power of Glow Lamp and Candle by means of Ritchie's Wedge.

white surface when viewed by the two sources of light. The simplest method of effecting this comparison is by the use of Ritchie's wedge. A block of wood is cut in the shape of a wedge with a rather obtuse angle, and has its two adjacent surfaces covered with fine white writing paper, the separating edge being made as sharp as possible (*see Fig. 10*). The two sources of light which are to be compared—say a candle and an incandescent lamp—are placed together in a darkened room. The wedge is held between them, with its edge vertical, and on looking at the vertical edge we see that one side of the wedge is illuminated by the candle and the other surface by the electric glow lamp. We can then move the

wedge to and fro until we find a position such that the two sides of the wedge respectively illuminated by the two agents appear to be of the same brightness. We are assisted in getting rid of any difficulty due to difference of colour in the light by adopting the principle suggested by Capt. Abney, viz., by oscillating the wedge or swinging it to and fro. When we have very nearly found the position of balance, if we move the wedge to and fro in slowly diminishing arcs, we shall alternately increase and diminish the relative brightness of the two surfaces, but we shall not affect any difference in their relative colour tints; hence the attention of the eye is by this device fastened upon that which is varying, namely, the relative luminosity or brightness of the two surfaces, and distracted from that to which we wish to give no attention, namely, their relative colour difference. Having found the position in which the two inclined surfaces of the wedge appear equally bright when each respectively is illuminated by one of the two sources, we measure the distance from the candle to the wedge and from the lamp to the wedge. The illumination produced by the candle is then obtained by dividing unity, since the candle is the unit of illuminating power, by the square of its distance in feet from the wedge; and, likewise, the illumination produced by the electric glow lamp is equal to its unknown candle-power divided by the square of its distance in feet from the wedge. Since these illuminations are equal, it follows that the candle-power of the glow lamp must be numerically given by dividing the square of the distance of the glow lamp from the wedge by the square of the distance of the candle from the wedge. Thus, if the balance has been found when the glow lamp is four feet from the wedge and the candle is one foot from the wedge, the candle power of the glow lamp is four times four divided by one times one, or 16.

Instead of employing a wedge, some observers make use of a method due to Rumford, in which the two illuminants are

made to cast the shadow of a stick upon a white screen. If a glow lamp and a candle are placed a short distance apart (*see* Fig. 11), and a pencil or other rod of wood is held an inch or two from a white card, which is placed so as to be equally illuminated by the two sources of light, it will be found that there are two shadows on the card. One is a shadow due to the lamp, and the other is a shadow due to the candle. The shadow due to the candle is a space into which candle-light does not shine, and upon which the light from the lamp shines, whilst the shadow due to the lamp is a portion of the surface which is illuminated by candle-light alone. By suitably moving the

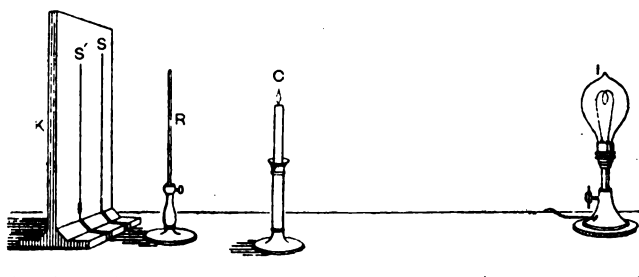


FIG. 11.—Comparison of Illuminating Power of Glow Lamp and Candle by means of Rumford Shadow Method.

candle and the lamp we find we can obtain positions in which the two shadows are of equal depth, and we then obtain the candle-power of the incandescent lamp by dividing the square of its distance in feet from the screen by the square of the distance in feet of the candle from the screen. All that we are doing in this case is to compare together the illumination produced upon a white surface which is partly illuminated from one source and partly from another, and the advantage which the rod gives us is that we can make these two shadows or illuminated surfaces just touch one another, and so be assisted in making a comparison between their relative brightnesses. If the same experiment is

tried with moonlight and a candle, the observer will very soon realise the difficulties that attend photometric measurement when the lights are of different qualities. On any bright moonlight night hold a white piece of card at the window, and place a candle so as to illuminate the card at about the same angle. Hold a pencil in front of the card, and it will be seen that there are two shadows, one of which is a bright blue and the other a bright yellow. The yellow shadow is, apparently, the shadow thrown by the moon, and the blue shadow is, apparently, the shadow thrown by the candle. In reality, that which we take for the shadow due to the moon is a space illuminated by candle-light alone, and the shadow we take for the shadow produced by the candle is a space on the card illuminated by moonlight alone.* These two surfaces differ not only in colour, but in brightness, because the light falling upon them is very different in quality. By moving the candle we can find a position, generally from seven to ten feet from the card, in which the blue and yellow shadows, though different in colour, have apparently the same depth. But the difficulty of determining when this relative luminosity is the same will give an observer who tries the experiment for the first time an insight into the difficulties of photometric measurement.

A third and more frequently-used method for the comparison of sources of illumination is the grease-spot photometer of Bunsen. If a piece of thin paper has a spot of grease placed upon it, and if we hold up the paper between our eye and the light, we find the grease-spot more transparent than the rest of the paper, and it looks lighter. But if we hold the paper so that the light falls on it, we see that the grease-spot is apparently darker than the rest of the paper.

* It was this and other similar phenomena which drew Goethe into some confusion of thought on the subject of colour, and led him in his *Farbenlehre* to dispute Newton's theory of colour.

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If the disc of paper with a grease-spot upon it (*see* Fig. 12) is held up between two lights, such as a gas flame and a candle, and looked at, first from one side and then from the other, it will be found possible, by moving the disc, to find a position for the piece of paper such that the grease-spot can hardly be seen from either side. When this is the case the two surfaces of the paper are equally illuminated by the two sources of light, and if one of these sources of light is a standard candle, then, as above explained, the candle-power of the other source of light is obtained by dividing the square of the distance in feet of the source of light from the screen by the square of the distance in feet of the candle from the screen.

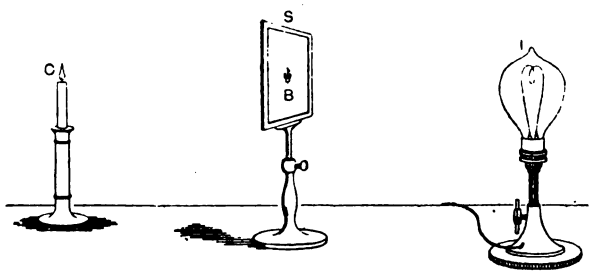


FIG. 12.—Comparison of Illuminating Power of Glow Lamp and Candle by means of Bunsen Grease-Spot Method.

In practice various devices are used, such as a pair of inclined mirrors at the back of the disc, and arrangements for reversing the disc, to enable the observer to rapidly make a fair comparison between the two surfaces of the disc, and to ascertain if, when equally viewed from both sides, the grease-spot is invisible. The grease-spot method is one which is largely used in the determination of the candle-power of illuminating gas. The Gas Companies are bound to supply gas of a certain candle-power when burnt from a particular standard burner, and official comparisons have to be made in order to determine if they comply with their obligations. If

the grease-spot or shadow methods are employed for the comparison of a candle with a glow lamp, we do not find any serious difficulties in determining the relative brightness of the white surfaces illuminated by the two sources of light, because the qualities of these two lights are not very different—that is to say, if we formed a spectrum from each light we should find that the different rays in these spectra, which are of the same wave length, can all be made to be equal in luminosity; or, in other words, if the luminosity of the red ray in the glow lamp is equal to the luminosity of the red ray of the same wave length in the candle, then the green, blue and violet rays of corresponding wave lengths are also equal as regards luminosity. If the experiment is tried with two lights of very different qualities, such as an electric glow lamp and an electric arc lamp, considerable difficulty will be found by the unpractised observer in determining when two adjacent surfaces illuminated by these two sources of light are of equal brightness. In order to diminish this difficulty, observers have sometimes employed coloured glasses to select particular rays from the arc lamp and compare them with similar rays in the candle-light, and numerous determinations exist of the candle-power of different arc lamps for red and green rays. But this really serves no useful purpose. The practical value of a light is not the relative intensities of certain rays when compared with rays of the same wave length in the light of a candle, but the total brightness which that light is capable of producing on a white surface when compared with the total brightness which the unit illuminant is capable of producing on the same surface.

We may further illustrate by an experiment the difficulty of comparing together the brightness of two surfaces of different colour in the following manner:—An electric glow lamp with a red glass bulb and an electric glow lamp with a

green glass bulb (*see* Fig. 13) are so placed as to equally illuminate a white surface. A rod or other opaque body is then placed so as to cast two shadows, one from each lamp. You observe that the two shadows are respectively red and green; the red lamp appears to cast a green shadow and the green lamp appears to cast a red shadow, and you will find considerably more difficulty in comparing together these two lights by this Rumford method of photometry than would be the case if the two lights were both of them white or both of them red. By adopting the method of oscillating one of the lights so as to

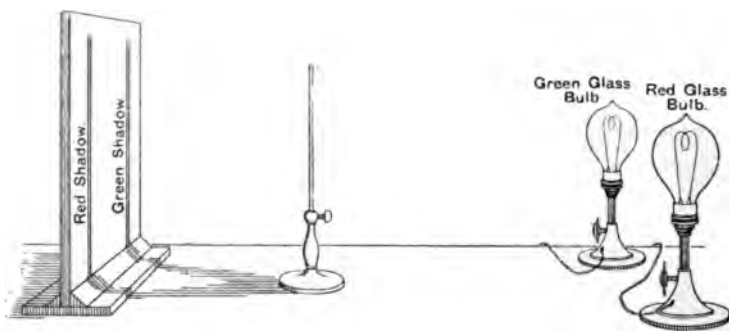


FIG. 13.—Comparison of Illuminating Power of Two Lights of Different Colours by Shadow Method. The two glow lamps have glass bulbs of different colours.

alternately increase and diminish the brightness of one of the shadows, the difficulty of making this luminosity comparison is partly overcome.

With regard to the standards of illuminating power, it has been previously stated that the British standard candle is an unsatisfactory standard. The French adopted a standard oil lamp, called a carcel standard lamp, burning 42 grammes of colza oil per hour with a wick of a certain size. This carcel standard gives a light equal to

about $9\frac{1}{2}$ British candles. A standard which has been adopted to some extent in Germany, and to a less extent in England, is the Hefner-Alteneck amyl-acetate lamp. This is a little spirit lamp burning pure pear oil from a wick of certain size, and yielding a flame 40 millimetres high, with an illuminating power of rather less than one British candle. But the objection to its use as a standard is that it gives a somewhat reddish light. Another standard largely used in England is the Methven gas standard. In this standard, ordinary coal gas, enriched with benzol or pentane, two hydro-carbon liquids, is burnt at a particular form of argand burner. In front of the flame is placed a metal plate, having a slit in it of such a size that it only permits light to pass from the centre of the flame, and this slit is adjusted so that the light is equal to two standard candles. Variations of quality and pressure in the gas are said not to affect the intensity of the light which is emitted through the slit. In practice this is not found to be exactly the case. By far the most satisfactory standard which has yet been proposed is the pentane air-gas standard of Mr. Vernon Harcourt. In this standard a mixture of volatile hydro-carbon (called pentane) and air is burnt at a jet of a certain kind in such a manner as to yield a flame, produced by the combustion of a definite chemical compound, such flame having a certain height and dimensions. By taking suitable precautions, such a pentane gas standard can be made to yield excellent results as a standard of light. It has been proposed by M. Violle, a distinguished French chemist, that an absolute standard of light should be obtained by taking the light emitted from one square centimetre of molten platinum. Such a standard, as a practical standard of reference, is not very convenient or suitable for use as a general primary standard of light, but it may be found valuable as an ultimate reference. For some purposes the only satisfactory method of comparing together two sources of light, and examining their qualities in regard

to the relative proportions and brightness of the different rays existing in those lights, is by employing the spectro-photometer. This instrument consists of an arrangement whereby the same prism is made to analyse the light from two sources of light, and to expand the rays sent out from these two sources of light into two rainbow bands or spectra in such a manner that these spectra are placed one above the other, the corresponding colours being vertically over one another. It is then possible to weaken one of these sources of light by interposing either a disc with an aperture in it of variable size, revolving at a rapid rate, or by employing a pair of nicol prisms, by means of which the light from one source can be weakened to any extent. The brighter of these sources of light is weakened until one particular ray, say the yellow ray, in that light is exactly equal in brightness to the corresponding yellow ray in the spectrum of the other source of light. In other words, the spectra are made to match each other exactly at this particular spot. It will then be found that if the two sources of light are of different qualities, say a glow lamp and an arc lamp, the two spectra will not match each other as regards brightness in any other places.

Prof. Nichols, of Cornell University, U.S.A., has carried out a series of observations with an apparatus of this kind. Taking an Edison 16-candle-power incandescent lamp as his standard, he calls the brightness of the spectrum of this light everywhere unity, and he compares with it the spectrum of any other light, say an arc lamp, of which the spectrum has been weakened so as to make it of identical brightness for one particular yellow ray with the corresponding ray in the spectrum of the glow lamp. It is then found that the spectrum of the arc lamp is less bright in the red rays, but much more bright in the green, and especially much more bright in the violet. In the spectrum of the arc lamp, especially when that spectrum is taken in such a way as to utilise a large portion of the light proceeding from the carbon

vapour between the poles, a very bright violet band is found at the extreme end of the spectrum. These observations have been expressed by Prof. Nichols in a series of curves (see Fig. 14). The horizontal line at the base of the diagram represents the spectrums; the letters A, B, D, E, &c., indicate

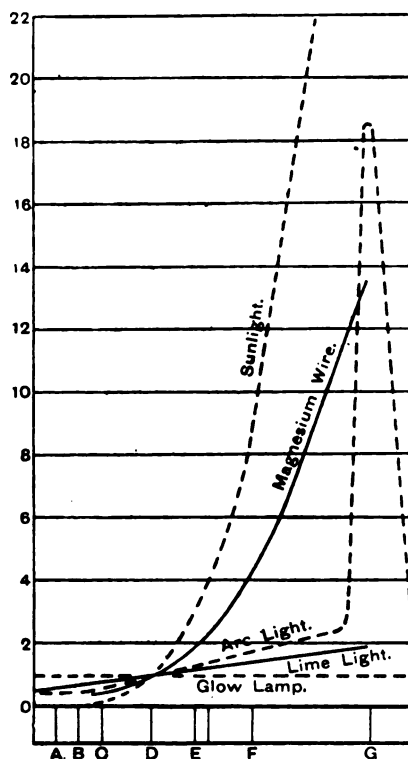


FIG. 14.

the position of the characteristic black lines or missing rays in the solar spectrum, which are taken as points of reference. The intensity of the spectrum of the glow lamp is taken everywhere as unity, and is represented by the altitude of the horizontal line marked *glow lamp*. The other curves then

represent, by their ordinates or heights at various points, the relative brightness of the respective rays in the different portions of the spectra of the light taken from the sun, the arc lamp, and magnesium wire, compared with the ray of the same wave length in the light from the electric glow lamp. It will be seen how enormously brighter the sunlight and the arc light spectrum are in the neighbourhood of the violet rays when compared with the corresponding ray of the glow lamp, but that their light, relatively speaking, is less bright in the red than the corresponding light of the glow lamp, provided that the spectra of all these sources of light have been equalised so as to make them of identical brightness in the neighbourhood of the D line of the solar spectrum—that is, of the yellow rays.





LECTURE II.

THE Evolution of the Incandescent Lamp.—The Nature of the Problem.—Carbon the only Solution.—Allotropic Forms of Carbon.—The Modern Glow Lamp.—Processes for the Manufacture of the Filament.—Edison-Swan Lamps.—The Expansion of Carbon when Heated.—Various Forms of Glow Lamps.—Focus Lamps.—High Candle-power Lamps.—Velocity of Molecules of Gases.—Kinetic Theory of Gases.—Processes for the Production of High Vacua.—Necessity for a Vacuum.—Mean free path of Gaseous Molecules.—Voltage, Current and Candle-power of Lamps.—Watts per Candle-power.—Characteristic Curves of Lamps.—Life of Glow Lamps.—Molecular Shadows.—Blackening of Glow Lamps.—Self-recording Voltmeters.—Necessity for Constant Pressure of Supply.—Changes Produced in Lamps by Age.—Smashing Point.—Efficiency of Glow Lamps.—Statistics of Age.—Variation of Candle-power with Varying Voltage.—Cost of Incandescent Lighting.—Useful Life of Lamps.—Importance of Careful "Wiring."—Average Energy Consumption of Lamps in various Places.—Load Factors.—Methods of Glow-lamp Illumination for Production of Best Effects.—Artistic Electric Lighting.—Molecular Physics of the Glow-Lamp.



THE illustrations and explanations in the previous Lecture will have prepared the way for a study of the electric incandescent lamp as a source of illumination. We shall not occupy time by discussing the stages by which the glow lamp has attained its present perfection. Neither is it necessary to dwell upon questions

of scientific or legal priority which are not most advantageously discussed from the lecture table. Suffice it to say that in its modern form the electric incandescent lamp always has been, and always will be, inseparably associated with the names of

a simple horse-shoe shape as originally adopted by Edison ; whilst in other cases it is given a double twist, as in the Edison-Swan lamps, one or more curls being used. For some purposes it is necessary to coil the filament into a tight compact coil, in order that, when the lamp is used in the focus of a mirror, the light may be all concentrated in one place, and

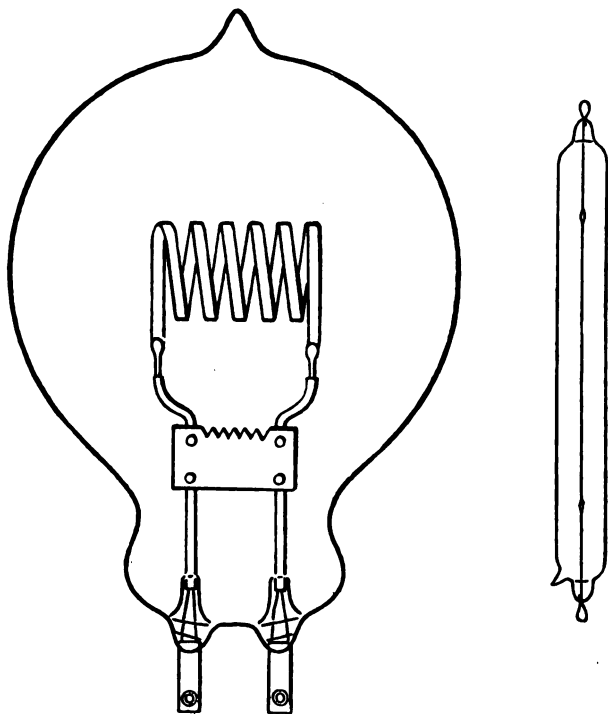


FIG. 18.—Glow Lamps with Zigzag and Straight Carbons.

be gathered up by the mirror or lens. Such lamps are called *focus* lamps, and are employed in optical lanterns, in ships' side-lights, and wherever it is necessary to collect all the rays from the filament by optical means. These lamps are very useful in optical lanterns, and, now that so many houses are

supplied with electric current, afford an easy means of showing lantern views in a drawing-room.* In cases in which high candle-power is required, it is obtained by putting two or more filaments in the same lamp bulb. In Fig. 19 is shown a representation of a multiple-filament lamp of 2,000 candle-

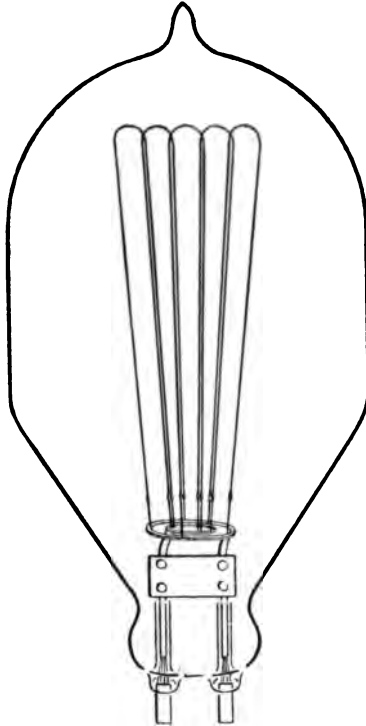


FIG. 19.—2,000 candle-power Glow Lamp with Multiple Filaments.

* Photographic amateurs who practise the art of making lantern slides from negatives, want an easy and ready mode of exhibiting them. If electric current is laid on to the house, the simplest method of doing this is to fit an ordinary optical lantern with a 100-volt focus lamp of 50 candle-power. This can be fitted on the tray which usually carries the oxyhydrogen jet. A screen of drawing paper six feet square makes an admirable surface for projection, and the views can be shown in a drawing-room without danger or smell.

power as constructed by the Edison-Swan Company. In these lamps a number of carbon conductors of the horse-shoe shape are arranged between two rings, so that the electric current can pass from one ring to the other through four, or five, or six, or more carbon conductors, each of which has one leg attached to one ring and the other leg to the other. Lamps of this high candle-power are made up to 3,000 candle-power or more. In such cases the glass bulbs become globes of glass nearly a foot in diameter, and have to be made of considerable thickness to withstand the enormous pressure of the air on the outside. The large lamp now shown to you in action has a globe having a surface of 280 square inches, and the air pressure on the exterior amounts to one and three-quarter tons! One limitation to the safe size of such lamps is this great external air pressure.

The final stage of manufacture of the lamp is the exhaustion of the lamp bulb. It was mentioned above that this was rendered essential, in the first place, by the fact that, if heated in the air to a high temperature, the carbon filament would soon be destroyed by combustion. Taking a non-exhausted lamp, I pass an electric current through the filament. You see that the lamp burns for a few seconds and then goes out. The oxygen of the air in the bulb has combined with the carbon of the filament in one or more places and destroyed it by producing combustion of the material. But if we take a filament of carbon not enclosed in a bulb and plunge it into a glass vessel full of a non-oxidising atmosphere, such as carbonic acid gas or coal gas, and heat it electrically, you will see that the carbon, though rendered incandescent, is not then rapidly destroyed. The carbon filament is enclosed in a vacuum, not merely to prevent its combustion, but there is another important reason for exhausting the bulb of all gases or air. Even if we suppose the bulb filled with a non-oxidising atmosphere of nitrogen or

hydrogen, such a lamp would yet not be so perfect as a vacuum lamp, and for the following reason:—Modern physical research has provided arguments well nigh irresistible to prove that gases consist of molecules or little particles which are in rapid motion. This *kinetic theory* of the structure of gases, as it is termed, is supported by an immense body of physical evidence, and no less an authority than Lord Kelvin has called it one of the surest articles of the scientific creed. Certain lines of investigation have shown how to determine the *average velocity* of these gas particles. It has been shown, for instance, that in hydrogen gas at ordinary pressure and at the temperature of melting ice the molecules are moving with an average velocity of about 6,000 feet per second, or 69 miles per minute, and in oxygen with an average velocity of 18 miles per minute. These gas molecules in their rapid flight collide against one another in moving to and fro, and the average space over which the molecule flies before it has a collision against some other molecule is called its *mean free path*.

In ten cubic inches of air or oxygen at the ordinary pressure, and at the temperature of melting ice, there are, in all probability, a number of gas molecules represented numerically by a billion times a billion, or by unity followed by twenty-four cyphers (10^{24}). The mean free path of the oxygen gas molecule is rather more than two-millionths of an inch, and the average number of collisions during the one second in which it darts over a distance in all of 1,500ft. is 7,600 millions. We are here dealing with minute portions of time and space, in which the unit is not a second and an inch, but a millionth part of each of these. The millionth of an inch is about the same fraction of an inch that one foot is of 200 miles; and the millionth part of a second is the same fraction of a second that one second is of about twelve days. All the air we breathe and feel consists, therefore, of these flying and colliding oxygen and nitrogen

molecules. Each particle or molecule is always having its direction changed by its collisions against neighbours, just as a billiard ball has when it "cannons" against another. It is the constant bombardment of these molecules against the walls of a containing vessel which creates the elastic pressure of a gas, and when any solid body is surrounded by a gas, an enormous number of collisions take place against that surface in every second of time.

If the surface is heated, the molecules which come up against it take heat from it in their myriads of contacts per second, and this removal of heat from the body by the gas molecules is called *convection*. If the carbon filament is enclosed in a glass bulb not exhausted of its air, a certain amount of energy is thus removed from the filament and given up to the glass, being conveyed by these molecular vehicles. The energy so abstracted from the filament is not in any way conveyed to the eye as light. Hence, for all illuminating purposes, it is lost, and a lamp so constructed with a non-vacuous bulb cannot in some respects be as efficient a device as one with a high vacuum, because of this additional loss of energy from the filament.

The process adopted for removing the air from the bulb is generally the employment of some form of mercury pump in combination with a mechanical air-pump. Time will not permit me to describe the many forms of this device which have been suggested. Briefly speaking, the principle upon which many of these mercury pumps act is as follows:—A vertical glass tube *F* (see Fig. 20) has a side tube *S* sealed into it, to which is attached the lamp to be exhausted. Drops of mercury are allowed to fall down the vertical tube, and as they fall down they push the air in the form of bubbles before them. The air is, therefore, gradually drawn out from the lamp bulb *L*, and the completion of the process of exhaustion is recognised

when the globules of mercury fall down the vertical tube in close contact with one another without carrying any air between them. Under these circumstances, if the bottom end of the vertical fall tube is placed in a vessel of mercury *V*, the mercury will gradually rise in the fall tube to the height

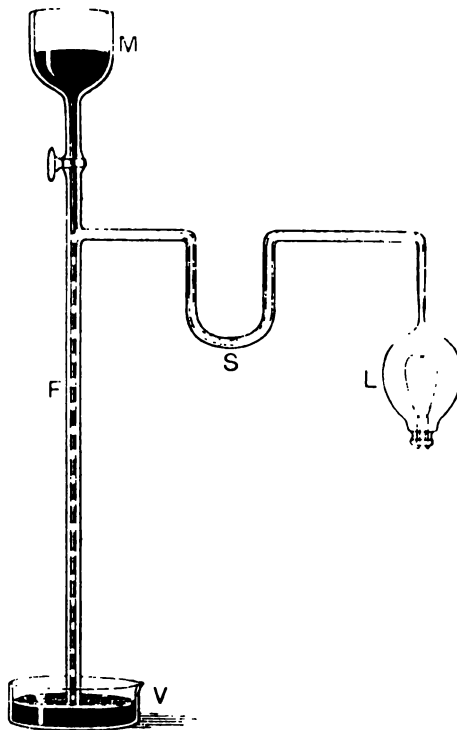


FIG. 20.—Diagram of arrangements of a Mercury Pump for Exhausting Lamps.

M Mercury Reservoir ; *F*, Fall Tube ; *L*, Lamp being Exhausted ;
S, Side Tube ; *V* Vessel of Mercury.

of the mercury barometer—that is, to about 30in.—and then each succeeding drop of mercury which falls down on to the top of the column, does so with a sharp click, thus showing

that the lamp bulb has been exhausted of its air to a very high degree. It is not difficult to abstract from such a glass lamp bulb all but one-millionth of the air originally contained in it. A highly exhausted space of this kind is commonly called a vacuum; nevertheless, when we realise that every ten cubic inches of air at ordinary pressure contains a billion times a billion molecules of air, it is easily seen that the reduction of this enormous number, which may be contained in a lamp bulb of about three inches in diameter before exhaustion, to one-millionth of their original value, still leaves included in the bulb a very respectable number of molecules, viz., a million times a billion. One important change that is, however, made in the physical state of the gas when so reduced to a very low pressure is that the *mean free path* of the molecules is enormously increased. By this reduction in pressure, the mean free path of the molecules in space, on an average, traversed before collision, is increased just in proportion as the pressure is diminished.

By the reduction of the pressure to one-millionth of its normal value by the abstraction of 999,999 millionths of the air, the mean free path of the air molecules is increased a million-fold—in other words, to something not far from two inches. In the interior, therefore, of a large glow lamp, which may have a volume, say, of ten cubic inches, there is good reason to believe that, when the bulb is exhausted to the highest point usually obtainable in practice, there still remain an immense number of molecules of air, but that the space is, comparatively speaking, so sparsely populated with molecules that each molecule in flying to and fro, may move, on an average, over a distance of an inch or two without collision against a neighbour. The rarefaction of the air must, therefore, decrease immensely the rate at which energy is taken from the filament by convection, because it decreases the number of molecular bombardments against the

filament, and this is one object of producing the exhaustion. When the proper vacuum has been obtained in the lamp, the glass inlet tube is melted, and the lamp is sealed off from the pump. The lamp is finished by adding to it a brass collar with two sole-plates of brass, which are fastened on to the ends of the platinum wires which protrude through the glass. In this way an electric current can be sent through the filament, and yet the highly perfect vacuum in the glass bulb be indefinitely preserved.

Lamps so made may take numerous forms. In Fig. 21 is shown a form of lamp called a candle-lamp, employed



FIG. 21.—Electric Candle Lamp, of which the Glass Bulb is shaped like a Candle Flame.

for decorative and other purposes. Lamps may be made so small that they can be employed for theatrical, decorative, microscopic, dental, and surgical purposes, in which case they are called *micro-lamps*; they may be made so large, and have so many carbon conductors in them, that they can yield a candle-power of two to three thousand, and the glass bulbs may, in both cases, be given any requisite form and colour.

Having thus briefly described the process of the construction of a glow lamp, we have now to study a little more closely

the laws which govern its operation. From what has been already stated in the First Lecture, it will be seen that, in order to possess a full knowledge of the behaviour of the lamp as an energy-translating device, it is necessary to know four things about it. First, we must know the current, in amperes, which is taken by the lamp when a certain electric pressure is produced between the ends of the carbon filament. Generally speaking, lamps are denominated by the pressure at which they must be worked to give satisfactory results as to duration. This pressure measured in volts is called the *marked volts* or *maker's volts*, and lamps may be distinguished as 100-volt lamps, 50-volt lamps, 10-volt lamps, &c. The current taken by a lamp can be measured by many kinds of instruments, which are called *ampere-meters* or ammeters. The pressure between the terminals of the lamp can be measured by suitable *voltmeters*. Approximately speaking, a lamp intended to work at 100 volts and to give 16 candles light has a carbon filament about five inches long and about one hundredth of an inch in diameter, and when worked at a pressure of 100 volts it takes from $\frac{3}{8}$ to $\frac{2}{3}$ of an ampere. From explanations previously given it will be remembered that if these two numbers—namely, the number representing the volts and the number representing the amperes—are multiplied together, we have a number representing the *electrical power*, estimated in watts, which is being taken up in the lamp, and such a 16-candle-power 100-volt lamp generally takes from 50 to 60 watts. We have already described the methods by which the candle-power can be measured. If we divide the number representing the electrical power in watts taken by the lamp by the number representing the candle-power of the lamp, the quotient gives us the number representing the *watts per candle-power*. In most modern lamps this last number varies from $2\frac{1}{2}$ to $3\frac{1}{2}$, when the lamp is being used at the marked volts. There are, therefore, four quantities in

connection with every lamp which it is important to know. First, the *marked volts* or voltage of the lamps; second, the *ampere current* taken by the lamp at the marked volts; third, the *candle-power* of the lamp; and, fourth, the *watts per candle-power*. Very briefly we may describe the process by which these four quantities are determined.

We have already described the instrument called an electrostatic voltmeter, which is used for measuring electric pressure.

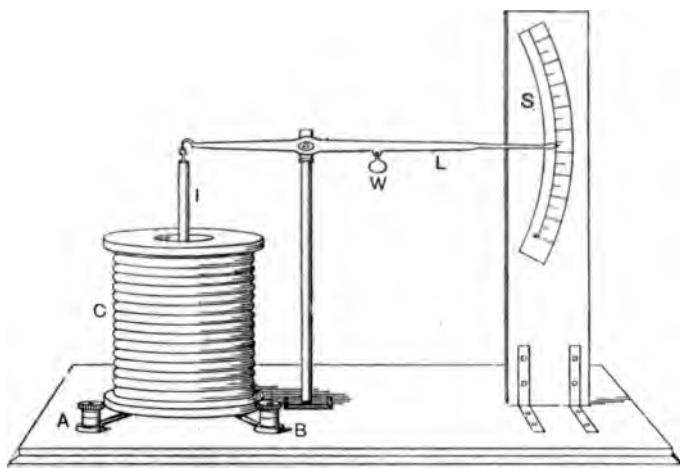


FIG. 22.—Simple Ampere-Meter for Measuring Lamp Currents.

One form of instrument which may be used for measuring the strength of the current taken by the lamp, and which is called an ammeter or an ampere-meter, is made as follows:—A coil of wire **C** (see Fig. 22) is wound on a bobbin, and through this coil of wire the current to the lamp is passed. This current creates a magnetic field round the wire, as described in our First Lecture. If a small piece of wire **I** is suspended on the end of a delicately balanced lever, so that the iron is held freely just at the entrance of the bobbin, the passage of an electric current through the bobbin will attract the iron into the coil.

This movement of the iron is resisted by a weight, **W**, properly placed on the balance arm, or by the weight of an indicating needle. Such an instrument may be graduated to show on a divided scale the strength of the current in amperes which is passing through the coil.

It is important that, in making thus a determination of the constants of a lamp—viz., the current passing through the lamp, the pressure difference at the terminals of the lamp, and the candle-power of the lamp by comparison with a suitable standard of light—these quantities should all be measured at the same instant. By varying the current through the lamp, as we can do by using pressures of different values, we can obtain a series of observations by which the candle-power, current, and volts on the terminals of the lamp are all measured simultaneously for different values of the pressure on the lamp terminals, and these values can be set out in a series of curves which are called the characteristic curves of the lamp. In Fig. 23 is shown a curve illustrating the manner in which the current of a 16-c.p. lamp varies as the current changes. It will be seen that a very small increase in the current is accompanied by a large increase in the light of the lamp, and it can be shown that the candle-power varies very nearly as the sixth power of the current. That is to say, if a lamp gives 1 candle light when a current of one ampere is passing through it, then it would give $2 \times 2 \times 2 \times 2 \times 2 \times 2 = 64$ candle-power if two amperes were passed through it. For a reason to be explained later, however, no lamp would bear such a relative increase in current. But the above-mentioned law holds good proportionately for all glow lamps, viz., that the candle-power varies approximately as the fifth or sixth power of the current. In Fig. 24 is shown another curve, illustrating the manner in which the candle-power varies with the volts, and in Figs. 25 and 26 curves showing the manner in which the candle-power varies with the watts and with the watts per

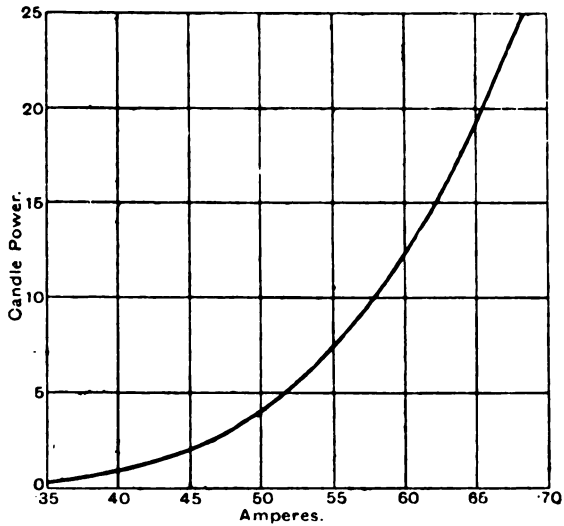


FIG. 23.—Curve showing Variation of Candle-power with Current in an Electric Glow Lamp.

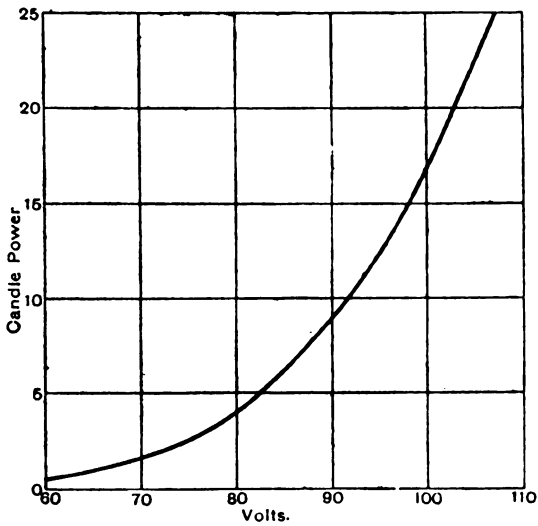


FIG. 24.—Curve showing Variation of Candle-power with Voltage in an Electric Glow Lamp.

candle-power. From the curve in Fig. 25 it will be seen that the candle-power of the lamp varies very nearly as the cube of the total power in watts supplied to it. In other words if we supply the lamp with twice the electrical power, the candle-power is increased eight times. If we supply it with three times the electrical power, it is increased twenty-seven times. These curves will show, amongst other things, the great importance of preserving the pressure at the terminals of a lamp quite constant when that lamp is used as an

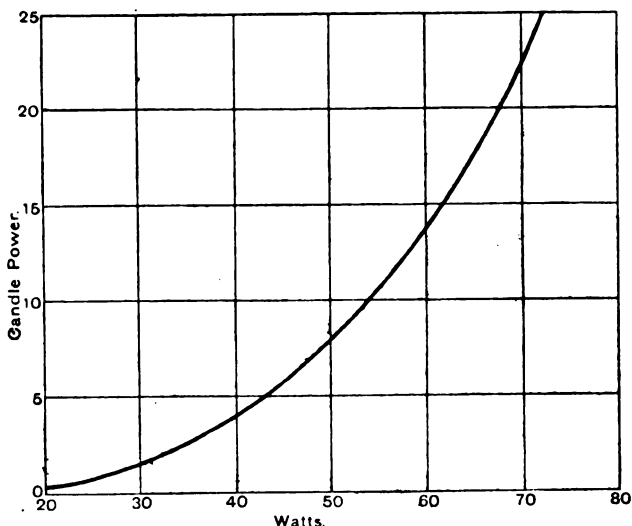


FIG. 25.—Curve showing Variation of Candle-power with Wattage in a Glow Lamp.

illuminating agent. Any variation of the pressure, however small, produces a very serious variation in the candle-power of the lamp. It is convenient to bear in mind the following figures for the 16-c.p. lamp as generally made. Such a lamp takes, when worked at a pressure of 100 volts, 0.6 or $\frac{3}{5}$ of an ampere, absorbs 60 watts of power, and therefore in $16\frac{2}{3}$ hours consumes an amount of energy which is $16\frac{2}{3}$ times $60 = 1,000$

watt-hours. This amount of energy is called one Board of Trade Unit, and is generally sold by the Electric Supply Companies in England at prices varying from 8d. to 8d.

As soon as we begin to use incandescent lamps in practice, even although the pressure at the terminals may be kept exceedingly constant, we find that lamps are like human beings—they have a certain life or duration, and moreover most of them undergo a process of deterioration in light-giving

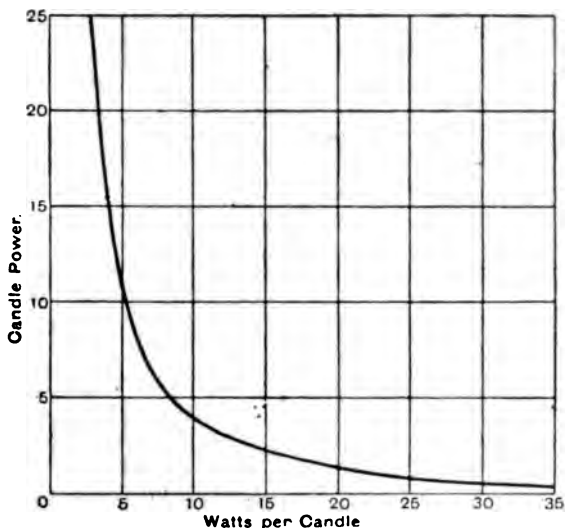


FIG. 26.—Curve showing the Variation of Candle-power with Watts per Candle-power in an Electric Glow Lamp.

power. After a certain time the filament of the lamp is destroyed, and the lamp ceases to work. The duration of any lamp when worked under certain conditions as to pressure and supply is called its *life*, and from a large number of observations of similarly-constructed lamps when worked in a particular manner we can deduce the *average life* of a lamp. We cannot predict from observations of one or two lamps

what the average duration will be, any more than we can tell from observations on one or two human lives what is the average duration of human life in any place. The life of any particular individual lamp may be only a few hours, or it may be many thousands of hours. We shall presently point out that this average duration is only one factor in determining the real value of the lamp as a translating device. Apart from accidental circumstances, the factors that determine the life of the lamp are the temperature at which the carbon filament is kept, and the nature of the surrounding gas. Assuming, however, that an incandescent lamp is supplied with electric current at a constant electric pressure: it is always found that two marked changes occur in the lamp in the process of time; first, the lamp begins to blacken by a deposit of carbon which is made upon the interior of the bulb; and, second, the carbon filament undergoes a change by which its electrical resistance becomes increased and its surface altered, so that, whatever its initial condition, its surface finally assumes a much darker and more sooty appearance than when first manufactured.

From these physical changes it follows that the candle-power of the lamp is diminished, first by an obstruction of light by the black deposit of carbon on the inner surface of the glass, and, second, owing to the increase in the resistance of the filament, the lamp taking less current and therefore furnishing less light. The deposit of carbon on the interior of the bulb is produced partly, or perhaps generally, by a process which is an ordinary evaporation of the carbon, but it may be produced partly by volatilisation of condensed hydrocarbons from the interior of the conductor. In addition, however, to this, there is a process of projection of carbon molecules from the filament in straight lines which is not of a kind generally included in the term volatilisation. Many blackened carbon lamps show lines of no deposit on the glass, which

have been called *molecular shadows*. It is not unusual to find lamps the interior surface of the glass of which has become very black by a deposit of carbon. On examining the lamp it will be found that on one side of the glass, lying in the plane of the loop, there is a line of clear glass on which no deposit has been made. This is very often shown well in lamps in which the filament has been worn away at a point about halfway down one leg of the loop (*see* Fig. 27); and on examining such a lamp it will be found that there is a well-marked line of no deposit upon the glass, just in the

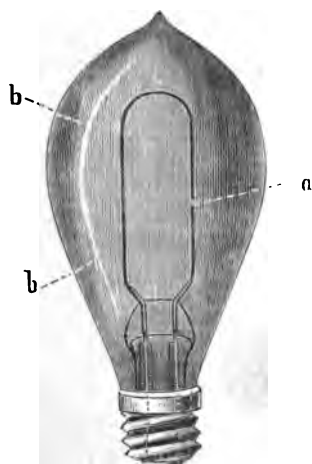


FIG. 27.—Glow Lamp with the glass bulb blackened by deposit of carbon, molecular scattering having taken place from the point marked *a* on the filament, and a shadow or line of no deposit thus produced at *b* on the glass receiver.

plane of the loop, and on the side farthest removed from the point of rupture. Blackened lamps of this kind, therefore, indicate clearly that carbon molecules have been shot off in straight lines from the place where the carbon has subsequently been ruptured. If in a carbon filament there happens to be a weak spot or place of high resistance, at that point the current will generate heat at a greater rate than at other

places, and the temperature of that spot will rise above the temperature of the rest of the filament. Such points of greater temperature can often be seen on carefully examining an old or bad filament. From these high-temperature spots, carbon molecules must be projected off by an action which may be partly electrical, and are deposited upon the glass on all sides. But it is clear that the unbroken leg would shield part of the glass from this carbon molecule bombardment, and a *shadow* of one-half of the loop would therefore be formed upon the glass.

This shielding action may be illustrated by a simple experiment. Here is a \cap -shaped rod (Fig. 28). This shall represent

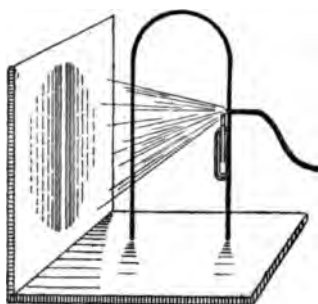


FIG. 28.—“Spray Shadow” of a rod thrown on Cardboard Screen to illustrate formation of molecular shadow in Glow Lamps.

the carbon conductor in the lamp, and this sheet of cardboard placed behind it the side of the glass bulb. I have affixed a little spray-producer to one side of the loop, and from that point blow out a spray of inky water. Consider the ink spray to represent the carbon atoms shot off from the overheated spot. We see that the cardboard is bespattered on all points except along one line where it is sheltered by the opposite side of the loop. We have thus produced a “spray-shadow” on the board. The existence of these molecular shadows

in incandescent lamps leads us, therefore, to recognise that the carbon atoms must be shot off in straight lines, or else, obviously, no such sharp shadow could thus be formed. It has been asserted by some writers that the whole of the black coating on the interior of the carbon glow lamps is produced solely by a process of evaporation of carbon; but the frequent existence of these shadows on blackened lamps show that under some conditions the projection of carbon from the filament is not merely a general and irregular volatilisation, but a copious projection of carbon molecules, which move out in perfectly straight lines from one part of the filament.*

It is clear, therefore, from all the foregoing considerations connected with the processes which are going on in the interior of the lamp and the resulting ageing of the filament, and from the changes that take place in the candle-power with change of current, that the following conditions must be observed in order to secure the best results in the employment of incandescent lamps for illuminating purposes. In the first place, the electric pressure which is supplied to the consumer must be exceedingly constant. Where a supply is drawn from the mains of a public electric lighting company the consumer will, usually, have no control over the pressure, at any rate in the case of continuous currents. The public supply companies of Great Britain are bound by the Electric Light-

* The author possesses a large collection of lamps showing these molecular shadows. In the case of the old Edison lamp, the bamboo filament was attached to the platinum leading-in wires by a deposit of copper made over the clamp. Under these conditions, if one clamp became loose, or made a bad contact, a scattering of copper molecules took place from that clamp, producing a green transparent deposit of copper over the interior of the bulb, which deposit generally showed a well-marked shadow of one-half of the carbon loop upon the glass. Shadows have also been produced by the writer by the volatilisation of aluminium plates in lamps, the volatilisation having been created by passing a sudden strong current through an aluminium wire or plate included in the lamp bulb. This aluminium deposit is of a fine blue colour.

ing Acts to furnish current at a certain constant pressure, which is generally 100 or 110 volts, and conditions are laid down by the British Board of Trade that the pressure shall not vary more than four volts above or four volts below this specified pressure. These are very wide limits of variation, and very few of the supply companies avail themselves of them. The consumer who takes current from a public supply

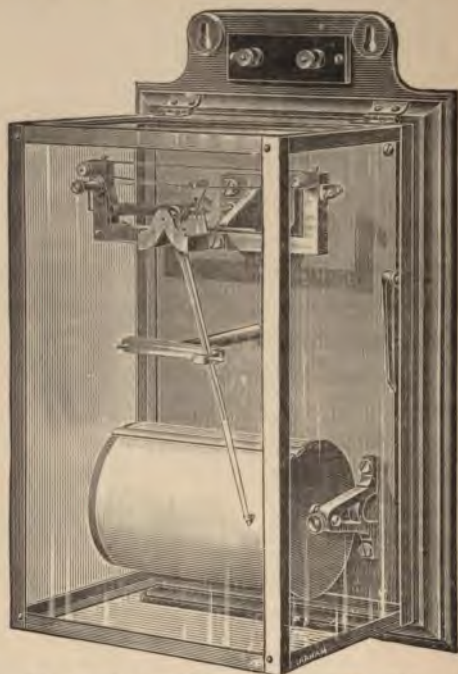


FIG. 29.—Holden's Self-recording Hot-Wire Voltmeter.

company should always take pains to ascertain for himself whether the pressure is constant. He can do this by the employment of a voltmeter, such as Lord Kelvin's electrostatic voltmeter, described in the previous Lecture, Fig. 8, or he can employ a self-registering voltmeter, of which there are many in use, which will record upon a drum covered with

paper a curve indicating the variation of pressure during the day and night. Two of these recording voltmeters which are now much in use are those devised by Prof. G. Mengarini and Captain Holden. In the latter instrument (*see* Fig. 29), a very fine wire is connected across between the two points between which pressure has to be determined. A small current flows through this wire and heats it, causing it to expand. The expansion of this wire is then made to move a lever, which, by means of a writing pen, records upon a drum driven by clockwork a curve, and any variation in the pressure or voltage is shown by the form of the line so drawn. The other instrument, due to Prof. Mengarini, is of a different type. In this apparatus a current taken from the two points between which the pressure has to be measured is passed through two coils of wire—one a fixed coil, and the other a coil suspended by two steel wires. The fixed and movable coils are so arranged (*see* Fig. 80) that when a current passes through the two it causes the movable coil to twist round, or tend to twist round, into a new position. This action of the current is resisted by the suspended wires. A writing pen is attached to the movable coil and draws a curve upon a revolving drum, as above described, and this curve records any variation in the voltage. The chart so drawn can then be examined at the end of every day, and a record is thus kept of the variation of pressure during the twenty-four hours. If the consumer finds that the light of his incandescent lamps varies a great deal, or if the life of the lamps is very irregular, it is well to have such a curve of pressure automatically drawn by a self-recording voltmeter for three or four days, and in this way to detect the limits within which the electric pressure varies. Consumers are very apt to lay the blame immediately upon the lamps for any variation in light or brevity in life, but as the electric pressure is partner with the lamp in producing the result, it is evident that the sins of the supply companies in giving a variable pressure ought not to

be laid upon the lamps. On the other hand, if the recording voltmeter shows a great uniformity in electric pressure, and if still the life of the lamps is not satisfactory, then either the lamps are in fault, or else the consumer is not using lamps of the proper voltage for his circuits. Before ordering lamps, it is well to ascertain also in this way what is the actual pressure

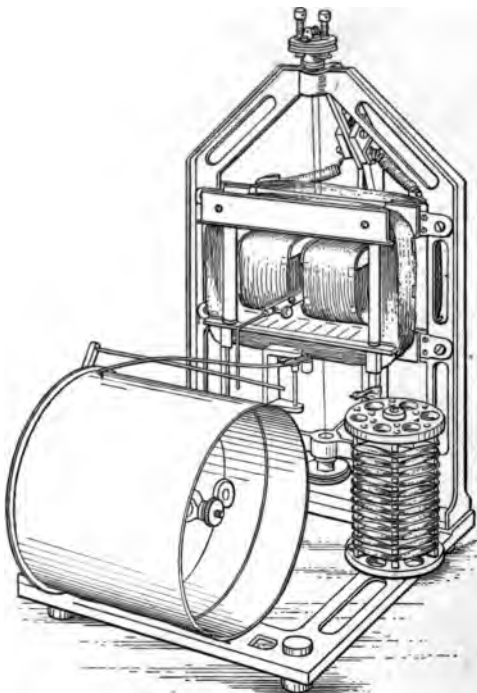


FIG. 30.—Mengarini's Self-recording Voltmeter.*

on the circuits, and then to order lamps of the proper marked volts for the particular pressure. Incandescent lamps which have been made and marked by the maker to work at 100 volts are not intended to be worked at 102 or 104, and, if they are, an abbreviation in the average life must certainly

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occur, or at any rate a more rapid blackening of the globe and diminution of the candle-power than would be the case if the consumer employed lamps of the proper voltage. If the supply is very irregular in pressure, it is never possible to obtain the same good results in the duration of the lamp and uniformity in brilliancy which can be obtained if the supply is maintained constant in pressure, at least within the limits of one or two volts.

Let us at this point note that, apart from the question of duration or cost, the physical value of a glow lamp is

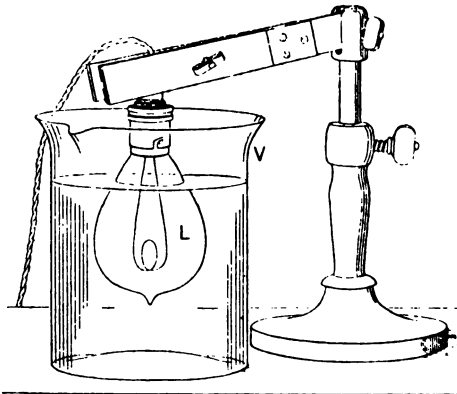


FIG. 31.—Glow Lamp Burning under Water.

measured by its utility as an energy-transforming device. An electric glow lamp is a machine for transforming energy from one form, viz., the energy of an electric current, into another form, viz., eye-affecting radiation. As explained in Lecture I., it transforms the electric energy into radiant energy, or energy of ether waves, only a fraction of which can impress the eye as light. This fraction varies from 8 to 7 per cent. This numerical expression for the "luminous efficiency" can be determined by suitable experiments in which the relative

amounts of luminous and dark heat sent out by a glow lamp are separately measured. If a glow lamp *L* (*see* Fig. 31) is immersed in water in a transparent vessel *V* the water absorbs or stops nearly all the dark or non-luminous radiation, but permits a large fraction of the luminous radiation or light to pass out. By photometering the lamp both when in and when out of the water the fraction of the luminous radiation which is absorbed can be determined. By measuring the rise of temperature of the water which is created by the absorption of the radiation, both when the luminous radiation is allowed to pass out and when it is wholly stopped by blackening the lamp, we can determine the ratio between the amount of the radiation which affects the eye as light and that which cannot affect it at all. A well-devised series of experiments of this kind have been carried out by Mr. E. Merritt, at Cornell University, U.S.A. The result has been to show that, on an average, about 5 per cent. of the energy supplied to the glow lamp as ordinarily used is utilized to make light, and the other 95 per cent. is non-effective as far as the eye is concerned. Hence, if a glow lamp is supplied with 60 watts and gives 16 candles light, only 3 watts are really transformed into light, and the mechanical value of the light so produced is rather less than a quarter of a watt per candle.

This luminous efficiency varies with the temperature of the filament. At a low red heat it is not more than 1 per cent., but increases to about 5, 6, or 7 per cent. at the normal working temperature. Thus, for an Edison 16-c.p. lamp Mr. Merritt found the following figures when the lamp was worked at various pressures so as to give different candle-power.

It will be seen that the luminous efficiency increases as the "watts per candle-power" decreases—that is, as the temperature of the filament increases.

The "luminous efficiency" must be carefully distinguished from that which is sometimes called "the efficiency" of the lamp, and which is the reciprocal of the watts per candle or the number of candles light yielded per horse-power or per watt expended in the filament.

Luminous Efficiency of a 16-c.p. Edison Glow Lamp.

Working volts.	Candle-power.	Power absorbed in watts.	Watts per candle-power.	Luminous radiation in watts.	Luminous efficiency as a percentage.
74.2	0.9	34.6	38.0	0.18	0.5 per cent.
91.6	4.8	56.2	12.0	0.68	1.2 "
97.3	7.3	64.6	9.0	1.13	1.7 "
100.3	8.9	69.3	7.8	1.62	2.3 "
107.6	14.6	81.6	5.6	2.97	3.6 "
109.3	16.3	84.4	5.2	4.57	5.4 "
124.1	38.2	115.4	3.0	7.46	6.5 "

It will be convenient for me at this stage to enter a little more fully into the changes that go on in an incandescent electric lamp during its life. We have pointed out above the causes which are at work to change the surface and ultimately to destroy the carbon filament. These causes commence to act from the very moment when the lamp begins to be used. If a carbon filament electric glow lamp is placed upon a circuit of perfectly constant pressure, and if the constants of the lamp—namely, the candle-power, the current, the resistance of the filament, the power taken up in watts, and the watts per candle-power—are measured at intervals, say of every 100 hours, it will be found that the changes which go on in the lamp are somewhat as follows:—During the life of the lamp there is a progressive rise in resistance, a progressive diminution in candle-power, and a progressive increase in watts per candle-power. An illustration of these changes is diagrammatically shown for certain forms of ordinary vacuum lamps in the curves given in Fig. 32. These curves

are deduced,* not from observations on one lamp alone, but from the average of a large number. If, instead of working the lamp at its marked volts or normal pressure, we raise the voltage, say, 10 per cent. above the normal, changes of the same character take place, only they are very much accelerated, and the result for a collection of lamps worked at 10 per cent. above the normal pressure is given in Fig. 83.

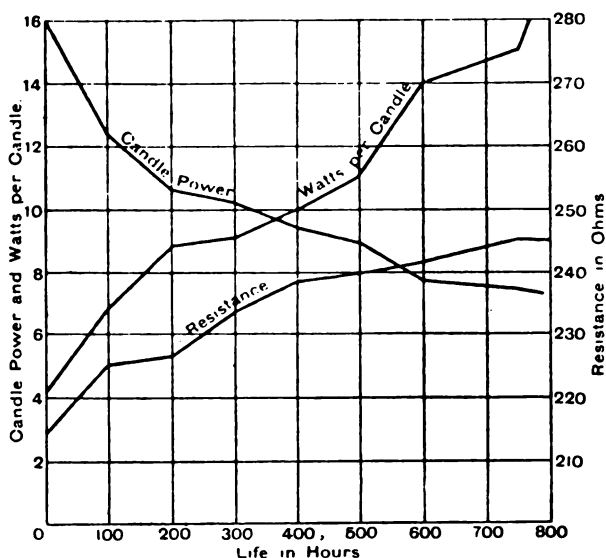


FIG. 32.—Curves showing the increasing resistance and diminishing candle-power of a Glow Lamp during its life.

On the other hand, if we desire to keep the candle-power always constant, we have to continually raise the pressure, and at the same time we are increasing the watts per candle and the resistance of the lamp. These progressive changes are illustrated in the two curves in Figs. 84 and 85, in the

* These curves are borrowed from a very instructive Paper by Prof. Nichols, of Cornell University, U.S.A., on "The Artificial Light of the Future."

first of which a 16-c.p. lamp was worked at continually-increasing pressures sufficient to keep the candle-power always constant, and in the second of which the candle-power was raised to four times its normal value and kept constant.

The value of a lamp to a user is not to be measured merely by the watts per candle-power it takes at the commencement

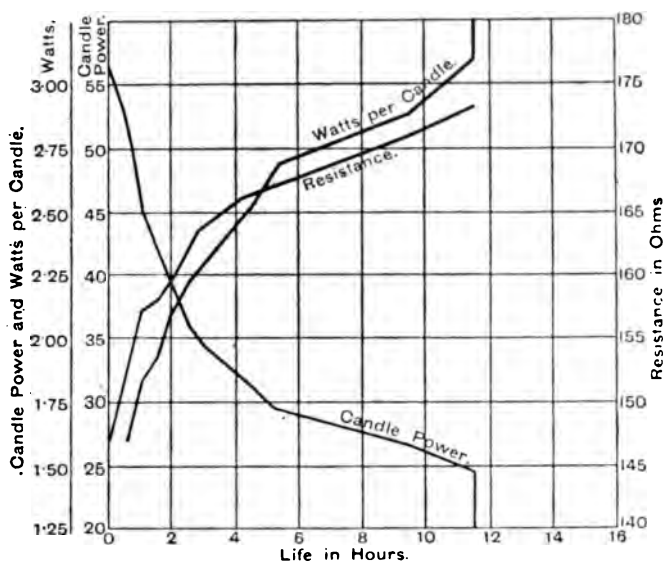


FIG. 33.—Curves showing the rapid decrease in candle-power and increase in resistance in a Glow Lamp worked at 10 per cent. above its normal pressure.

of its life, but also, to a large degree, by the extent to which this consumption of energy and production of light remains constant during the life of the lamp. This is best expressed by the percentage of the variation of the candle-power during the life of the lamp when taken at equal intervals of time. A 16-c.p. glow lamp, taking say 60 watts at the commencement of its life, but which after two or three hundred hours

in use is giving no more than 8 candles light, although taking the same power, is costing the consumer twice as much per candle at the end of this time as at the beginning. Some writers have, therefore, suggested that at the end of a certain period, when the candle-power has suffered a definite diminution in value, say, of more than 80 per cent., the lamp should be "smashed," and a new lamp put in its place. This, however, is not always advisable or necessary.

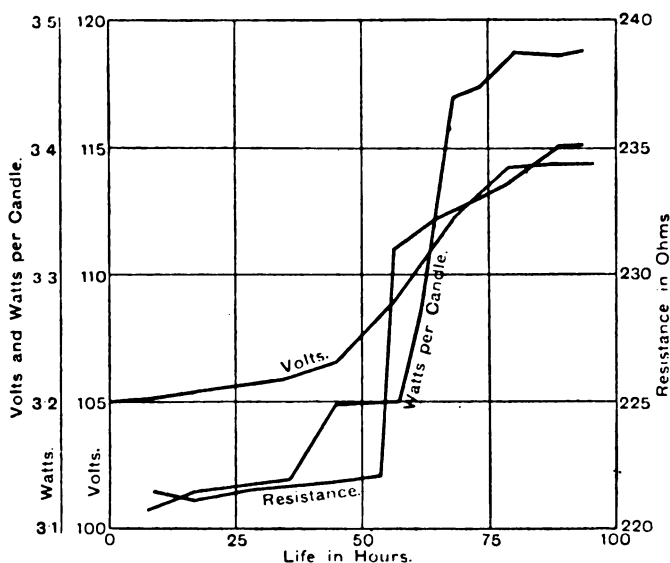


FIG. 34.—Curves showing the increase in Voltage required to keep the candle-power of a Glow Lamp constant.

If the lamp in the position in which it is being used is giving light enough for the purpose for which it is required, then, although its candle-power may be diminished, if the total power taken by the lamp is diminished also, the total sum which is being paid by the consumer for the energy supplied to that lamp is on the whole diminished, and therefore the lamp is costing the consumer less *on the whole* to maintain

in action than a new lamp would do if it replaced the old one, although at the same time the cost of the lamp per candle-light may be considerably increased. Generally speaking, users of electric glow lamps have an idea that the lamp which lasts the longest is necessarily the best. From what has been stated above, it will be seen that it is not merely long life, but constancy of candle-power, combined with high luminous efficiency and low cost, which are really the factors deter-

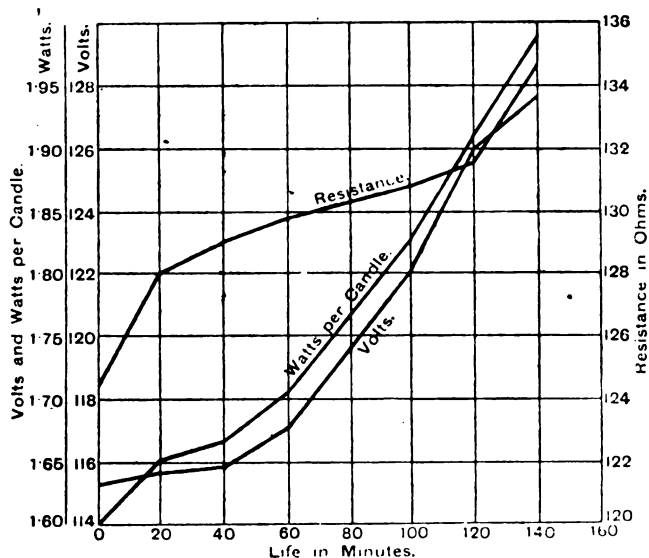


FIG. 35.—Curves showing the increasing Voltage necessary to keep the candle-power of a Glow Lamp constant and equal to four times its normal amount.

mining the value of the lamp to the consumer. The duration of the lamp is intimately connected, apart from accidents, with the temperature at which the carbon filament is maintained, and this is determined by the watts per candle-power being expended in the lamp. Makers of glow lamps sometimes make statements which lead the uninitiated to believe

that there is a certain "efficiency" or peculiarly low value of the watts per candle-power belonging especially to a certain make of lamp. It should be, however, clearly understood that any glow lamp may be worked at values of the watts per candle-power varying over very wide limits, but experience has shown that carbon filaments, as at present made, cannot be worked at temperatures higher than those corresponding to about 2 to $2\frac{1}{2}$ watts per candle-power without very rapidly being destroyed.

A number of tests have been made by Messrs. Siemens and Halske on lamps of various types, and in which groups of these lamps were worked at different watts per candle-power from 2 to 4 and upwards. The lamps were measured at intervals of 100 hours, and the diminution in the observed candle-power expressed as a percentage of the original candle-power was laid down in the form of curves. The results of these tests seemed to show that, on the whole, lamps worked at from 3 to $3\frac{1}{2}$ watts per candle-power had a greater average constancy in candle-power than those worked at a lower number of watts per candle-power. That is to say, if we take a 16-c.p. glow lamp and work its filament at such a temperature that it is taking on the whole 32 watts, or 2 watts per candle-power, it may take less power to begin with, but it will not preserve nearly the same constancy of candle-power as if it had been worked at such a temperature that it absorbed from 48 to 56 watts at the commencement of its career; in other words, absorbed 3 to $3\frac{1}{2}$ watts per candle-power. No hard and fast rule, however, can be laid down about these matters. No deductions can be made which have any great value from tests made upon a very small number of lamps. As I have previously stated, the lives of electric lamps are in this respect like human lives, and just as the insurance companies have to draw their deductions and to make their calculations for

life insurance on the results of figures obtained from a very large number of human lives, so inferences made from the results of statistical tests carried out on incandescent lamps are only of use if they are made on a very large number of lamps. Broadly speaking, however, the duration of an incandescent lamp is dependent upon the electric pressure at which the lamp is worked, and if a lamp is marked to be used on a 100-volt circuit, then if it is used on a circuit at a pressure of 105 volts, its life would be greatly abbreviated, probably to one quarter or one fifth of the time which it would last if worked at the normal pressure. On the other hand, if it is worked at a pressure of 95 volts, its duration, accidents apart, will be very greatly increased. It is useful, therefore, to have, under these circumstances, a general knowledge of what would be the variation in candle-power if the same lamp is employed at different voltages. The tables on the next page give the candle-power yielded by two particular electric glow lamps when worked at varying pressures, the first table being for a 100-volt 16-c.p. lamp, and the second table being for a 100-volt 8-c.p. lamp. A very cursory examination of these tables will show that the light given by the lamp varies enormously with the pressure, and that a very small diminution of pressure suffices to deprive the user of a very large fraction of the candle-power for which the lamp is constructed.

It may not be inappropriate at this point to call the attention of users of incandescent lamps to the fact that price is not the only factor to be considered in connection with the choice of a glow lamp. Consideration should also be given to the rate at which the candle-power decays when constant pressure is kept on the terminals of the lamp, as well as to the initial efficiency, and the *useful* life of the lamp. The useful life of the lamp, so far as most consumers are concerned, must be considered to be limited to that period of the total lamp life in which its candle-power is not diminished by

lends itself so readily to decorative purposes. In order that it may be employed in this sense to the best advantage, certain general guiding principles have to be held in view and followed. Unfortunately, this is not always done, and the result is that the effectiveness of the light, from the artistic point of view, is greatly diminished.

The first principle which should be allowed to guide us in the production of an artistic effect by electric lighting is that, while a proper and sufficient illumination is thrown upon the surfaces to be illuminated, the source of light must be itself, as far as possible, concealed. The illuminating power of a light for the purpose of vision does not depend merely upon the candle-power of the lamp, but it depends upon the amount of light which is received by the eye from the surfaces from which it is reflected. The pupil of the eye is capable of being varied over wide limits, probably from about four square millimetres, or the 150th part of a square inch, to 70 square millimetres or the tenth part of a square inch. This expansion or contraction of the pupil of the eye is effected by muscles over which the will has no control, and is termed in physiology a "reflex action." If the eye has been kept in darkness for some time, then the pupil becomes expanded to its greatest extent. Under these conditions, if the eye is opened, and especially if it is exposed to a bright light, this light falling on the retina causes an immediate muscular contraction of the pupil to be effected. It is very easy to see this adjusting process going on in the eye by shutting the eyes for a few minutes and then opening them in a bright light whilst a hand mirror is held before the face so as to examine the pupils of the eyes. The pupils will then be seen to be rapidly contracting in area immediately the eyes are opened. Hence, if the eye is turned upon a brilliant line of light, the contraction of the pupil which immediately sets in imposes a limitation upon the amount of light which can enter the eye; and if the

eye is turned immediately towards a dull, badly illuminated surface, the pupil is not at once suddenly expanded again.

One part, at any rate, of the injurious effect produced by trying to read or write in the twilight is due to the strain produced in the eyes by the muscular effort thus demanded in the eye. When, therefore, we enter a room in which there are a number of bright lights, such, for instance, as a room in which incandescent lamps are being employed, the filaments of which can be seen, the moment that the eye is turned upon these bright lines of light, muscular contraction of the pupil sets in, and on turning the eye away again to other less illuminated surfaces, the retina of the eye does not receive sufficient light to observe details. In common language, the eye is "dazzled." There is a certain fascination about brilliant points or lines of light which captivates and attracts the eye even against the will.

It will thus be seen that the presence of brilliant lines or points of light in a room has a tendency to keep the pupil of the eye in a state of partial contraction, which unfits it for obtaining the best visual effect when turned upon surfaces less illuminated. Some eyes are especially sensitive in this respect. If, however, the lamp globes are made of frosted glass, or in other ways protected, so that the image of the incandescent filament cannot be directly thrown upon the retina, the actual visual effect may be increased in spite of the fact that such frosted globes or screens may cut off from 30 to 50 per cent. of the light. This is a common experience with everybody who tries to read or write in the neighbourhood of a brilliant incandescent lamp. By covering the globe with tissue paper, or with a ground glass or porcelain shade, although a diminution to a large extent is caused in the total light actually distributed from the lamp, yet, nevertheless, we are able to see better and more comfortably by it. The process, however,

which is commonly adopted of cutting off a large proportion of the light by a semi-opaque globe is a wasteful one, because there can be no object in making light merely to absorb it.

The proper method is to so place the incandescent lamp that no light from the filament can directly enter the eye, but so that all the light sent out from the filament shall be reflected to the eye from the objects to be seen, such as the walls of the room and the various objects in the room,



FIG. 36.—View of the Interior of a Dining Room, Illuminated with Shaded Electric Candles.

and that the light shall reach the eye only after reflection from these surfaces. Bare or uncovered lamps, and especially if suspended half way down a room or on a level with, or a little above the eye, are a crude and exceedingly disagreeable method of illumination.

In all cases where incandescent lighting is carried out by those who understand the proper methods of using it, so as to pro-

duce the best artistic effects, the guiding principle is followed of so placing the lamps that the whole of the light emitted from them only reaches the eye after reflection from surrounding objects. This result can be attained in a variety of ways. The lamps may be placed in shades so as to throw the light upwards upon the ceiling of the room or upon the walls, from which surfaces it is ultimately reflected down on to the objects in the room. Wherever highly decorated and light ceilings exist in a room to be lit, this is a very effective method of distributing the light. In Figs. 36, 37, 38, 39, and 40, are shown photographs of interiors so illuminated. It is always possible to arrange glow lamps protecting them by metallic or other ornamental shields, so so as to obtain this desired effect. A natural shell, properly supported, forms a very effective screen and reflecting surface. The white pearly interior of the shell acts as a reflecting surface, and if the shell is very slightly translucent, then a pleasing effect is produced by the small and diffused light which passes through.

The second important principle of decorative electric lighting is to distribute the light properly and to prevent the concentration of light in large masses, thereby avoiding the production of harsh shadows. Everyone is familiar with the exceedingly disagreeable effect which is produced by sharp and strong shadows. A soft gradation of light and shade is essential in the production of an artistic effect in a room. The inartistic productions of many amateur photographers, especially in the region of portraiture, are much more due to the fact that they have not the means at their disposal for producing the right effects of light and shade on their subjects, than to any defects in their photographic manipulation. Any concentration of the light in large masses, in public or private rooms,



FIG. 37.—View of the Interior of a Drawing Room Lighted Electrically with Screened Lamps.



FIG. 38.—View of a Ball Room Illuminated with Shaded Electric Candles in Chandeliers.



FIG. 39.—View of a Reception Room Illuminated with Inverted Incandescent Lamps throwing Light upwards.



FIG. 40.—View of a Drawing Room Illuminated with Shaded Electric Candles.

and especially when thrown downwards, invariably produces disagreeable shadows upon the face; and, therefore, in places like ball-rooms or drawing-rooms no such process of employing downwardly-suspended incandescent lamps should ever be permitted if it is desired to obtain the best results. All artists know the immense importance of obtaining proper lighting by diffused light, and the northern aspect which they invariably select for their studios is determined by this consideration. It is essential that the light should be so distributed in small units, in order that there shall be no sharp shadows cast upon any object. A simple test for the effectiveness of the distribution of light in any room may be obtained as follows: Take a white card or a sheet of white writing paper and hold it horizontally about the level of the eyes, then hold a pencil or other small rod vertically on the card, and note if any marked shadow of the pencil is thrown in any direction: if it is, then the light is insufficiently diffused, and the lamps ought to be re-arranged. The best effects as to distribution of light are generally obtained by the employment of 5 candle-power lamps. These glow lamps are constructed with bulbs of semi-opaque glass having the shape of a candle flame. (see Fig. 21). These candle lamps are carried on the extremity of an artificial porcelain candle, so as to resemble in some degree an ordinary wax candle alight. These lamps are then arranged in groups on chandeliers or brackets, being protected in many cases from direct visibility by plain or ornamental shades. In Figs. 36 and 41 are shown photographic illustrations of rooms which are lit by such candle lamps, and in which the light from these is thrown up against the walls of the room, the candle lamps not being themselves directly visible.

A third great guiding principle of artistic lighting is to proportion the light to the nature of the surfaces from which it is reflected. We require light in our living rooms to see the



FIG. 41.—Drawing Room Illuminated with Electric Lamps, so shaded that no direct view of the lamp filament can be obtained.



FIG. 42.—Interior of Freemason's Hall Illuminated by Shaded Incandescent Lamps on Frieze and Ceiling.

pictures, decorations, objects, and, above all, to see one another; and we can only see these objects by the light which they reflect back to the eye. The aim should, therefore, be to have sufficient but not excessive incident light. It should be borne in mind that the different surfaces with which the walls of rooms are covered have very different reflective powers. Dull walls absorb eighty per cent. of the light falling on them, and reflect only about twenty per cent. of the light; this is the case, for instance, with walls covered with dark oak panelling or dull-coloured paint or paper. Ordinary light tints of paper or paint may reflect from forty to sixty per cent.; clean white surfaces, such as white painted and varnished wood or plaster, may reflect as much as eighty per cent., and mirrors from eighty to ninety per cent.

The following figures have been given by Dr. Sumpner for the reflecting power of various surfaces:—

Table of Reflecting Powers of Various Surfaces.

White blotting paper 82 per cent.	Plane deal (clean) 40-50 per cent.
Ordinary foolscap ... 70 " "	Plane deal (dirty) ... 20 " "
Newspapers 50-70 " "	Yellow painted wall
Yellow wall paper ... 40 " "	(dirty) 20 " "
Blue paper 25 " "	Yellow painted wall
Dark brown paper ... 13 " "	(clean) 40 " "
Deep chocolate paper 4 " "	Black cloth 1·2 " "
	Black velvet '4 " "

Everyone is familiar with the way in which the laying of a white table-cloth brightens up a dark dining room, or in which the substitution of white for dark curtains lightens up a room. The reason for this is obvious from the above table. Hence, rooms with dark oil paintings, dark woodwork, hangings, curtains, or portières require relatively much more illumination per square foot than rooms with very light decorations, water-colour paintings, mirrors, &c. Experience has shown that in a room of the latter description an illumi-

nation is required which is equal to an expenditure in the lamp of about $\frac{3}{4}$ or 1 watt per square foot of floor surface, on the assumption that the lamps are placed, on an average, about 8 or 9 feet above the floor. A room of the former description, such as a picture gallery with dark oak walls, may require from 2 to 3 watts per square foot of floor surface. These figures, however, cannot be taken too absolutely. Experience is the only guide as to the amount of light to be placed in a room. As a broad general rule, 100 square feet of floor surface will be barely illuminated by one 16-c.p. lamp placed about 8 feet above the floor. It will be well illuminated by two such lamps placed 8 feet above the floor, and brilliantly illuminated by four such lamps. In other words, one 16-c.p. lamp per 100 square feet of floor surface is hardly sufficient, one 16-c.p. lamp per 50 square feet of floor surface is fairly good, and one for every 25 square feet of floor surface is brilliant. These figures are given on the assumption that the lamp is placed about 8 feet above the floor, and that the floor surface is fairly reflective. The lighting of pictures is a very special study, and can only be properly undertaken by those who have given considerable attention to the method of placing and using incandescent lamps. The mere illumination of a room by a number of lamps hung head downwards by cord pendants from the ceiling, a method which is not unusually adopted, is the most crude and imperfect method of applying the electric light, and ought never to be permitted in any case where the smallest decorative effect is really desired.

The brief limits of a lecture will not permit us to enlarge at greater length on this part of the subject, but it opens up a wide field for the exercise of the inventive and æsthetic faculties.

Before leaving the subject of incandescent lamps it will not be without interest to refer briefly to the investigations

which have been made into a curious effect, found to exist in lamps which have a third or idle metallic wire sealed through the glass. Our starting-point for this purpose is a discovery made by Mr. Edison in 1884, and which received careful examination at the hands of Mr. Preece in the following year,* and by the author more recently. Here is the initial experiment. A glow-lamp having the usual horseshoe-shaped carbon (Fig. 43) has a metal plate held on a platinum wire sealed through the glass bulb. This plate is so fixed that it stands up between the two sides of the carbon arch without touching either of them. We shall illuminate the lamp by a

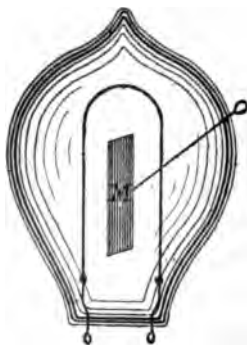


FIG. 43.—Glow Lamp having insulated metal middle plate *M* sealed into the bulb to exhibit the “Edison effect.”

continuous current of electricity, and for brevity's sake speak of that half of the loop of carbon on the side by which the current enters it as the positive leg, and the other half of the

* Mr. Preece's interesting Paper on this subject is published in the *Proceedings of the Royal Society* for 1885, p. 219. See also *The Electrician*, April 4, 1885, p. 436, “On a Peculiar Behaviour of Glow-lamps when Raised to High Incandescence.” The following statements of experiments are chiefly taken from a Friday evening discourse given by the author at the Royal Institution, February 14, 1890, on “Problems in the Physics of an Electric Lamp,” and from a Paper on “Electric Discharge between Electrodes at different Temperatures in Air and in High Vacua.” (*Proceedings of the Royal Society*, 1890.)

loop as the negative leg. The diagram in Fig. 45 shows the position of the plate with respect to the carbon loop. There is a distance of half-an-inch, or in some cases many inches, between either leg of the carbon and this middle plate. Setting the lamp in action, I connect a sensitive galvanometer between the middle plate and the *negative terminal* of the lamp, and you see that there is no current passing through the instrument. If, however, I connect the terminals of my galvanometer to the middle plate, and to the *positive electrode* of the lamp, we find a current of some milliamperes is passing through it. The diagrams in Fig. 45 show the mode of con-

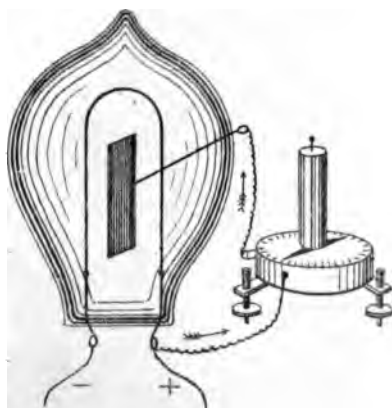


FIG. 44.—Sensitive Galvanometer connected between the middle plate and positive electrode of a Glow Lamp, showing current flowing through it when the lamp is in action ("Edison effect").

nection of the galvanometer in the two cases. This effect, which is often spoken of as the "Edison effect," clearly indicates that an insulated plate so placed in the vacuum of a lamp in action is brought down to the same potential or electrical state as the negative electrode of the carbon loop. On examining the direction of the current through the galvanometer we find that it is equivalent to a flow of negative electricity taking place through it *from* the middle plate to the

positive electrode of the lamp. A consideration of this fact shows us that there must be some way by which negative electricity gets across the vacuous space from the negative leg of the carbon to the metal plate, whilst at the same time a negative charge cannot pass from the metal plate across to the positive leg.

Before passing away from this initial experiment, I should like to call your attention to a curious effect at the moment when the lamp is extinguished. Connecting the galvanometer, as at first, between the middle plate and the negative electrode of the lamp, we notice that, though made

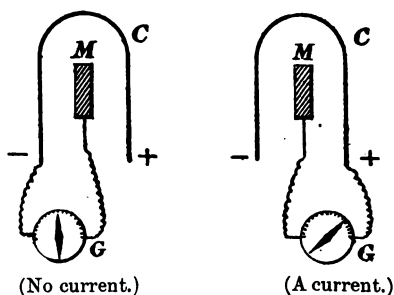


FIG. 45.—Mode of connection of galvanometer *G* to middle plate *M* and carbon horseshoe-shaped conductor *C* in the experiment of the "Edison effect."

highly sensitive, the galvanometer indicates no current flowing through it whilst the lamp is in action. Switching off the current from the lamp produces, as you see, a violent kick or deflection of the galvanometer, indicating a sudden rush of current through it.

In endeavouring to ascertain further facts about this effect one of the experiments which early suggested itself was directed to determine the relative effects of different portions of the carbon conductor. Here is a lamp (*see* Fig 46) in which

one leg of the carbon horseshoe has been enclosed in a glass tube the size of a quill, which shuts in one half of the carbon. The bulb contains, as before, an insulated middle plate. If we pass the actuating current through this lamp in such a direction that the covered or sheathed leg is the *positive* leg, we find the effect existing as before. A galvanometer connected between the plate and positive terminal of the lamp yields a strong current, whilst if connected between the negative terminal and the middle plate there is no current at all. Let us, however, reverse the current through the lamp so that the

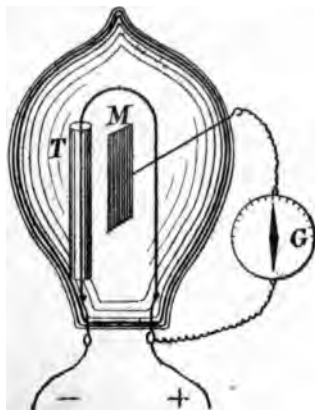


FIG. 46.—Glow Lamp having negative leg of carbon enclosed in glass tube *T*, the "Edison effect" being thereby annulled or greatly diminished.

shielded or enclosed leg is now the negative one, and the galvanometer is able to detect no current, whether connected in one way or in the other. We establish, therefore, the conclusion that it is the negative leg of the carbon loop which is the active agent in the production of this "Edison effect," and that, if it is enclosed in a tube of either glass or metal, no current is found flowing in a galvanometer connected between the positive terminal of the lamp and this middle collecting plate

Another experiment which confirms this view is as follows: The lamp (Fig. 47) has a middle plate, which is provided with a little mica flap or shutter on one side of it. When the lamp is held upright the mica shield falls over and covers one side of the plate, but when it is held in a horizontal position the mica shield falls away from the front of the plate and exposes it. Using this lamp as before, we find that, when the positive leg of the carbon loop is opposite to the shielded face of the plate, we get the "Edison effect" as before in any position of the lamp. Reversing the lamp current, and making

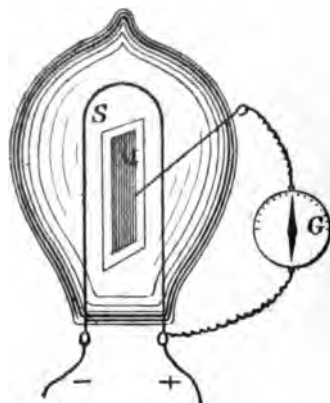


FIG. 47.—Glow Lamp having mica shield *S* interposable between middle plate *M* and negative leg of carbon, thereby diminishing the "Edison effect."

that same leg the *negative* one, we find that, when the lamp is so held, the metal plate is shielded by the interposition of the mica, and the galvanometer current is very much less than when the shield is shaken on one side and the plate exposed fully to the negative leg.

At this stage it will perhaps be most convenient to outline briefly the beginnings of a theory which may be proposed to reconcile these facts, and leave you to judge how far the

subsequent experiments seem to confirm this hypothesis. Very briefly, this theory is as follows: From all parts of the incandescent carbon loop, but chiefly from the negative leg, carbon molecules are being projected which carry with them, or are charged with, negative electricity. In a few moments a suggestion will be made to you which may point to a possible hypothesis on the manner in which the molecules acquire this negative charge. Supposing this, however, to be the case, and that the bulb is filled with these negatively-charged molecules, what would be the result of introducing into their midst a conductor such as this middle metal plate which is charged positively? Obviously, they would all be attracted to it and discharge against it. Suppose the positive charge of this conductor to be continually renewed, and the negatively-charged molecules continually supplied—which conditions can be obtained by connecting the middle plate to the positive electrode of the lamp—the obvious result will be to produce a current of electricity flowing through the wire or galvanometer by means of which this middle plate is connected to the positive electrode of the lamp. If, however, the middle plate is connected to the negative electrode of the lamp, the negatively-charged molecules can give up no charge to it, and produce no current in the interpolated galvanometer. We see that, on this assumption, the effect must necessarily be diminished by any arrangement which prevents these negatively-charged molecules from being shot off the negative leg or from striking against the middle plate.

Another obvious corollary from this theory is that the "Edison effect" should be annihilated if the metal collecting plate is placed at a distance from the negative leg much greater than the mean free path of the molecules.

Here are some experiments which confirm this deduction. In this bulb (Fig. 48) the metal collecting plate, which is to be

connected through the galvanometer with the positive terminal of the lamp, is placed at the end of a long tube opening out of and forming part of the bulb. We find the "Edison effect" is entirely absent, and that the galvanometer current is zero. We have, as it were, placed our target at such a distance that the lowest range molecular bullets cannot hit it, or at least that but very few of them do so. Here, again, is a lamp in which the plate is placed at the extremity of a tube opening out of the bulb, but bent at right angles (Fig. 49). We find in this case, as first discovered by Mr. W. H. Preece, that there is no "Edison effect." Our molecular marksman

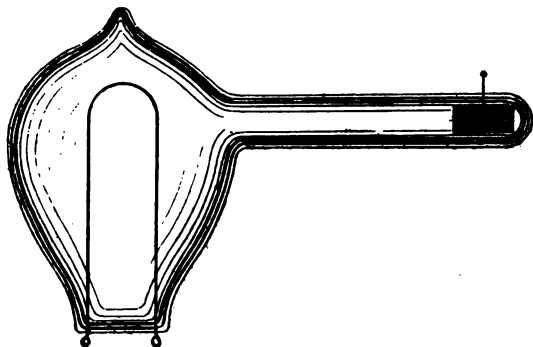


FIG. 48.—Collecting plate placed at end of tube, 18in. in length, opening out of bulb.

cannot shoot round a corner. None of the negatively-charged molecules can reach the plate, although that plate is placed at a distance not greater than would suffice to produce the effect if the bend were straightened out. Following out our hypothesis into its consequences would lead us to conclude that the material of which the plate is made is without influence on the result, and this is found to be the case.

We should expect also to find that the larger we make our plate, and the nearer we bring it to the negative leg of the carbon, the greater will be the current produced in a circuit

connecting this plate to the positive terminal of the lamp. I have before me a lamp with a large plate placed very near the negative leg of the carbon of a lamp, and we find that we can collect enough current from these molecular charges to work a telegraph relay and ring an electric bell. The current which is now working this relay is made up of the charges collected by the plate from the negatively-charged carbon molecules, which are projected against it from the negative leg, across the highly perfect vacuum. I have tried experiments with lamps in which the collecting plate is placed in all kinds

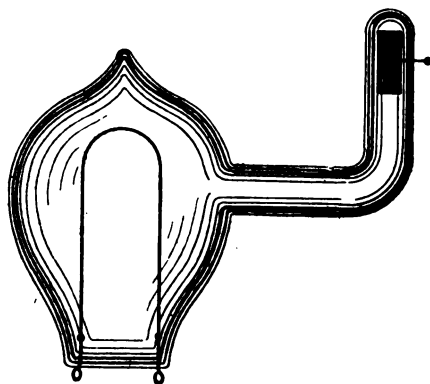


FIG. 49.—Collecting plate placed at end of elbow tube opening out of bulb.

of positions, but the results may all be summed up by saying that the greatest effects are produced when the collecting plate is as near as possible to the base of the negative end of the loops, and, as far as possible, encloses, without touching, the carbon conductor. It is not necessary to make more than a passing reference to the fact that the magnitude of the current flowing through the galvanometer when connected between the middle plate and the positive terminal of the lamp often "jumps" from a low to a high value, or *vice versa*, in a remarkable manner, and that this sudden change in the current can be produced by bringing strong magnets near the outside of the bulb.

Let us now follow out into some other consequences the hypothesis that the interior of the bulb of a glow lamp, when in action, is populated by flying crowds of carbon atoms all carrying a negative charge of electricity. Suppose we connect our middle collecting plate with some external reservoir of electric energy, such as a Leyden jar, or with a condenser equivalent in capacity to many hundreds of Leyden jars, and let the side of the condenser which is charged positively be first placed in connection through a galvanometer with the middle plate (*see* Fig. 50), whilst the negative side is placed

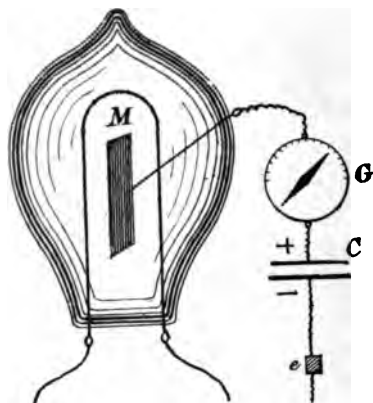


FIG. 50.—Charged condenser *C* discharged by middle plate *M*, when the positively-charged side of condenser is in connection with the plate and the other side to earth *e*.

in connection with the earth. Here is a condenser of two microfarads capacity so charged and connected. Note what happens when I complete the circuit, and illuminate the lamp by passing the current through its filament. The condenser is at once discharged. If, however, we repeat the same experiment, with the sole difference that the negatively-charged side of the condenser is in connection with the middle plate, then there is no discharge.

These very interesting experimental results may also be regarded from another point of view. In order that the condenser may be discharged, as in the first case, it is essential that the negatively-charged side of the condenser shall be in connection with some part of the circuit of the incandescent carbon loop. This experiment with the condenser discharged by the lamp may be, then, looked upon as an arrangement in which the plates of a charged condenser are connected respectively to an incandescent carbon loop and to a cool metal plate, both being enclosed in a highly-vacuous space; and it appears that the discharge takes place when the incandescent conductor is the negative electrode of this arrangement, but not when the cooler metal plate is the negative electrode of the charged condenser. The negative charge of the condenser can be carried across the vacuous space from the hot carbon to the colder metal plate, but not in the reverse direction.

This experimental result led me to examine the condition of the vacuous space between the middle metal-plate and the negative leg of the carbon loop in the case of the lamp employed in our first experiment. Let us return for a moment to that lamp. I join the galvanometer between the middle plate and the negative terminal of the lamp, and find, as before, no indication of a current. The metal plate and the negative terminal of the lamp are at the same electrical potential. In the circuit of the galvanometer we will insert a single galvanic cell, having an electromotive force of rather over one volt. In the first place let that cell be so inserted that its negative pole is in connection with the middle plate, and its positive pole in connection through the galvanometer with the negative terminal of the lamp (*see* Fig. 51). Regarding the circuit of that cell alone, we find that it consists of the cell itself, the galvanometer wire, and that half-inch of highly vacuous space between the hot carbon conductor and the middle plate. In that circuit the cell cannot send any

sensible current at all, as it is at the present moment connected up. But if we reverse the direction of the cell, so that its positive pole is in connection with the middle plate, the galvanometer at once gives indications of a very sensible current. This highly-vacuous space, lying between the middle metal plate on the one hand and the incandescent carbon on the other, possesses a kind of unilateral conductivity, in that it will allow the current from a single galvanic cell to pass one way but not the other. It is a very old and familiar fact, that in order to send a current from a battery through a highly-rarefied gas by means of metal electrodes

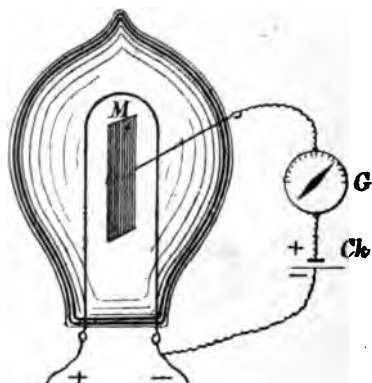


FIG. 51.—Current from Clark cell *Ck* being sent across vacuous space between negative leg of carbon and middle plate *M*. Positive pole of cell in connection with plate *M* through galvanometer *G*.

the electromotive force of the battery must exceed a certain value. Here, however, we have indication that if the negative electrode by which that current seeks to enter the vacuous space is made incandescent the current will pass at a very much lower electromotive force than if the electrode is not so heated.

A little consideration of the foregoing experiments led to the conclusion that, in the original experiment, as devised by Mr.

Edison, if we could by any means render the middle plate very hot, we should get a current flowing through a galvanometer when it is connected between the middle plate and the negative electrode of the carbon. This experiment can be tried in the manner now to be shown. Here is a bulb (Fig. 52) having in it two carbon loops: one of these is of ordinary size, and will be rendered incandescent by the current from the mains. The other loop is very small, and will be heated by a well-insulated secondary battery. This smaller incandescent loop shall be employed just as if it were a middle metal plate. It is, in fact, simply an incandescent middle conductor. On repeating

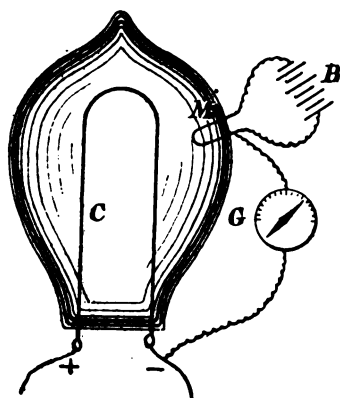


FIG. 52.—Experiment showing that when the “middle plate” is a carbon loop rendered incandescent by insulated battery *B*, a current of negative electricity flows from *M* to the positive leg of main carbon *C* across the vacuum.

the typical experiment with this arrangement, we find that the galvanometer indicates a current when connected between the middle loop and either the positive or the negative terminal of the main carbon. I have little doubt but that, if we could render the platinum plate in our first-used lamp incandescent by concentrating on it from outside a powerful beam of radiant heat, we should get the same result.

A similar set of results can be arrived at by experiments with a bulb constructed like an ordinary vacuum tube, and having small carbon loops at each end instead of the usual platinum or aluminium wires. Such a tube is now before you (*see* Fig. 53), and will not allow the current from a few cells of a secondary battery to pass through it when the carbon loops are cold. If, however, by means of well-insulated secondary batteries, we render both of the carbon loop electrodes highly incandescent, a single cell of a battery is sufficient to pass a very considerable current across that vacuous space, provided the resistance of the rest of the circuit is not large. We may embrace the foregoing facts by saying that if the electrodes, but especially the negative electrode, which form the means of ingress and egress of a

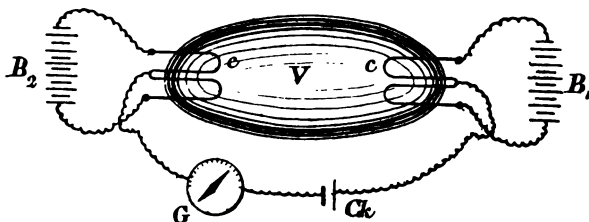


FIG. 53.—Vacuum tube having carbon loop electrodes, e c , at each end, rendered incandescent by insulated batteries, B_1 B_2 , showing current from Clark cell, Ck , passing through the high vacuum when the electrodes are incandescent.

current into a vacuous space, are capable of being rendered highly incandescent, and if at that high temperature they are made to differ in electrical potential by the application of a very small electromotive force, we may, under these circumstances, get a very sensible current through the rarefied gas. If the electrodes are cold, a very much higher electromotive force will be necessary to get the discharge or current through the space. These facts have been made the subject of elaborate investigation by Hittorf and Goldstein, and more recently by Elstér and Geitel. It is to

Hittorf, I believe, that we are indebted for the discovery of the fact that by heating the negative electrode we greatly reduce the apparent resistance of a vacuum.

Permit me now to pave the way by some other experiments for a little more detailed outline of the manner in which I shall venture to suggest these negative molecular charges are bestowed. This is really the important matter to examine. In seeking for some probable explanation of the manner in which these wandering molecules of carbon in the glow-lamp bulb obtain their negative charges, I fall back for assistance upon some facts discovered by the late Prof. Guthrie. He showed some years ago new experiments on the relative powers of incandescent bodies for retaining positive and negative charges. One of the facts he brought forward* was that a bright red-hot iron ball, well insulated, could be charged negatively, but could not retain for an instant a positive charge. He showed this fact in a way which it is very easy to repeat as a lecture experiment. Here is a gold-leaf electroscope, to which we will impart a positive charge of electricity, and project the image of its divergent leaves on the screen. A poker, the tip of which has been made brightly red hot, is placed so that its incandescent end is about an inch from the knob of the electroscope. No discharge takes place. Discharging the electroscope with my finger, I give it a small charge of negative electricity, and replace the poker in the same position. The gold leaves instantly collapse. Bear in mind that the extremity of the poker when brought in contiguity to the knob of the charged electroscope becomes charged by induction with a charge of the opposite sign to that of the charge of the electroscope, and you will at once see that this experiment confirms Prof. Guthrie's statement, for the negatively-charged electroscope induces a positive charge on the incandescent iron, and this charge

* "On a New Relation between Electricity and Heat," *Phil. Mag.*, Vol. XLV., 1873, p. 308.

cannot be retained. If the induced charge on the poker is a negative charge it is retained, and hence the positively-charged electroscope is not discharged, but the negatively-charged electroscope at once loses its charge.

Pass in imagination from iron balls to carbon molecules. It may be asked whether it is a legitimate assumption to suppose the same fact to hold good for them, and if a hot carbon molecule or small carbon mass just detached from an incandescent surface behaves in the same way, and has a greater grip for negative than for positive charge? If this can possibly be assumed, we can complete our hypothesis as follows: Consider a carbon molecule or small congerie of molecules

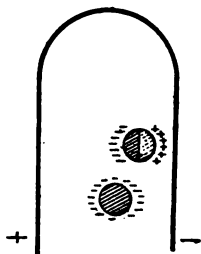


FIG. 54.—Rough diagram illustrating a theory of the manner in which projected carbon molecules may acquire a negative charge.

just set free by the high temperature from the negative leg of the incandescent carbon horseshoe. This small carbon mass finds itself in the electrostatic field between the branches of the incandescent carbon conductor (Fig. 54). It is acted upon inductively, and, if it behaves like the hot iron ball in Prof. Guthrie's experiment, it loses its positive charge. The molecule then being charged negatively is repelled along the lines of electric force against the positive leg. The forces moving it are electric forces, and the repetition of this action would cause a torrent of negatively-charged molecules to pour across from the negative to the positive side of the carbon horseshoe.

If we place a metal plate in their path which is in conducting connection with the positive electrode of the lamp carbon, the negatively-charged molecules will discharge themselves against it. A plate so placed may catch more or less of the stream of charged molecules which pour across between the heels of the carbon loop. There are many extraordinary facts, which as yet I have been able only imperfectly to explore, which relate to the sudden changes in the direction of the principal stream of these charged molecules, and to their guidance under the influence of magnetic forces. The above rough sketch of a theory must be taken for no more than it is worth, viz., as a working hypothesis to suggest further experiments.

It has been noticed that such a middle plate placed between the legs of the carbon horseshoe becomes black soonest on the side facing the negative leg of the carbon. It has also been noticed that, in the case of lamps with treated filaments, the surface of the deposit assumes a dead lamp-black appearance soonest on the negative leg. These observations tend to support the contention that there is an ejection, or more rapid ejection, of carbon from the negative leg of the carbon horseshoe. Much has yet to be done before the full meaning of this "Edison effect" is unravelled; but, as far as it has yet been examined, the following summary of facts includes most of those as yet observed.

If a platinum wire is sealed through the glass bulb of an ordinary carbon filament lamp, and carries at its extremity a metal plate so placed as to stand up between the legs of the carbon horseshoe without touching either of them, then, when the lamp is actuated by a continuous current, it is found that:—

(1.) This insulated metal plate is brought down instantly to the potential of the base of the negative leg of the carbon, and no sensible potential difference exists between the insulated

metal plate and the negative electrode of the lamps, whether the test be made by a galvanometer, by an electrostatic voltmeter, or by a condenser.

(2.) The potential difference of the plate and the positive electrode of the lamp is exactly the same as the working potential difference of the lamp electrodes, provided this is measured electrostatically, *i.e.*, by a condenser, or by an electrostatic voltmeter taking no current; but, if measured by a galvanometer, the potential difference of the plate and the positive electrode of the lamp is something less than that of the working lamp electrodes.

(3.) This absolute equality of potential between the negative electrode of the lamp and the insulated plate only exists when the carbon filament is in a state of vivid incandescence, and when the insulated plate is not more than an inch or so from the base of the negative leg. When the lamp is at intermediate stages of incandescence, or the plate is considerably removed from the base of the negative leg, then the plate is not brought down quite to the same potential as the negative electrode.

(4.) A galvanometer connected between the insulated plate and the *positive* electrode of the lamp shows a current increasing from zero to four or five milliamperes, as the carbon is raised to its state of commercial incandescence. There is not any current greater than 0.0001 of a milliampere* between the plate and the *negative* electrode when the lamp has a good vacuum.

(5.) If the lamp has a bad vacuum this inequality is destroyed, and a sensitive galvanometer shows a current flowing through it when connected between the middle plate and either the positive or negative electrode.

(6.) When the lamp is actuated by an *alternating* current, a *continuous* current is found flowing through a galvanometer

* A milliampere is the $\frac{1}{1000}$ th of an ampere.

connected between the insulated plate and *either* terminal of the lamp. The direction of the current through the galvanometer is such as to show that negative electricity is flowing from the plate through the galvanometer to the lamp terminal. This is also the case in (4); but, if the lamp has a bad vacuum, then negative electricity flows *from* the plate through the galvanometer *to* the positive terminal of the lamp, and negative electricity flows *to* the plate through the galvanometer *from* the negative terminal of the lamp.

(7.) The same effects exist, on a reduced scale, when the incandescent conductor is a platinum wire instead of a carbon filament. The platinum wire has to be brought up very near to its point of fusion in order to detect the effect, but it is found that a current flows between the positive electrode of a platinum wire lamp and a platinum plate placed in the vacuum near to the negative end of that wire.

(8.) The material of which the plate is made is without influence on the result. Platinum, aluminium, and carbon have been indifferently employed.

(9.) The active agent in producing this effect is the *negative* leg of the carbon. If the negative leg of the carbon is covered up by enclosing it in a glass tube, this procedure entirely, or nearly entirely, prevents the production of a current in a galvanometer connected between the middle plate and the positive terminal of the lamp.

(10.) It is a matter of indifference whether a glass or metal tube is employed to cover up the negative leg of the carbon; in any case this shielding destroys the effect.

(11.) If, instead of shielding the negative leg of the carbon, a mica screen is interposed between the negative leg and the side of the middle plate which faces it, then the current produced in a galvanometer connected between the positive terminal of the lamp and the middle plate is much reduced. Under the same circumstances hardly any effect is produced

when the mica screen is interposed on that side of the metal plate which faces the positive leg of the carbon.

(12.) The position of the metal plate has a great influence on the magnitude of the current traversing a galvanometer connected between the metal plate and the positive terminal of the lamp. The current is greatest when the insulated metal plate is as near as possible to the base of the negative leg of the carbon, and greatest of all when it is formed into a cylinder which embraces, without touching, the base of the negative leg. The current becomes very small when the insulated metal plate is removed to 4 or 5 inches from the negative leg, and becomes practically zero when the metal plate is at the end of a tube forming part of the bulb, which tube has a bend at right angles in it. Copious experiments have been made with metal plates in all kinds of positions.

(13.) The galvanometer current is greatly influenced by the surface of the metal plate: being greatly reduced when the surface of the plate is made small, or when the plate is set edgewise to the negative leg, so as to present a very small apparent surface when seen from the negative leg. In a lamp having the usual commercial vacuum, the effect is extremely small when the insulated metal plate is placed at a distance of 18 in. from the negative leg, but even then it is just sensible to a very sensitive galvanometer.

(14.) If a charged condenser has one plate connected to the insulated metal plate, and the other plate connected to any point of the circuit of the incandescent filament, this condenser is instantly discharged if the positively-charged side of the condenser is connected to the insulated plate and the negative side to the hot filament. If, however, the negative leg of the carbon horseshoe is shielded by a glass tube, this discharging power is much reduced, or altogether removed.

(15.) If the middle plate consists of a separate carbon loop, which can itself be made incandescent by a separate insulated

battery, then, when this middle carbon is rendered incandescent, and is employed as the metal plate in the above experiment, the condenser is discharged when the negatively-charged side of it is connected to the hot middle carbon, the positively-charged side of it being in connection with the principal carbon horseshoe.

(16.) If this last form of lamp is employed as in (4), the subsidiary carbon loop being used as a middle plate, and a galvanometer being connected between it and either the positive or negative main terminal of the lamp, then, when the subsidiary carbon loop is cold, we get a current through the galvanometer only when it is in connection with the positive main terminal of the lamp; but, when the subsidiary carbon is made incandescent by a separate insulated battery, we get a current through the galvanometer when it is connected either to the positive or to the negative terminal of the lamp. In the first case the current through the galvanometer is a negative current flowing from the middle carbon to the positive main terminal, and in the second case it is a negative current from the negative main terminal to the middle subsidiary hot carbon.

(17.) If a lamp having a metal middle plate held between the legs of the carbon loop has a galvanometer connected between the negative main terminal of the lamp and this middle plate, we find that, when the carbon is incandescent, there is no sensible current flowing through the galvanometer. The vacuous space between the middle plate and the hot negative leg of the carbon possesses, however, a curious unilateral conductivity. If a single galvanic cell is inserted in series with the galvanometer, we find that this cell can send a current deflecting the galvanometer when its negative pole is in connection with the negative main terminal of the lamp, but if its positive pole is in connection with the negative terminal of the lamp, then no current

flows. The cell is thus able to force a current through the vacuous space when the direction of the cell is such as to cause negative electricity to flow across the vacuous space from the hot carbon to the cooler metal plate, but not in the reverse direction.

(18.) If a vacuum tube is constructed, having horseshoe carbon filaments sealed into it at each end, and which can each be made separately incandescent by an insulated battery, we find that such a vacuum tube, though requiring an electromotive force of many thousands of volts to force a current through it when the carbon loops are used as electrodes and are *cold*, will yet pass the current from a single galvanic cell when the carbon loop which forms the negative electrode is rendered incandescent. It is thus found that a high vacuum terminated electrically by unequally-heated carbon electrodes possesses a unilateral conductivity, and that electric discharge takes place freely through it under an electromotive force of a few volts when the *negative* electrode is made highly incandescent.





LECTURE III.

THE Forms of Electric Discharge : Brush, Glow, Spark, Arc.—Vacuum Tubes.—Sparking Distance.—The Electric Arc.—The Optical Projection of the Arc.—The Arc a Flexible Conductor.—The High Temperature of the Arc.—Non-Arcing Metals Lightning Protectors.—The Distribution of Light from the Arc.—Continuous and Alternating Current Arcs.—Voltage Required to Produce an Arc.—The Physical Actions in the Arc.—The Changes in the Carbons.—The Distribution of a Potential in the Arc.—The Unilateral Conductivity of the Arc.—The Temperature of the Crater.—Comparison with Solar Temperature.—The “Watts per Candle” of the Sun.—Intrinsic Brightness and Dissipative Power of Heated Surfaces.—Comparison of Glow Lamp, Arc Lamp, and Sun, in respect of Brightness and Radiation.—Arc Lamp Mechanism.—Arc Lamp Carbons.—The Hissing of Arc Lamps.—The Application of Arc Lamps.—Inverted Arcs.—Series and Parallel Arc Lighting.



WE must now forsake the study of the incandescent electric lamp, and turn our attention to the older form of electric illumination, namely, the electric arc lamp. There are three principal ways in which electric discharge

seems to be capable of taking place across a space filled with air or gas. These are: firstly, by the *glow discharge*; secondly, by an electric spark, or *disruptive discharge*; and thirdly, by the *electric arc*. When two conductors which are at different electric pressures are brought within a certain distance of one another, and when the electric pressure difference

between these conductors is gradually raised, one of the above three modes of electric discharge is always established, and takes place in a manner depending upon the gaseous pressure of the air or gas in which the conductors are immersed, upon the electric pressure difference between the conductors (called the electrodes), and upon the strength of the current which the source of the electric pressure can produce. If the two conductors are, say, two brass balls which are connected with the terminals of an ordinary electrical machine, and capable of making a considerable difference of electric pressure between them, but only capable of generating a very feeble electric current, the following effects are observed when the balls are brought near one another : In the air, at ordinary pressures, the conductors, if viewed in the dark, will be found to be covered with a violet glow of light, and this is more visible if their ends are small—if, for instance, they are the rounded ends of two wires attached to the electrical machine. If one of the conductors has a large surface—if, for instance, it is a brass plate—and the other conductor is not a very good conductor—as, for example, a wooden ball or knob—then in the dark we find a form of glow discharge taking place between the ball and the plate, which is called a *brush discharge*. It has a violet, broom-shaped glow of light, and Sir Charles Wheatstone, on examining it by reflection in a rapidly revolving mirror, showed that it consisted of a series of very rapid electric discharges between the air particles. It is generally accompanied by a slight hissing sound.

This form of brush discharge occurs in nature under some conditions, and is known by the name of St. Elmo's Fire. In the high Alps, when in the neighbourhood of a thunder-storm, travellers have often observed these whizzing brushes of purple light proceeding from their axes and alpenstocks, and from sharp-pointed rocks ; and a similar phenomenon

has been seen proceeding from the masts and yardarms of ships under certain electrical conditions of the atmosphere. When the knob or ball is positively electrified the electric brush is larger and more brilliant than when the ball is negative. The finest brushes are formed in nitrogen gas. On the other hand, if the conductor is very small, and made of metal, then the glow that is seen on the end of this conductor in the dark is not intermittent, but is associated with a current of air proceeding from the conductor. A blunt metal point attached to an electrical machine can thus be made to blow out a candle flame.

The form and nature of the electric discharge in rarefied gases has been studied by many observers. The phenomena are exceedingly complicated, but generally speaking are somewhat as follows :—Into the extremities of a closed glass tube or bulb are sealed platinum wires, which can be connected to an electrical machine or battery or any source of electric current. If the tube is exhausted of its air, or filled with highly rarefied gas of any kind, and an electric current is passed through it, the following facts can be noticed : A glow of light surrounds each end of the platinum wire, and, if the air pressure is gradually diminished until the air is rarefied to about $\frac{1}{700}$ th part of its ordinary pressure, the end of the negative electrode or conductor is seen to be surrounded by a bluish violet light separated from the electrode by a dark space which increases in width as the rarefaction of the air is increased. The positive conductor, or electrode, is also surrounded with a glow of light, frequently of a different colour, and the space in between may be filled up with a glow of light which may or may not be cut up by a series of dark spaces, and is then said to be stratified. This phenomena of stratified discharge can be shown to you very beautifully by connecting the terminals of an induction coil to two platinum wires, which



FIG. 55.—Vacuum Tube, showing Stratification.

are sealed into the ends of a long glass tube filled with rarefied carbonic acid (*see* Fig. 55). This tube is one of a large collection which belonged to the late Mr. Warren de la Rue, and was presented to this Institution. You will see that the space in between the platinum wires is filled with a light-coloured violet glow, which is stratified or cut up into a number of saucer-shaped layers of light. During this form of discharge the negative electrode, or wire by which the current leaves the tube, is found to become hot, and becomes, in time, raised to a red heat; whilst, at the same time, it is more or less disintegrated, depositing a film of metal upon the glass in its neighbourhood.

This glow discharge in gases is a very complex and yet attractive subject, and has engaged the attention of numerous able physicists, such as Faraday, Plücker, De la Rive, Hittorf, Spottiswoode, Moulton, Crookes, and, more lately, J. J. Thomson and many others. It is found that this phenomenon of discharge, both in rarefied gases and in gases at ordinary pressure, can be produced not only by the current of electricity supplied by an ordinary statical electrical machine

or by an induction coil, but also by the current given by a battery or series of galvanic cells. When the electric discharge takes place in air at ordinary pressures, and when the supply of current is not very rapid, and especially if one of the conductors is a poor conductor of considerable surface, the discharge takes the form of a series of electric sparks. The form of this spark and the distances over which it can be produced are very much determined by the nature of the surfaces. Rounded metallic knobs give bright snapping sparks, whereas metallic points give thin whizzy sparks. Besides the form of surface, the interposed gases affect the result. Faraday found that different gases have different restraining powers upon the spark. At the same distance of conductors, hydrogen gas permits a discharge through it more easily—that is, at a lower electric pressure—than air, whereas some other gases, such as hydrochloric acid gas, have peculiar restraining powers. An electric spark is in reality a brief current of electricity taking place between the two conductors, during which time the air or other gas between them is rendered incandescent and therefore luminous, and portions of the material of the conductors between which the discharge is taking place are carried across from one side to the other. For each particular pressure there is a certain *sparking distance*, which, however, is partly determined by the form of the conductors between which the sparks jump. By the term “sparking distance,” corresponding to any electric pressure, is meant the distance over which the spark will spring when the difference of electric pressure between the conductors is gradually raised to that value. If we bring two brass balls, connected with the terminals of an electrical machine, within a certain distance of one another, and then darken the room, you will see these two balls surrounded with a glow of light. On bringing them within a certain lesser distance of one another a snapping spark passes, which is repeated at intervals as the machine is worked.

The first accurate experiments on the sparking distance corresponding to different electric pressures were made by Lord Kelvin. He showed that between slightly curved metal plates it required a pressure of about 5,000 volts to jump across the eighth part of a centimetre, or the twentieth part of an inch. Mr. De la Rue found that between flat plates the difference of pressure required to produce a spark was 8,000 volts when the surfaces were a fifth part of a centimetre, or $\frac{8}{100}$ ths of an inch, apart; and 11,300 volts when the surfaces were one-third of a centimetre, or $\frac{2}{15}$ ths of an inch, apart. Similar experiments have been made by numerous other physicists with like results.

The following table of results for sparking distances under certain conditions has been calculated from the observations of MM. Bichat and Blondlot. If metal balls $\frac{1}{10}$ th of an inch in diameter (one centimetre) are separated by a distance beginning at $\frac{1}{25}$ th of an inch, and increasing by equal amounts—viz., by one millimetre at a time—the electric pressure difference in volts which will just make a spark jump across at the various distances will be as follows (the distances are given in millimetres; one millimetre is nearly $\frac{1}{25}$ th of an inch):—

Distances of the balls in millimetres.	Sparking pressure in volts.	Distances of the balls in millimetres.	Sparking pressure in volts.
1	4,765	12	27,024
2	8,140	13	27,765
3	11,307	14	28,359
4	14,119	15	28,949
5	16,664	16	29,363
6	19,210	17	29,837
7	21,823	18	30,133
8	22,792	19	30,547
9	24,153	20	30,932
10	25,071	21	31,198
11	26,255	22	31,494

Hence we see that, between such metallic balls, an electric pressure of 30,000 or 40,000 volts is required to make a spark jump over a distance of about one inch. It has, however, been shown that these sparking distances are greatly affected by the state of polish of the surfaces and by their form, and that sparking is promoted by light of particular kinds falling on the balls. The light from other electric sparks, as also that from burning magnesium wire, has a strong effect in breaking down the insulation of the air and promoting the passage of a spark between polished metal balls across a distance which it would otherwise not pass. Although an electric spark of a few inches in length represents an electric pressure of many thousands of volts, it represents very little current or quantity. It is somewhat surprising to see a torrent of noisy sparks passing between the discharger of an electrical machine, and yet to know that this represents in quantity, probably, not a thousandth of an ampere of electric current, and that it could hardly effect the chemical decomposition of one drop of water or solution of a metallic salt, although its pressure or potential may be hundreds or thousands of volts. An electric pressure of 100 volts, such as is used for incandescent lamps, will hardly produce a spark over any visible distance; certainly the distance would be less than $\frac{1}{100}$ th part of an inch.

These various forms of discharge—the spark, the brush, and the glow discharge—have all been the subject of elaborate investigations. Faraday particularly studied them in the twelfth and thirteenth series of his “Experimental Researches in Electricity.” Whatever be the means by which the difference of electric pressure is produced, it is found that, if there exists a sufficient resistance in the circuit limiting the rate at which the supply of electricity can take place, then no other form than the spark, brush, or glow discharge is possible. If, however, we employ a primary or secondary battery—that is, a collection of

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galvanic cells—for producing the difference of pressure between the two conductors, such an arrangement differs from an ordinary electrical machine only in the fact that it can supply a stronger or more powerful electric current. The galvanic battery, when compared with an electrical machine such as the Wimshurst machine, must be thought of as a very large pump compared with a very small one. Both these pumps can create a difference of level, pumping up water from one level to a cistern or reservoir at a higher level, but the pump of larger capacity differs from the other in that it can supply faster and can keep up the difference of level in spite of out-flow of water. Accordingly, if a large number of galvanic cells are joined together, and the ends of this battery are connected to two brass balls, it is found that, if the battery is one which has a high internal resistance, it will produce a series of small sparks, which jump over continually from one ball to the other, provided the difference of pressure is great enough and the balls are brought near enough together. When the battery is one of low internal resistance, then the spark discharge cannot be maintained; but if the conductors are brought near enough together, and the pressure sufficiently raised, the spark or glow discharge passes spontaneously and immediately into a third form of discharge, which is called the *electric arc*.

It is possible that this arc discharge was first observed by Curtet in 1802, one year after the invention of the galvanic battery by Volta; but the first careful study of the phenomena of the electric arc in air and in vacuum was made by Sir Humphry Davy at the beginning of this century. At that time Sir Humphry Davy was engaged in the remarkable series of electro-chemical discoveries which resulted in the production of metallic potassium and sodium from caustic potash and soda for the first time.

Finding the necessity for a larger battery than he possessed, he laid a request before the managers of the Royal Institution in 1808, asking them to provide the means of constructing a battery of 2,000 cells with which to continue these researches.* Shortly afterwards this battery was provided, and with it Davy not only continued his electro-chemical discoveries, but studied the phenomena of the electric arc. Connecting the ends of this large battery of 2,000 pairs of plates to two pieces of hard charcoal, which had been heated and then plunged into quicksilver to make them better conductors, Davy observed for the first time on a large scale the phenomena of the electric arc. In his "Elements of Chemical Philosophy" (Vol. IV.) he thus describes the first production of this very large 2,000-volt electric arc :—

"The most powerful combination (battery) that exists in which number of alternations (plates) is combined with extent of surface is that constructed by the subscriptions of a few zealous cultivators and patrons of science in the laboratory of the Royal Institution. It consists of two hundred instruments connected together in regular order, each composed of ten double plates arranged in cells of porcelain and containing in each plate 92 sq. in. ; so that the whole number of double plates is 2,000 and the whole surface 128,000 sq. in. This battery, when the cells are filled with 60 parts of water mixed with one part of nitric

* Sir Humphry Davy laid a request before the managers of the Royal Institution on July 11, 1808, that they would set on foot a subscription for the purchase of a large galvanic battery. The result of this suggestion was that a galvanic battery of 2,000 pairs of copper and zinc plates was set up in the Royal Institution, and one of the earliest experiments performed with it was the production of the electric arc between carbon poles, on a large scale. It is probable, however, that Davy had produced the light on a small scale some six years before, and, according to Quetelet Curtet, observed the arc between carbon points in 1802. See Dr. Paris' "Life of Sir H. Davy."

acid and one of sulphuric acid, afforded a series of brilliant and impressive effects. When pieces of charcoal about an inch long and one-sixth of an inch in diameter were brought near each other (within the thirtieth or fortieth part of an inch) a bright spark was produced and more than half the volume of the charcoal became ignited to whiteness, and by withdrawing the points from each other a constant discharge took place through the heated air in a space equal at least to 4 in., producing a most brilliant ascending arch of light, broad and conical in form in the middle. Fragments of diamond and points of charcoal and plumbago rapidly disappeared and seemed to evaporate in it, but there was no evidence of their having previously undergone fusion."

We may repeat his experiments on a smaller scale. Taking two pencils of hard carbon formed of gas retort coke, these are connected to the two ends of a battery furnishing a pressure of about 50 volts. If these two carbons are brought within a short distance of one another, say $\frac{1}{100}$ th of an inch, about the thickness of a thin visiting card, it will be found that this electric pressure of 50 volts is not sufficient to create a spark capable of jumping over this distance. The very thin layer of air existing between the two carbon terminals is sufficient to insulate entirely this pressure. If the two carbons are touched together at the point of contact they immediately become red hot, and then, on separating them again a slight distance, the electric discharge continues to take place between them, and this is called the electric arc. Instead of bringing the carbons in contact we can break down the insulation of the air by passing a small electric spark from one carbon to the other, or even at some distance away. This can be done by taking a discharge from a small electrical machine, and when the experiment is performed before you you will notice that the electric arc discharge

follows immediately upon the passage of a spark which is too small to be visible.

A very convenient arrangement for producing and examining an electric arc is one in which one of the carbon rods is fixed and the other is moved by a rack and pinion, so as to be brought in contact with the first, and then moved away as required when the arc is produced between them. This arrangement, one form of which, as devised by Mr. Davenport,* is shown in Fig. 56, is needed in an experimental study of the arc.



FIG. 56.—The Davenport Arc Lamp.

In order to view more comfortably the whole phenomena that are taking place in the interior of this dazzling source of light, we will project an image of the electric arc by means of a lens upon the screen, and examine the whole of the effects taking place in it. The arc discharge is most easily obtained and maintained between conductors which are good but not very good conductors, and which are only

* This convenient form of hand-regulated arc lamp was described in *The Electrician*, Vol. XXXII., p. 393.

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volatilised at a high temperature. As we shall see presently, it cannot be maintained at all between certain metals when close together. No substance has been found superior for the purpose of producing an electric arc to a dense variety of carbon, produced either from the graphitic deposit of carbon found in the interior of gas coke ovens, or by the production of a hard variety of carbon from gas coke, lamp-black, or sugar. Twenty-five years ago the carbon rods which were used for this purpose were generally produced by sawing lumps of very hard retort carbon into small rods a quarter of an inch square, but of late years an immense manufacture has sprung up in the manufacture of arc-light carbon rods. In the manufacture of these rods some form of dense and very pure carbon is made into a paste with syrup or coal tar, and from this rods are pressed out which, after being dried and baked at a high temperature, furnish carbon rods of various sizes which are used for the production of the electric arc. These carbon rods are generally sold in lengths, varying in size from a quarter of an inch to an inch in diameter, and from nine to eighteen inches in length. Taking two of these rods, and producing between them an electric arc in a closed lantern, we have now projected upon the screen an enlarged image of the electric arc.

Four things can be noticed in this electric arc as soon as it has been formed by bringing the carbons into contact (*see* Fig. 57). Looking at the optical image of the arc projected upon the screen, we notice that both the ends of the two carbons are brilliantly incandescent, but that, if the arc is formed by a current of electricity flowing always in one direction, called a continuous current arc, then the positive carbon, or the one attached to the positive pole of the battery or dynamo, and marked + in the illustration, is more brilliantly incandescent than the other. This positive pole soon has its end hollowed out into a cup-shaped depression,

which is called a *crater*, and this positive or crater carbon, after being in use for a few minutes, will be found to have its extremity converted into graphitic carbon or plumbago, which will mark upon paper like a pencil, in a way that it would not do before being so used. At the



FIG. 57.—The Electric Arc.

same time the negative carbon appears to get more pointed, and becomes surrounded a little below its tip with small globules, which are, in all probability, condensed carbon vapour. If the arc is maintained for some time, it will be

found that both these carbons are being worn away ; but the positive or crater carbon wears away about twice as fast as the other when the arc is formed in air. In the space between the two carbons we notice a band of brilliant violet light, which is surrounded by a less luminous aureole of a golden colour. In a carefully-regulated, steady electric arc it will be seen that the violet-coloured portion seems to spring from the white-hot crater, and that a well-marked dark space separates this purple part from the golden-coloured, wing-shaped flames which appear to spring from the negative carbon. This intermediate central portion is called the true arc, and hence we have four separate portions of the arc to consider : namely, the crater, the violet-coloured core of the arc, the aureole, and the negative carbon. On looking closely at an arc so projected upon the screen, it will be seen that little bits of carbon are continually becoming detached from one pole, and they immediately fly over to the opposite carbon. It can be shown by several experiments that there is probably a double transport of material going on in the arc. The material of the positive pole is being carried across to the negative side, and the material of the negative pole is being carried across to the positive side.

At this stage it will be interesting to analyse the light of the arc by means of the prism. Placing a vertical slit before the lens which is projecting the image of the arc upon the screen, we cut off all but a linear band of light, the upper part of which proceeds from the crater carbon, the middle part from the violet core of the arc, and the bottom part from the negative carbon. The prism being then placed in front of the lens, we expand this linear image of the slit into a spectrum which is divided longitudinally into three parts. The upper portion, you will observe, is a brilliant continuous spectrum of light, in which all the prismatic rays are present. This spectrum proceeds from the light emitted from the crater of the arc.

The middle portion of the spectrum is much less bright in the orange, red, yellow and green, but, as you will see, is distinguished by two remarkably brilliant violet bands, which are characteristic bands in the spectrum of carbon vapour. The bottom of the spectrum, formed from the negative carbon, is a continuous spectrum similar to the upper one, only less brilliant. On lengthening out the arc by drawing the carbons apart, we find that the two upper spectra move away from one another, the middle spectrum increasing in width. At a certain point the arc breaks down, the middle spectrum with its two violet bands disappearing instantly; but for a short period of time the upper and under spectra linger as the carbons cool, because the carbons continue to glow faintly; and as these carbons cool you observe the two spectra shrinking rapidly from the violet end, until at last only faint traces of the red and green remain, and these finally vanish. But you will see that the lower spectrum, which proceeds from the light of the negative carbon, vanishes first. This only indicates that the crater carbon has a much higher temperature than the negative carbon, and is in accordance with all that we know about the arc.

If we examine an electric arc of this kind, we notice that, as the carbons are gradually drawn apart and the arc lengthened, the aureole of golden vapour increases and forms a sort of large lambent flame, which disappears when the arc is lengthened beyond a certain point. After such a continuous current arc has been extinguished, we can always tell which has been the positive carbon, not only from the crater hollowed in it, but also from the fact that it remains red hot for a longer time than the negative carbon. The exceedingly high temperature of the positive carbon is shown by the fact, above mentioned, that after being used for some time it is converted into graphite at the extremity and will mark paper.

The electrical power taken up in the continuous current electric arc is measured in an exactly similar way to the power taken up and dissipated in the filament of an incandescent lamp—namely, by measuring the current in amperes flowing through the arc, and the difference of electric pressure in volts between the carbons. Multiplying these numbers together, we obtain the value of the power, measured in watts, being absorbed in the arc. By far the best way to denominate electric arcs is by the power in watts absorbed. We may thus speak of a 300-watt arc, or a 500-watt arc, or of a 1,000-watt arc. The length of the arc can be measured most conveniently by projecting the image of the carbons upon the screen. If a sheet of polished metal is then held behind the arc, a faint diffused light will be thrown upon the screen, which will enable us not only to see the arc, but to see also the image of the two carbon rods. By means of a pair of compasses or a foot rule we can measure the length of the magnified image of the arc—that is to say, the distance between the image of the tips of the carbon rods. We can also measure the width of the magnified image of either of the carbon rods. Let us suppose that the rods are each of them half an inch in diameter, and that their image upon the screen is eight inches in diameter; then the arc has been magnified 16 times. If, then, we find that the image of the arc is, say, two inches in length when measured upon the screen, it shows us that the real length of the arc is $\frac{1}{8}$ th part of this, namely, one-eighth of an inch, because the length of the arc is magnified in the same ratio as the width of the carbons. This optical method of measuring the length of the arc affords a very easy means of accurately ascertaining the length of the arc without danger or difficulty. Electric arcs may, roughly, be distinguished as long arcs and short arcs, according to the distance between the carbon poles, although the dividing line between the two is not in any way accurately determined.

Before passing on to consider the physical measurements connected with the electric arc, let us notice that the violet core of the arc, which consists of a torrent of incandescent carbon vapour, acts like a perfectly flexible conductor. If I bring a magnet near to such an electric arc, you will notice that the magnet apparently impels the arc sideways, and if it is brought near enough it will actually blow it out. The reason for this is because the arc, as a conductor traversed by a current, exerts an electro-magnetic force upon the magnet, and in like manner the magnet exerts an electro-magnetic force upon the arc. The arc always tends to move across the lines of force of the magnet, and in this way may actually be bent into a bow shape, or be drawn out into a kind of blow-pipe flame.

Brief references must now be made to the generation of light and heat in the arc. Experiment seems to show that in the continuous current arc—that is, the arc produced by a current of electricity always flowing continuously in the same direction—by far the larger proportion, perhaps 85 per cent., of the light thrown out comes from the highly incandescent crater carbon, about 15 per cent. comes from the negative carbon, and very little indeed, certainly not more than 5 per cent., from the true electric arc. The great source of light, therefore, in the electric arc is the intensely luminous crater formed at the end of the positive carbon, which is brought up to an exceedingly high temperature. Various estimates have been made of the temperature of this electric crater, but the physical difficulties connected with these experiments are very great. Rossetti in 1879 stated, as the result of some of his experiments, that the positive carbon had a temperature of $8,200^{\circ}\text{C}.$, and the negative carbon $2,500^{\circ}\text{C}.$ More recently, Violle has given a temperature of $8,500^{\circ}\text{C}.$ as the temperature of the crater carbon, and other observers have given even higher results than these.

This temperature is sufficient not only to melt but actually to boil away even the most refractory metals. We can easily show the high temperature of the crater by boiling in it a piece of copper or silver. If an electric arc is formed with the crater downwards, and a small piece of copper is placed in the crater, and then the image of the arc is projected upon the screen, you will see the little piece of copper boiling violently and sending out a torrent of copper vapour, which, in its incandescent condition, emits a brilliant green light. In the same way, silver, platinum, and even the refractory metal iridium, can be melted and boiled into vapour in the electric arc. It has been already mentioned that there is a transport of material in the arc, material passing over from the positive to the negative pole, and *vice versa*. It is only natural, therefore, to expect that, since the electric arc is a phenomenon of electric discharge taking place in a highly-conducting vapour, the longest arcs will be capable of being maintained between the most volatile metals, and experience shows this to be the case generally. It has been found that the difference of temperature between the two poles is greater in proportion as they are worse conductors and more easy of disaggregation; also that, to produce an arc of a given length requires a greater current in proportion as the infusibility of the electrode is greater.

It is, however, a very striking fact, as discovered by Wurts, that there are four metals—namely, zinc, cadmium, magnesium and mercury—which are called non-arcng metals, and between which it is difficult or impossible to maintain the electric arc, if the surfaces of these metals are very near together. This fact is taken advantage of in the production of a very ingenious lightning protector.

When overhead circuits, consisting of copper wires carried on insulators on posts, are employed for the distribution of

electric currents, such overhead lines are very much exposed to lightning stroke. If they should be struck by lightning, the lightning discharge will generally pass back into the generating station, and do damage to the dynamo machines. In order to prevent this, a lightning protector of the following kind has been designed, especially for use with overhead alternating-current circuits, the object of which is to allow the lightning striking the overhead circuits to pass to earth, but yet at the same time to prevent the current from the dynamo machine following it. On a slab of marble or porcelain are placed an odd number of small zinc cylinders, one inch in diameter and about three inches long. These cylinders are placed close to one another, being separated by a distance only of $\frac{1}{16}$ th of an inch. Each cylinder is insulated. The middle cylinder is connected by a wire with a plate sunk in the earth. The two outside cylinders are connected to the two lines which form the overhead circuit. If the lightning strikes the line on either side, the discharge jumps over from cylinder to cylinder until it reaches the middle cylinder, and then it goes to earth by the earth-plate. In so doing it will attempt to start an electric arc between the cylinders; but, as a permanent arc cannot be maintained between zinc surfaces, this arc is almost instantaneously extinguished, and the dynamo machine, therefore, is unable to produce an electric arc across between the two mains. In this way a safe passage for the lightning to earth is provided.

It will, in the next place, be necessary to direct attention to the distribution of light from an electric arc. On examining an electric arc formed by a continuous current, the positive or crater carbon being uppermost, a very cursory examination shows us that the light sent out from such an arc is not uniform in all directions. Very little light is sent upwards. By far the larger portion of the light is sent downwards at an angle of from 30 deg. to 40 deg. below

the horizon. In the light of explanations given above, it is very easy to see the reason for this. It is really due to the fact that the greater portion of the light is sent out from the bottom surface of the top carbon; hence, when viewed in a horizontal position, but very little of this incandescent crater can be seen. On elevating the arc, so as to look up underneath it, the amount of apparent surface of the crater which is seen increases up to a certain position as we elevate the arc. But if the arc is elevated beyond a certain amount above the eye, then the bottom or negative carbon begins to stand in the way of the light sent out from the crater, and to diminish the apparent

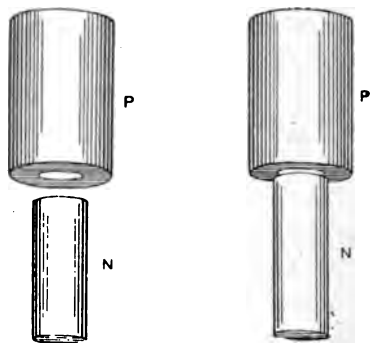


FIG. 58.—Model showing the Relative Areas of Crater seen at different angles of view.

area. By taking two cylinders of pasteboard, *N* and *P*, held by threads a little distance from one another (*see* Fig. 58) and fixing on the underneath end of the upper cylinder a disc of red paper to represent the incandescent crater, we may easily convince ourselves, by looking at this model arc in different directions, that there is a certain position in which the eye can be placed in which it will see the largest apparent area of the crater.

There is, therefore, a position in which the maximum amount of light will be thrown out by a real arc represented

by this model. Accordingly, if the intensity of the light given by an electric arc is measured in different directions, say for every ten degrees of inclination above and below the horizontal, we find different photometric values for these rays in these directions. We can represent this different photometric intensity in different directions by a curve (see Fig. 59), which is called the photometric curve of the arc. The magnitude of the radii of this curve, taken in different directions, represent the intensity of the light pro-

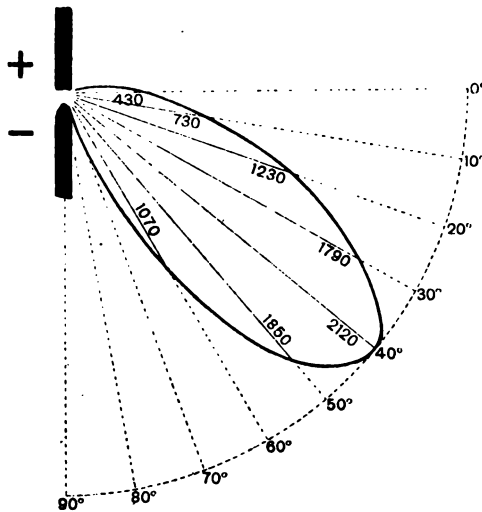


FIG. 59.—Curve Showing the Distribution of Light from a Continuous Current Arc. The figures printed on the radial lines represent the candle-power in different directions.

ceeding from the arc in these directions. The exact position of the maximum intensity will depend upon the length of the arc, the size of the crater, and the relative widths of the negative and positive carbons. Accordingly, in street arc lighting advantage is taken of this fact. If the electric arcs are put up on high poles, or street supports, the positive or crater carbon being upermost, they will throw down their

maximum amount of light in a direction inclined at 40 or 50 degrees to the horizon, and illuminate a circular area around them with a maximum intensity. Just underneath the arc there will be more or less shadow, and in a direction above the horizon very little light will be thrown out.

In order to obtain the best results and the most uniform distribution of light, Mr. Crompton has found it best to use a very small negative carbon, in order that the negative carbon may stand as little as possible in the way of the light sent out from the crater in a downward direction, and that the arc may, therefore, illuminate the largest possible area underneath it. This unequal distribution of the light from the arc can, to a great extent, be modified by the use of a somewhat peculiar glass globe; but we shall point out in an instant a method by which superior results are gained for internal lighting.

Not only can we form an electric arc by means of a continuous current of electricity—that is to say, one always flowing in the same direction—but we can employ an alternating current of electricity. In such an alternating current there is a very rapid change in the direction of the electric current, the current flowing first in one direction for a short period, then being arrested, and then flowing in the other direction for an equally short period. An alternating current of electricity may therefore be compared to the ebb and flow of water in a tidal river, in which the water flows in one direction for a certain number of hours and is then reversed and flows in the opposite direction for about an equal period.

In what are called alternating current systems of electric lighting distribution it is very usual to employ an alternating current having a frequency of from 40 to 100 periods per second—that is to say, the electric current changes its direction from 80 to 200 times in one second. A continuous

current of electricity may be likened, on the other hand, to a non-tidal river in which the water flows always in one direction. We can form an electric arc not only with a continuous but with an alternating current of electricity, and the reason for this is that there is a certain persistence in the electric arc. It is found that, when the arc is made with a continuous current of electricity, the current can be interrupted for a short fraction of a second without putting out the arc. Hence the current of electricity can be reversed in direction without extinguish-

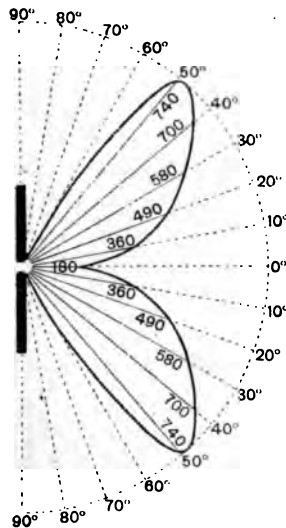


FIG. 60.—Curve Showing the Distribution of Light from an Alternating Current Arc. The figures marked on the radial lines indicate candle-power in these directions.

ing the arc. If, however, it is reversed very rapidly, then the result will be that, instead of having one positive and one negative pole or carbon, each carbon will be alternately positive and negative, and, instead of having only one crater, we shall have both carbons cratered out and sending out light. Hence an alternating current arc has a very different distribution of light from a continuous current arc. In an alternating current

arc the light is not only sent downwards, but it is sent upwards as well, the distribution of light being found to be about equal in directions above and below the horizon. The curve representing the distribution of the light of an alternating arc lamp is shown in Fig. 60, from which it will be seen that the alternating current arc sends its maximum illumination in directions both upwards and downwards equally inclined to the horizon. Objections are sometimes taken to the use of an alternating current arc on the ground that it makes a disagreeable noise or humming. This, however, is much less evident in some arcs than in others, and, by the employment of suitable carbons and a proper frequency of current, can very nearly be abolished. It appears to be proved, however, that an alternating current arc does not yield the same total illumination for a given expenditure of energy in it as does a continuous current arc.

Returning, then, for the moment, to the continuous current arc, we must note some more physical facts connected with the formation of an electric arc between various terminals. When the arc is formed between carbon poles it is very easy to show that the true electric arc cannot be formed for much less than 35 volts; that is to say, it is not possible to produce a separation of the two carbons from one another, producing a true electric arc as opposed to a mere incandescence of the tips of the carbons, unless the pressure difference between the carbons amounts to about the above value. When the electric arc is formed with a greater pressure difference, it is found that there is a certain constant pressure not far from 35 or 39 volts, according to most experimentalists, which does not, as it were, take any part in the production of the current through the arc. If electric arcs of various lengths are formed, and if at the same time we measure the current flowing through the arc and the difference of pressure between the carbons, it has been found by various

observers that the result can always be expressed in the following way: Let the difference of pressure between the carbons, measured in volts, be called the working volts of the lamp; then the length of the arc is found to be proportional to the working volts diminished by a certain constant amount, and to be inversely proportional to the current flowing through the arc. The question then arises, What is this constant pressure which is first to be produced before any true electric arc can be formed?

It has been maintained by some that the electric arc acts as if it contained a source of electric pressure which operates against the external electro-motive force producing an arc. There are various reasons, however, why this position is untenable, and there are also arguments supporting the view that this constant pressure, which must be first applied before any true arc can be formed, represents one factor in the measurement of the physical work that must be done in order to volatilise the carbon in the crater, and to maintain that production of carbon vapour. The view now generally held is that the temperature in the interior of the crater of the electric arc is the temperature of boiling carbon, or of the material of the poles between which the electric arc is formed. This theory receives support, amongst other results, from the experiments of Victor von Lang. In his experiments he formed electric arcs between poles of various metals, and measured this constant pressure difference below which no arc of measurable length can be maintained. His results are seen in the table below, and it will be at once noticed that the order in which the metals stand as regards the magnitude of this so-called back electromotive force is the order of their infusibility or volatility. Carbon, which is the most infusible and non-volatile, heads the list, and the most fusible and volatile metals, such as cadmium and silver, are found at the bottom. If, then,

Table of Electric Pressures or Voltages required to begin an Electric Arc between poles of the materials named. (Victor von Lang.)

Material used for poles between which the Arc is formed.	Initial voltage for production of an Arc between these poles.
Carbon	35 volts.
Platinum	27 "
Iron.....	25 "
Nickel.....	26 "
Copper	23 "
Silver	15 "
Cadmium	10 "

the temperature of the crater is the temperature of the boiling metal or material between which the electric arc is formed, we should expect such a crater to have a constant temperature and to emit light of a perfectly constant composition. Very good reasons have accordingly been shown by different physicists for believing that, in the carbon arc, the temperature of the crater of the positive carbon has a perfectly constant value, which is that of the boiling point of carbon.

From the above table of results it will be seen that the most infusible material (carbon) requires the highest voltage to begin a true electric arc, and generally that the order of the initial voltage is the order of infusibility of the materials. It has been shown that the lower the electric conductivity of the material of which the poles are formed, the greater is the difference of temperature between the positive and negative poles when an arc is produced between them. The following, therefore, are the reasons why carbon, in its hard graphitic form, is so peculiarly suitable for use as the poles or electrodes in forming an electric arc :—

1. It is a rather poor conductor of electricity (compared with metals). Hence we obtain a great difference of tem-

perature between the poles ; in other words, we gain the advantage of having a highly-heated crater on one pole.

2. It is easily disintegrated. Hence we obtain a longer arc for a given electric pressure difference between the poles.

3. It is very infusible, and its oxide is a gas ; and therefore does not form a deposit of oxide in and around the arc.

4. It is an inferior conductor for heat. Hence the high temperature is localized at the ends of the poles.

Just as no material has yet been found which is superior to carbon for the incandescing conductor of glow lamps, so no substance has been yet discovered superior to carbon for use in the production of an electric arc.

Hence we must conclude that the processes at work in the production of an electric arc are somewhat as follows :—In the crater, carbon is being boiled into vapour at a temperature probably about 3,500° and 4,000° centigrade. On looking carefully at a magnified image of the crater projected by a lens on to white paper, we can see an apparent seething and effervescence of the surface of the intensely white hot floor of the crater. This torrent of carbon vapour passing towards the negative pole forms the violet core of the arc, the violet colour being that of the vividly-incandescent carbon vapour, possibly superheated to a higher temperature than the boiling point of carbon by the passage of the electric discharge through it. At the cooler negative pole some of the carbon vapour is condensed ; but a portion of the torrent of carbon molecules carried across is deflected back and returns to the positive pole, causing the golden aureole or flame, and creating thus a double carbon current in the arc. The negative carbon gives evidence after use of having been worn away by a kind of sand-blast action, and is so worn away into a series of terraces. The negative carbon also undergoes a curious series of changes, by which a point or knob of carbon is gradually formed on it.

This knob or mushroom end finally breaks off, and is thrown out of the arc. In Fig. 61 are shown a series of diagrams illustrating the gradual changes the negative carbon undergoes.

It has been shown by the author that, if a third carbon pole is dipped into the continuous current arc, so that its tip is in the core of the arc but not touching either positive or negative carbon, or if the electric arc is projected sideways by a magnet on to the third carbon, then a great difference of electric pressure exists between the positive or crater carbon and this

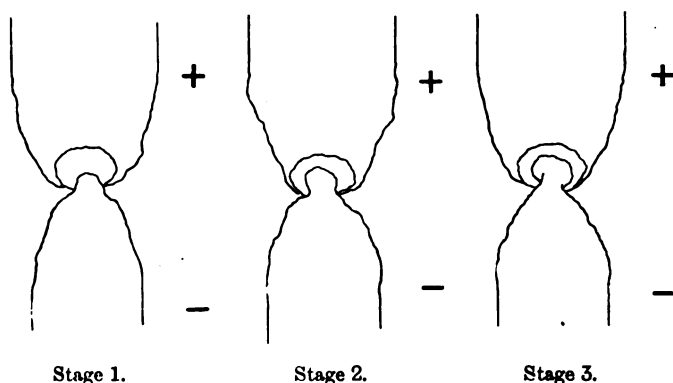


FIG. 61.—Diagrams showing the Changes in Form of the End of the Negative Carbon.

third pole, sufficiently great to illuminate a small incandescent lamp or to ring an electric bell; but that no sensible difference of electric pressure exists between the negative carbon and the third pole.

If a continuous current electric arc is formed in the usual way, and if a third insulated carbon, held at right angles to the other two, is placed so that its tip just dips into the arc (see Fig. 62), we can show a similar series of experiments to those described in Lecture II. under the head of the “Edison effect.” The arc is rather more under control if

we cause it to be projected against the third carbon by means of a magnet. We have now formed on the screen an image of the carbon poles and the arc between them in the usual way. Placing a magnet at the back of the arc, the flame of the arc is deflected laterally and is blown against a third insulated carbon held in it. There are three insulated wires attached respectively to the positive and to the negative carbons of the arc and to the third or insulated carbon, the end of which dips into the flame of the arc projected sideways by the magnet. On starting the arc this third carbon is instantly brought down to the same electrical potential as the negative carbon of the arc, and connecting an amperemeter in between the negative carbon

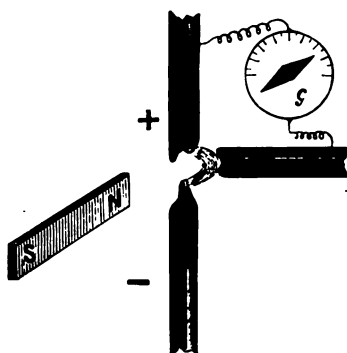


FIG. 62.—Electric Arc projected by magnet against a third carbon, and showing strong electric current flowing through a galvanometer, *G*, connected between the positive and third carbon.

and the third or insulated carbon, we get, as you see, no indication of a current. If, however, we change the connections and insert the circuit of the galvanometer between the positive carbon of the arc and the middle carbon, we find evidence, by the violent impulse given to the galvanometer, that there is a strong current flowing through it. The direction of this current is equivalent to a flow of negative electricity from the middle carbon through the galvanometer to the positive carbon of the arc. We have here, then, the "Edison effect" repeated

portion of its length, and therefore a practical limit is placed to the use of very small negative carbons. It appears probable, from the experiments which have been made, that the hissing of an arc lamp—at any rate, of a continuous current arc lamp—is caused by a too great current density; that is to say, by employing a current too large in comparison with the cross-section of the carbons. Generally speaking, in most of the ordinary arc lamps used for outdoor illuminating purposes, the positive or crater carbon is about 14 or 15 millimetres in diameter—that is to say, $\frac{3}{8}$ ths of an inch—and the negative carbon about 10 or 11 millimetres, or $\frac{7}{16}$ ths of an inch. Under these circumstances the current density in the carbon is somewhere about 50 amperes per square inch. Under some circumstances an electric arc burning quietly sets to work suddenly to hiss loudly, and at that moment the current through the arc also suddenly increases and the difference of pressure between the carbons decreases. M. Blondel thus explains the cause of this effect. At the moment when the hissing begins the voltage falls and the arc loses its transparency, and is transformed into an incandescent mist which hides the crater. When the hissing ceases the arc suddenly becomes violet and transparent again, and the surface of the crater appears to be covered with black specks, which gradually disappear. Whilst the hissing lasts, the brightness of the crater is diminished, and the violet colour of the arc becomes replaced by a very characteristic blue-green tint which seems to point to a lowering of the temperature. M. Blondel thinks that in this case the uniform flow of carbon vapour is replaced by an intermittent disruptive discharge in which the carbon is torn away as it were in pieces, and not evaporated. Hissing occurs when the current density exceeds about an ampere per square millimetre of crater surface, and is more likely to occur with soft carbons than with hard.

In the employment of arc lamps for illuminating purposes, owing to the fact that the light is radiated from an exceed-

some time, we find that its extremity is cratered out and converted into graphite, just as if it had been employed as the positive carbon in forming an electric arc.

Much interesting work yet remains to be done in studying the physics of the electric arc, and we cannot say yet that we fully understand the mechanism which underlies this familiar electric phenomenon.

We must not pass on from this part of the subject without making reference to experiments which have been made to determine the intrinsic brightness of the crater of the electric arc. It has been estimated by Mr. A. P. Trotter that each square millimetre of the incandescent crater sends out light equal in brightness to 170 candles. Other observers, such as M. Blondel, have determined a similar value, viz., 160 candles per square millimetre. That is to say, an arc light crater, having an area, say, of 30 square millimetres, would send out light which would produce an illumination or brightness on a white surface placed parallel to it equal to about 5,000 candles placed at the same distance. The intrinsic brilliancy of the arc crater exceeds that of any other source of light with which we are acquainted, except the sun. Certain materials exist for making a comparison between solar temperature and brilliancy, and that of the arc light crater. It is well known that the sun's edge (limb) is much less bright than the centre (40 per cent. less); but, as a first rough approximation, we may take it as equally bright, and assume that the sun radiates light as if it were an incandescent circular disc of 852,900 miles in diameter. The actual area of its apparent surface is then $1\frac{1}{2}$ billion billion square millimetres ($1.48 \cdot 10^{24}$ sq. mm.). Prof. Young, in America, has made an estimate of the illuminating power of the sun, and his conclusion is that it is equal to 1,575 billion billion candles ($1575 \cdot 10^{24}$ c.p.), correction being made for terrestrial atmospheric absorption.

Hence the mean intrinsic brilliancy of the sun's apparent surface is about 1,000 candles per square millimetre.

The rate at which energy is being sent out from the sun's surface has been estimated from data given by Herschell and Pouillet; and more recently Forbes and Langley have corrected these data, so that from them Lord Kelvin estimates the solar radiation as equal to 133,000 horse-power per square metre. This corresponds very nearly to 100 watts per square millimetre. Accordingly every square millimetre of solar surface is sending out energy as radiation at a rate equal to 100 watts, and producing a candle-power of 1,000 candles. The sun is, therefore, working at an "efficiency," as a glow-lamp maker would say, of $\frac{1}{10}$ th of a watt per candle. If we take the mean value of the estimates that have been made for the intrinsic brilliancy of the crater of the electric arc as 160 candles per square millimetre, and the rate of energy radiation as 30 watts per square millimetre, which is probably not far from the truth,* we see that the solar surface radiation is 100 watts, and the solar intrinsic brilliancy 1,000 candles per square millimetre; whilst the electric arc light crater has a radiation of about 30 watts and a brilliancy of 160 candles per square millimetre. The brilliancy of the sun is, therefore, according to these figures, six times that of the electric arc crater, and its rate of surface radiation about three times. Compare this again with the incandescent carbon of the glow lamp. A 16-c.p. glow lamp with circular filament has a carbon filament five inches long and $\frac{1}{160}$ th of an inch in diameter. The area of

*A series of experiments made in the author's laboratory seemed to show that in various sized arcs, using from 300 to 1,000 watts, the crater area was always of such size that in every case about 27 to 30 watts was dissipated per square millimetre of crater surface. This is on the assumption that the whole work done in the arc is in the first place expended in the crater, and that the rest of the heating of the carbon is due to secondary effects.

its apparent surface is, then, about 30 square millimetres, its total surface is very nearly 100 square millimetres, and it will radiate energy at a rate equal to about 50 watts when giving a light of 16 candle power. Hence, per square millimetre of apparent surface, its intrinsic brilliancy is about half a candle, and its rate of radiation half a watt. Comparing the radiation power and intrinsic brilliancy of the three illuminants—sun, electric arc crater, and glow lamp filament—per square millimetre of surface, we have a rough comparison as follows:—

—	Rate of radiation of energy per square millimetre of surface.	Intrinsic brilliancy or candle-power per square millimetre.
Sun	100 watts.	1,000 candle-power.
Electric Arc Crater	30 „	160 „
Glow Lamp Filament...	$\frac{1}{2}$ „	$\frac{1}{2}$ „

The law connecting radiation with temperature at these high temperatures is not yet definitely known, but the above figures warrant the conclusion that the solar temperature, though much higher, is not enormously higher than that of the crater of the electric arc. The temperature of the sun has sometimes been assumed to be millions of degrees Centigrade. All such guesses must be exceedingly wide of the mark. The solar temperature, at least at the surface, is probably much nearer 6,000° or 7,000°C.

It is necessary to note the proper manner of comparing various illuminating agents in respect of their specific energy radiation and light-producing powers. If we consider a straight cylindrical carbon filament which is, say, 5in. long and $\frac{1}{100}$ th of an inch in diameter, this filament will have a *total* surface of nearly $\frac{1}{100}$ ths of a square inch, but its *apparent* or *projected* surface will only have an



per square unit of area from the incandescent surface the *dissipating power* of that surface, we see that the ratio of dissipating power to brightness is about one watt per candle for a carbon filament when worked under the conditions of temperature at which its "efficiency" would generally be called three watts per candle-power. Hence it happens that in the above table, comparing solar, arc, and glow lamp radiation, the brightness of the carbon filament is given as half a candle per millimetre, and the dissipating power as half a watt per millimetre. It is necessary to distinguish between the ratio of dissipating power to brightness and total radiation of energy to observed candle-power. We shall call the former the watts per normal candle-power and the latter the watts per observed candle-power.

We cannot assume that the law which experiment shows connects together the candle-power and the rate of energy waste in watts in a glow-lamp filament is followed at temperatures much higher than that at which the incandescent filament is usually worked. The temperature of the carbon filament when being used at about 3 watts per candle-power is probably not far from $1,500^{\circ}\text{C}.$ or $1,700^{\circ}\text{C}.$ Up to that point the candle-power appears to increase nearly as the cube of the power in watts taken up. We cannot, however, heat the carbon to a temperature higher than that which corresponds to one watt per observed candle-power without rapidly volatilising it in vacuo. In the electric arc crater the temperature appears to correspond to about one-fifth of a watt per normal candle-power; and this temperature M. Violle has stated to be about $3,500^{\circ}\text{C}.$, or about double that of the incandescent carbon of the glow lamp at normal temperature. These estimates of arc crater temperature, however, probably err on the side of being too low, and the true crater temperature may be even higher than the above value. Stefan, in 1879, suggested that the temperature

certain narrow lines, and not to attempt, in any way, to give even an outline of the whole of the methods which are at the present moment employed for this purpose. Avoiding technicalities, however, the endeavour will be made to enable the reader to grasp certain broad general principles which underlie the construction of all the appliances used for the generation and distribution of electric current for industrial purposes, and then we shall proceed to describe generally the arrangements employed in the electric lighting of a modern city. We must return, in the first place, to facts touched upon in the First Lecture, and examine more in detail the relation of magnetism and electric currents. In so doing it may be an advantage, for the sake of brevity and clearness, to depart a little from the usual phraseology. We must also trace our way in the light of a series of preliminary experiments.

It has already been pointed out that a wire, conveying what we call an electric current, exerts a magnetic influence, and at the same time becomes heated. In fact, the phrase, "an electric current," is merely a comprehensive expression used to denote all the effects of heat and magnetism that are known to exist in and around a wire which we speak of as being traversed by an electric current. If we take a conducting wire—say a copper wire—and pass through it a strong electric current, we find that such a wire, when dipped into iron filings, takes them up, the filings clinging to the wire forming a bunch around it. This fact was discovered about 1820 by the French experimentalist Arago. We can more carefully and deliberately explore the nature of this magnetic action of a wire conveying a current in the following way: Passing the copper wire through a hole in a glass plate, or card, so that the wire stands vertically to the plate, we then sprinkle the glass plate uniformly with steel filings, and, by causing a powerful current to pass through the copper wire, and by gently

prised the use of clockwork, and this involved many objections. Modern arc lamp mechanism almost entirely depends upon the employment of electromagnets, which act as the source of power to move the carbons. Without attempting any extended description of even a portion of the devices employed, a very general idea may be obtained by considering the simplest form of shunt and series coil lamps. One of the carbon rods, say the upper one, is fixed to an iron rod or core *I* (see Fig. 64), which has one end inserted just inside the hollow of a bobbin wound over with thick insulated wire, *Se*, and the other end inserted just inside a bobbin, *Sh*, wound over with fine wire. The

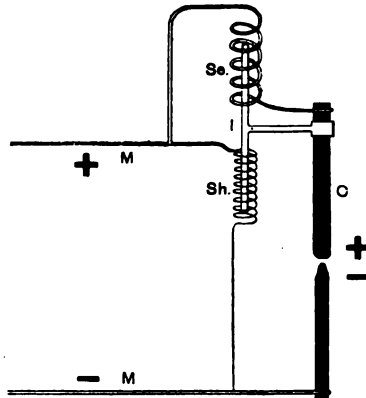


FIG. 64.—Diagram of Shunt and Series Coil Arc Lamp Mechanism.

M, M, are the two Electric Supply Mains ; *Se*, the Series Coil ; *Sh*, the Shunt Coil ; *I*, the Iron Core, and *C*, the Carbons.

former bobbin is inserted in the line, bringing the current to the arc, and is called the *series* coil. The fine wire coil has its wire joined across between the two carbons, and is called the *shunt* coil. The arrangement will be understood by reference to Fig. 64.

When the carbons are placed in contact, and a current is passed through them, this current passes through the series coil on its way to the carbons. It therefore energises that coil, and causes it to attract or pull in the iron core. It will be

understood by reference to the figure that this action immediately separates the carbons a little way. At the same time the shunt coil is traversed by a current, the action of which is to attract its half of the iron core, and therefore to exert an influence drawing the carbons together. The carbons therefore become separated by a small amount, thus striking the arc, and the amount by which they become so separated is determined by the relative magnetic power of the two coils. If the current becomes weakened by the carbons burning away, then the action of the shunt coil preponderates, and draws the carbons together again, thus shortening the arc and increasing the current. When the current increases in strength the series coil exerts the most powerful action of the two, and separates the carbons again by a small amount. It will be seen that the iron core is kept floating between two balanced attractions, one of which tends to draw the carbons attached to it together, the other to separate them. By this simple means, modified in practice by various details of construction, an arc lamp can be constructed which is self-feeding and which sets itself in action by the simple process of switching it on to the circuit.

There are an immense variety of arc lamps depending upon this principle, in all of which various means are taken to make the feed of the lamp properly gradual. If the feed of the lamp is not sufficiently regular it shows itself by rapid changes in the appearance of the arc. As the arc lengthens the violet light of the true arc preponderates, and the lamp burns with a much more bluish colour. If the carbons then come together suddenly the lamp is very liable to hiss and to burn for some time with a more reddish light until the carbons are separated. In lamps with imperfect feeding mechanism these changes give rise to irregularities in the light which are very annoying. In a good arc lamp the feeding mechanism should work with such perfect regularity that the

movement of the carbons can hardly be discerned with a magnifying glass. The best manner of examining the feed of an arc lamp is to project, by means of a lens, an image of the arc lamp upon the screen, and then to examine this optical image. The motion produced by the feeding mechanism is then magnified in the same ratio as the dimensions of the arc, and any irregularity of motion is very easily detected.

One word ought to be said with regard to the quality of carbons used in arc lighting. A vast amount of ingenuity and capital has been expended in investigations intended to discover the best processes for the manufacture of arc light carbons. These carbons are now made as above mentioned by forming into a paste some very pure form of carbon by means of a fluid, such as tar or syrup, capable of being carbonised. This paste is squeezed out into rods by powerful hydraulic presses, and these rods are then dried and baked. Carbons are classified as cored or non-cored carbons. In the cored carbons the centre of the carbon rod is perforated by a longitudinal hole, thus forming a carbon tube, and this carbon tube is filled up with a less dense form of carbon. These cored carbons are generally used for the positive carbon of the arc, and the object of making a softer central portion to the carbon is to determine the position of the crater and compel it to keep a central position.

As already explained, in operating continuous current arc lamps the positive or crater carbon is generally placed uppermost. Under these circumstances there is a cone of shadow directly under the lamp which is due to the negative carbon. Mr. Crompton has found it to be a great advantage to use very much smaller negative carbons than is usually done, and in his arc lamp he uses the smallest negative carbon which can be employed practically. On the other hand, a too small negative carbon is liable to become heated over a greater

the secondary conducting circuit, whilst the first is called the primary circuit. Faraday discovered that, as long as the electric current flowing in the primary circuit remains constant or unaltered, no effect will be produced by it upon the secondary bobbin. The aperture, or central hollow, of this secondary bobbin remains traversed by part or all of the magnetic induction, or, as we have called it, the magnetic current of the first bobbin, but otherwise it is not affected. If the electric current traversing the primary bobbin is either stopped or reversed in direction, or is altered in strength, the immediate result is to produce a current of electricity circulating in the secondary bobbin; which current is, however, only a very brief or transitory current, and does not continue for any length of time. It is, as it were, a sort of wave of electricity passing through the secondary circuit. By appropriate instrumental means we can measure the quantity of electricity which is in this way set in motion. We can also measure, in the unit called an ohm, as explained in the First Lecture, the electrical resistance of this secondary circuit. Faraday showed that the strength of the magnetic induction which is traversing the secondary bobbin could be exactly measured by the product of the resistance of this secondary circuit and the whole of the quantity of electricity set flowing in it when the primary current was suddenly arrested. Hence the use of a secondary circuit of this kind affords us the means of exploring the magnetic condition of the space in the interior of such a primary bobbin.

In Fig. 68 is shown a pair of spiral conducting circuits, **P** and **S**, which consist of spirals of wire wound through suitable holes in a card. If a current of electricity is passed through the circuit **P**, and if iron filings are sprinkled on the card, we find that the lines of magnetic induction, due to the primary circuit, are marked out. Some of these lines will be seen to pass through, or be linked with, the secondary circuit **S**.

It may be well to state at this point that the quantity of electricity which flows past any point on a conducting electric circuit in one second, when it is being traversed by a current of one ampere, is called *one coulomb*. We are able to measure quantity of electricity by means of an instrument called a ballistic galvanometer, and, if we are provided with such an appliance, it is a comparatively simple matter to determine the whole quantity of electricity which flows through a circuit when any change takes place in the magnetic current or induction linked with it.

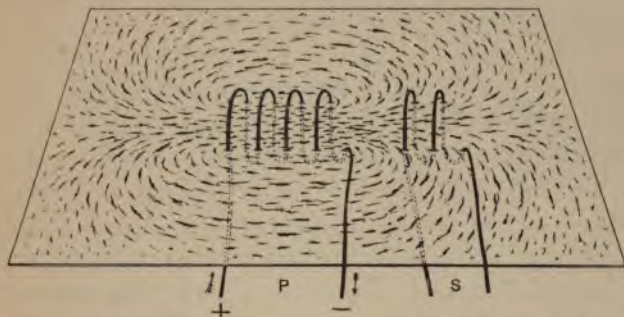


FIG. 68.—Curves delineated by Iron Filings on a Card, showing the line of Magnetism of a Spiral Primary Circuit, P, passing through and linked with a Secondary Circuit, S.

Returning, then, to the case of our primary and secondary bobbin: let us, for the sake of simplicity, assume that the secondary bobbin is wound closely on the outside of the primary bobbin, and that the primary bobbin has a certain given number of turns, and is traversed by a current of a certain strength measured in amperes. When a primary current of, say, ten amperes is passing through the primary bobbin, we have a magnetic current of a certain strength, or a magnetic induction, flowing in the direction of the axis of that primary bobbin, part or all of which magnetic induction or current perforates or passes through the secondary circuit. If the secondary circuit is connected to a ballistic galvanometer,

and the primary current is suddenly stopped, the galvanometer will give an indication that a brief wave of electricity has passed through the secondary circuit, and this is called a secondary current. The product of the quantity of electricity in this secondary current and the resistance of the secondary circuit is, under these circumstances, a measure of the amount of the total magnetic induction, or the total magnetic current due to the primary current in the primary bobbin which is flowing through or linked with the secondary circuit. In the above case the magnetic circuit consists of air—that is to say, the magnetic current flows in a circuit which is merely the air space in and around the primary bobbin.

Supposing that we insert in the interior of the primary bobbin a thick iron rod wholly filling up the interior space, and either straight or bent round so as to form a closed ring. Experiment shows that under these circumstances the state of affairs is greatly altered. On passing through the primary bobbin the same current of, say, ten amperes, and then stopping that current suddenly, we should find that the electric impulse produced in the secondary circuit is immensely increased, and that the quantity of electricity circulated in the secondary circuit would then be a thousand or more times greater—multiplied probably 50 times if the iron bar were a short straight rod, and multiplied 2,000 times if the iron was bent round in the form of a ring. This fact shows us that a given number of ampere-turns in the primary bobbin produces an electric impulse, or electromotive force, in the secondary circuit which is dependent upon the nature of the material filling the space in the interior of, and outside of, the primary bobbin.

These facts can all be summed up in the following statements: If there be two circuits of insulated wire, which are called respectively the primary and secondary circuits, and

which are placed near to one another or wound over one another, then the flow of a continuous current of electricity through the primary circuit produces no effect whatever upon the secondary circuit, so long as that primary current remains unaltered in strength. If the primary current is changed in strength, either being increased or diminished, reversed or reduced to nothing, this change in the number of ampere-turns in the primary bobbin gives rise to an electromotive force, or force setting electricity in motion, acting in the secondary circuit, and this electromotive force, or electric impulse, depends essentially upon the *rate of change* of the current strength, or the rapidity of the change which is made in the strength of the primary current, and, therefore, in the rate of change of strength of the magnetic current or magnetic induction which is produced by the primary current, and which passes through the secondary circuit.

Following a conception of Faraday's, it is usual to speak of the direction of the magnetic current which surrounds the electric current as the direction of the magnetic *lines of force* due to the primary current, and these lines of force can, as we have already explained, have their directions made manifest by the employment of steel or iron filings. The peculiar property which iron possesses, when used as a magnetic circuit, is that it permits the production through it of a given magnetic current by the employment of a far less number of ampere-turns per unit of length than does any non-magnetic material.

We may examine the behaviour of iron in this respect most easily by returning to the fundamental experiment by which it was discovered. Let us take a wooden ring having a circular section, and wind over it insulated copper wire to form a primary circuit, and over that again another insulated secondary circuit. Through the primary circuit let an electric

current be passed, producing, therefore, a certain magnetising force, measured by the number of ampere-turns per unit of length of that bobbin. From what has been above explained it will be seen that this primary current gives rise to a magnetic induction or magnetic current which passes round the ring, and some or all of which traverses the secondary circuit. If the primary current is stopped, the arrest of the magnetic current flowing through the secondary circuit creates in it an electromotive force which causes a brief flow of electricity to take place through that circuit, if it is complete, and that electromotive force at any instant is

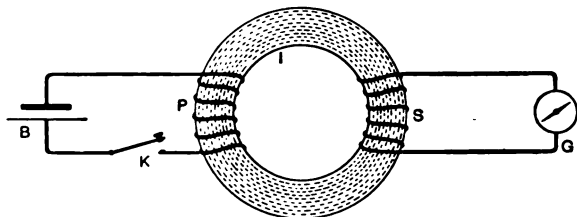


FIG. 69.—Iron Ring I, having wound on it two Circuits,—a Primary, P, connected to a battery, B, through a switch or key, K, and a Secondary Circuit, S, connected to a galvanometer, G. When the Primary Current is stopped or started it creates an Induced Current in S.

measured by the rate at which the magnetic field or magnetic current passing through the secondary circuit is being changed. Supposing, however, that these circuits, instead of being wound on a wooden ring, are wound on an iron ring of exactly the same size (*see* Fig. 69), and that we pass through the primary circuit a current of the same magnitude as before. The arrest or stoppage of this current would now be found to produce a vastly greater electromotive force in the secondary circuit, and the explanation of this fact is that the primary ampere-turns are now able to produce a much greater flow of magnetic current through the secondary circuit, and that, therefore, the arrest of the primary current creates a greater rate of change in the magnetic current flowing through the secondary circuit. The nature of the material filling the space, either partly or

wholly, in and around the primary and secondary circuits, has a very important influence on the inductive effect of the primary circuit upon the secondary.

It is sometimes customary to speak of the primary circuit as exerting a *magneto-motive force* on the magnetic circuit. In the above simple instance of the iron ring, we see that we have two electric circuits and one magnetic circuit, which are linked together like the links of a chain, and that between the three circuits there is a fixed relation of action which is as follows: The electromotive force acting in the primary circuit produces in the primary circuit an electric current. This electric current, flowing a certain number of times round the primary circuit, produces a magneto-motive force of a given magnitude, which creates a magnetic current, or, as it is called, a magnetic induction in the magnetic circuit. The magnetic circuit, in turn, exercises an influence upon the secondary circuit, but only when the magnetic induction or magnetic current is being changed. In the secondary circuit an electromotive force is set up whenever the magnetic current is being altered in magnitude, and the electromotive force acting to produce an electric current in the secondary circuit is measured by the rate at which the magnetic current or magnetic induction throughout is being changed.

We have, therefore, the following operations related to one another in these three circuits, the primary circuit, the secondary circuit, and the magnetic circuit. The primary electromotive force acting in the primary circuit produces a primary current; the primary current acting on the magnetic circuit produces a magneto-motive force; the magneto-motive force produces a magnetic current or magnetic induction in the magnetic circuit; the magnetic induction, *when changing in amount*, produces a secondary

electromotive force in the secondary circuit, and therefore a secondary current. In order to distinguish that quality of iron and air in virtue of which a given magneto-motive force produces a much greater magnetic current or magnetic induction in the iron than it does in the air, we say that the iron possesses a less magnetic resistance than the air. The magnetic resistance of iron may be as much as from one to two thousand times less than that of air.

Supposing, in the next place, we make a cut across each side of our iron ring (as seen in Fig. 70), separating the iron ring into two portions by a narrow air-gap across each

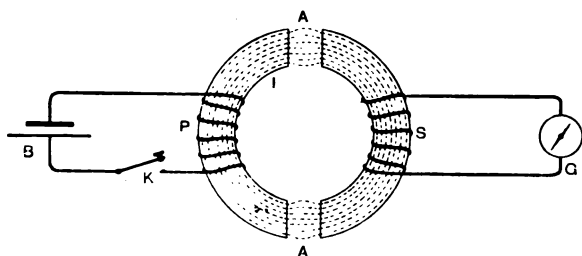


FIG. 70.—Iron ring I, as in Fig. 69, but having a pair of air-gaps made by cutting the ring in two at the points **A A**.

side. It would then be found that a given electric current flowing in the primary circuit would, when arrested, produce a much smaller secondary electromotive force in the secondary circuit than it does when the iron ring is uncut. The reason for this is because the air-gap on both sides has increased the magnetic resistance of the magnetic circuit; and, accordingly, although the magnetic current or induction is able to traverse the air-gaps, the magneto-motive force of the primary circuit is now not able to produce so much magnetic current or magnetic induction in the magnetic circuit; and hence, when the primary current is arrested, the change in the magnetic induction or magnetic current flowing through the secondary circuit is not so great as before.

The two cases we have just examined—namely, the iron ring complete, with a primary and a secondary circuit wound over it, and an iron ring with two air-gaps in it—that is to say, two half-iron rings, on one of which is wound a primary circuit, and on the other a secondary circuit—constitute two typical models which will enable us to explain the action of appliances which are called dynamo-electric machines and transformers.

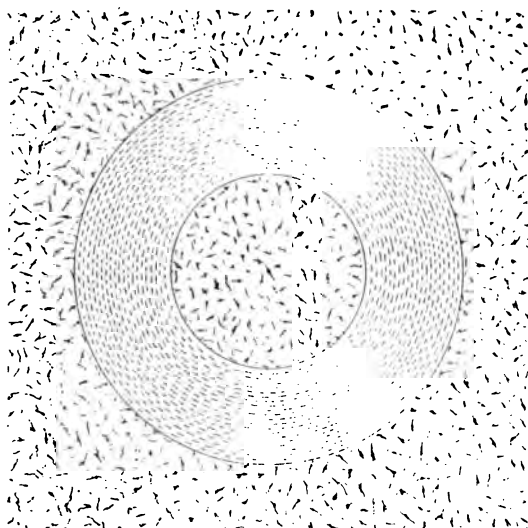


FIG. 71.—Iron Ring magnetised in lines round the ring, but producing no external magnetic field or influence.

Attention should, at this point, be directed to four diagrams, which are reproductions from photographs of experiments easily made.

In Fig. 71 is shown an iron ring, circular in form. Let a spiral of insulated wire be wound on this ring and an electric current be passed through it. This current will magnetise the ring along lines shown as dotted lines. If iron filings are sprinkled over a card held over the ring they will not

arrange themselves in any particular form in the space outside the ring, thus showing that the magnetism is wholly confined to the ring. Next, let a narrow cut be made in the ring (*see* Fig. 72). At once we find, by applying the iron filings test, that there is now a development of magnetism in the space outside the ring, and that there are lines of magnetic induction or current passing across from one side of the air-gap to the other, producing a powerful magnetic

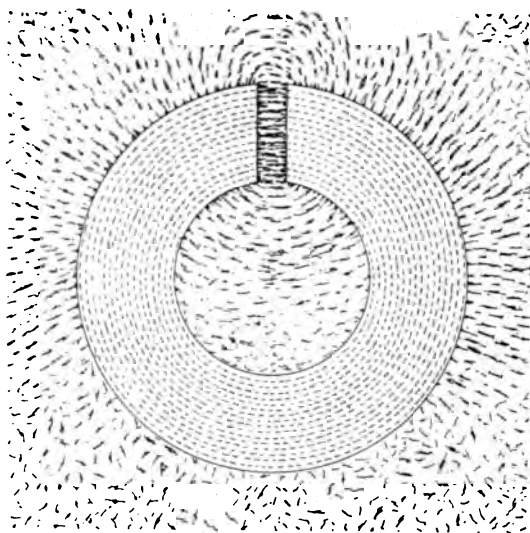


FIG. 72.—Magnetised Iron Ring with cut or air-gap. The ring is magnetised along the dotted lines. The external magnetic field or induction is delineated by iron filings.

field in the gap space. By making wider gaps, as shown in Figs. 73 and 74, the varying forms of the external fields can be detected by the use of the sprinkled filings.

We have already seen that, in order to produce an induced electromotive force, and therefore an electric current, in the secondary circuit, it is necessary to produce a change in the magnetic current or magnetic induction passing through that

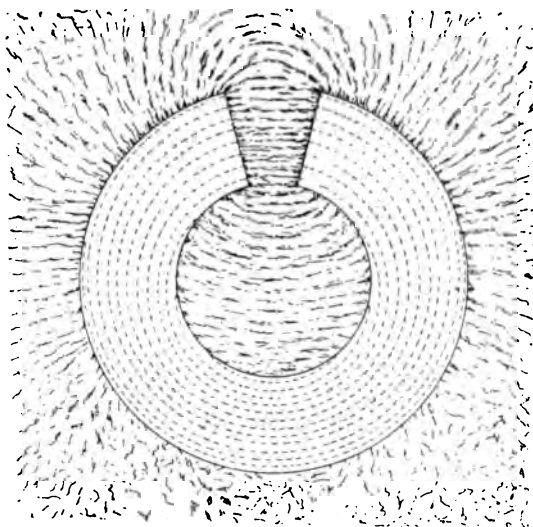


FIG. 73.—Magnetised Iron Ring, as in Fig. 72, but with wider air-gap.

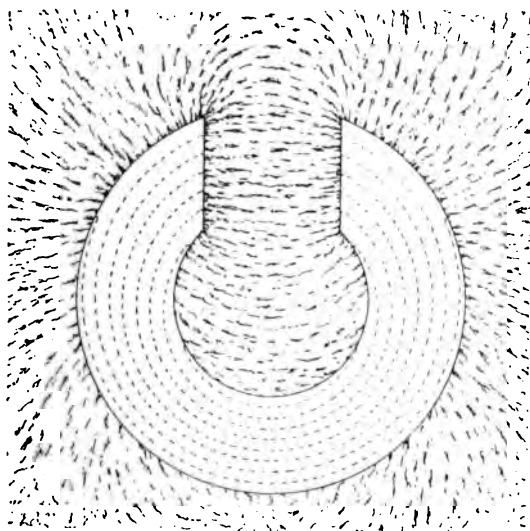


FIG. 74.—Magnetised Iron Ring, as in Fig. 72, but with still wider air-gap.

circuit. This change may be produced in two ways—either by stopping the primary current or by reversing its direction. In the second case the amount of change in the induction is doubled, because the reversing of the primary current amounts to first stopping it, and then starting it again in the opposite direction. Take the first case, viz., the complete iron ring, and imagine the direction of the primary current continually being changed or alternated. This would produce an alternating flow of magnetism round the ring, the direction of the magnetisation changing its direction with every change in the direction of the primary current. This changing magnetism passing through the secondary circuit would set up in the secondary circuit an alternating secondary electromotive force, and therefore, if the secondary circuit is completed, an alternating secondary current. It will be at once evident, however, that there are two ways in which we may set up this changing magnetism in that half of the ring on which the secondary circuit is wound. We may either employ the undivided ring, and keep continually changing the direction of the primary current in the primary circuit, or we may maintain the primary magnetising current constant in one half of the ring and turn round that half of the ring on which the secondary circuit is wound, so as to keep on reversing the direction of the magnetic current through the secondary circuit. A little consideration of these two modes of producing the changing induction through the secondary circuit will show that they really amount to the same thing.

In the light of the above remarks it will be found very easy to understand the general action of the alternate current transformer and dynamo. It has been explained in the previous Lectures that the power conveyed by a continuous electric current is measured by the product of the current strength measured in amperes and the difference of pressure or potential between the ends of the circuit measured in volts. The

same amount of power, say one horse-power, can, therefore, be conveyed, either in the form of a large current having a small fall in pressure or in the form of a small current having a large fall in pressure. We are familiar enough with this effect in the case of water. A waterfall may consist of a very large body of water falling down a very moderate height, such as the well-known falls of the Rhine at Neuhausen, in Switzerland; or it may present itself in the form of a much less quantity of water falling from a much greater height, like the falls of the Anio, at Tivoli, in Italy. In either case the power which the water would be capable of exerting, if utilised by means of turbines or waterwheels, would depend, not on the height of the fall nor on the quantity of the water flowing down, taken alone, but on the product of the numbers representing respectively the height and the quantity. The height of the fall determines the velocity or pressure of the water at the base of the fall. It is impossible to get any work out of water unless we are able to let it fall from one level to a lower one. Hence it is that stores of water at a height above the level of the sea represent available sources of energy.

The power which originally lifted the water up before it can fall must have been the evaporative power of the sun's heat. We commonly speak of using *water-power*; but, as a matter of fact, we are really employing in these cases *sun-power*. The falls of Niagara represent the accumulated water drainage of half a continent falling over a precipice of 150 feet in height on its way to the sea. This water, however, was originally evaporated from lakes, rivers, or the sea, before it could be precipitated as rain over Canada and North America. The power which lifted these water molecules from the surface of the Atlantic or Pacific Oceans into a position in which we are able to make them give back part of their energy of position was the radiation proceeding from the sun. The dynamos now being placed in

position to tap off some of the available power of the Falls of Niagara will in reality be driven by sun-heat, radiated to the earth at some previous time.

These hydraulic facts have an exact parallel in electrical science. A current of electricity can only do work by falling down from one electric pressure or potential to a lower one, and the rate at which it can do work is measured, not by the current strength or by the fall in potential alone, but by the product of the numbers representing these two quantities. A current of electricity of 10 amperes falling in pressure by 1,000 volts, represents the same power of doing work as a current of 1,000 amperes falling 10 volts in pressure. There is, however, a considerable difference between the two agents. The small current of high pressure is very different in many respects from the large current of low pressure. In the first place, the high-pressure current is more dangerous to handle or deal with, and requires much better insulation than the low-pressure current. On the other hand, when a current of electricity flows through a conductor a part at least of its energy is frittered away into heat by reason of the necessary resistance of the conductor. The amount so dissipated is, by Joule's law, measured by the product of the square of the current strength and the resistance of the conductor measured in ohms. Hence, if flowing in conductors of the same size in cross-section and length, a current of 1,000 amperes would generate 10,000 times more heat per second than a current of 10 amperes. Suppose, then, that we wish to convey a power of 10,000 watts to a distance, we can do it either by using a current of 10 amperes flowing out and back along electric mains between which 1,000 volts pressure is maintained, or by using a current of 100 amperes flowing in mains between which 100 volts pressure difference is preserved. If, however, we wish to dissipate or waste only the same fraction of our 10,000 watts power, say 10 per cent., in conveying it, we shall have,

in the case of low pressure current, to employ electric mains of 100 times less resistance in the second case than in the first. In other words, we must provide a conducting channel for the large current of 100 amperes which shall be 100 times the cross-section, and therefore 100 times the conductivity, of that which it would be necessary to lay down for conveying the small current of 10 amperes, with the same proportionate loss of energy. It will, therefore, be evident that, as far as mere cost of copper is concerned, it is cheaper to convey electric energy in the form of small electric currents at high pressure than as a large electric current at lower pressure. On the other hand, the high pressure conductors cost more to insulate, so the advantage is not simply proportionate to the economy in copper.

The point to which I wish now to direct your attention is that we have in the *transformer* an appliance which enables us to convert small currents of high pressure into large currents of low pressure. In so doing we do not create any energy—in fact, we waste some of it; but the transformer is a device which enables us to change the form of electrical power just as a pulley or other simple machine enables us to change the form of mechanical power. By the use of a pulley and tackle a man exerting a small force through a great distance can raise a very heavy weight through a smaller distance.

The alternate-current transformer is an appliance which operates on electric power just as a lever or pulley does on mechanical power. It consists of an iron ring or circuit of some form or other which is wound over with two wires, a thick wire and a thin one. These circuits are carefully insulated one from the other. The thin wire generally makes some ten or twenty times more turns round the iron ring than the thick one. If a small alternating current of elec-

tricity from a high pressure source is passed through the thin wire, it creates an alternating magnetism in the ring of iron, magnetising it first one way and then the other in a direction round the axis of the ring. This changing magnetism creates, as before explained, another alternating electric current in the thick-wire circuit, called a secondary current, and this current can be made to have a lower pressure and greater strength than the primary current by suitably proportioning the number of windings of the two circuits. Hence we can transform a small current produced by a pressure of 2,000

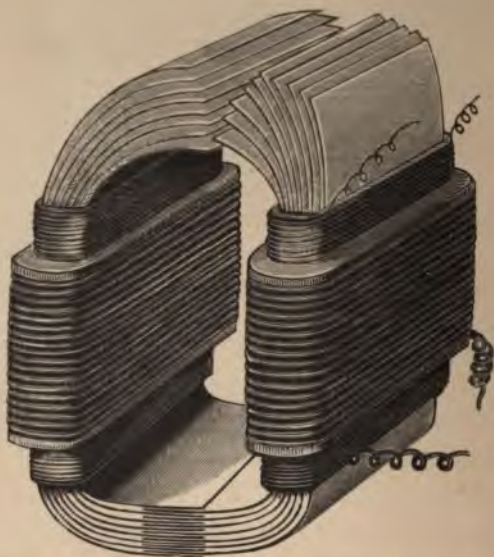


FIG. 75.—One form of an Alternate-Current Transformer in process of manufacture.

volts into a current nearly 20 times as strong, but having a pressure only of 100 volts. In this transformation there is a certain loss of power, or dissipation of energy, but in good transformers this will not exceed from 1 to 4 per cent. of the amount of power, transformed. We can, then, by means of

the transformer, carry electric power at high pressure and transform it down to a low pressure at the place where it is to be used. In transformer systems of electric supply this

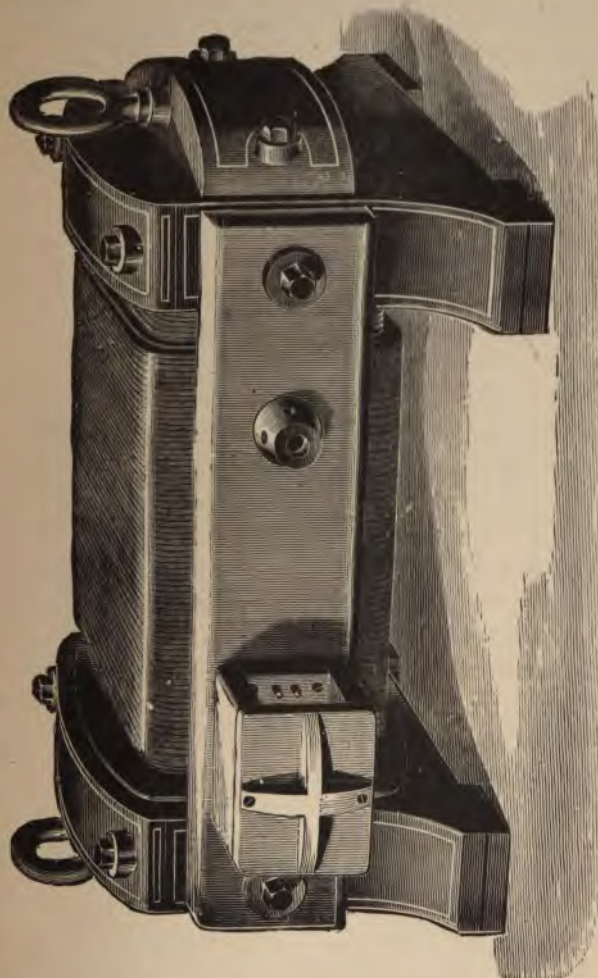


FIG. 76.—One form of an Alternate-Current Transformer, complete.

is the method adopted, as opposed to the other or direct-current supply, in which continuous currents flowing uniformly

engines are coupled direct to this flywheel, the cylinders of the engines being one on either side. In Figs. 82 and 83 are shown views of these 600 H.P. alternators with the direct-coupled engines. Smaller alternators of 250 H.P. are used during the daytime, or at a time when the demand for lighting is not great. These alternators generate an alternating current at a frequency of 40 and a pressure of 2,000 volts.



FIG. 84.—View of the Hydraulic Tower at Tivoli.

They send their current into three primary cables or feeders, which are concentric steel-armoured conductors, and these feeders deliver current to certain transformer centres and to transformers which are placed in the various buildings to be lighted. The total length of mains laid down is about 25 kilometres, or 16 miles. At the end of 1892 there was a total

the secondary circuit by making two cuts or air-gaps in the iron ring, and then rotating that half of the ring on which the secondary circuit is bound. This can, perhaps, be better understood by reference to another diagram (*see* Fig. 77). In this figure the shaded horse-shoe shaped part marked **M** is supposed to represent an iron core, which is magnetised by an electric current passed through a wire wound round that core, but which wire, for the sake of simplicity, is not shown on the diagram. This magnetic circuit is completed by placing another iron mass, represented in the figure by the shaded rectangle in between the poles or ends of the horse-shoe magnet. We thus obtain an iron circuit having two cuts or air-gaps in it, corresponding with the split ring shown in Fig. 70.

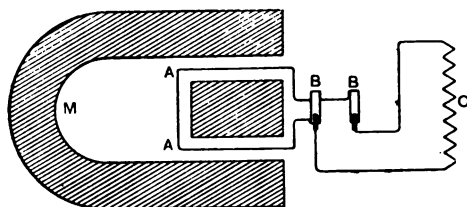


FIG. 77.—Skeleton Diagram, showing the principle of the Dynamo Machine

Suppose, now, a magnetic induction or magnetic current to exist in one direction round this air-iron circuit, and to be produced by a primary current flowing in a primary circuit of insulated copper wire which is wound on the horse-shoe shaped iron core. This magnetic induction or current will pass across the air-gaps, producing there a powerful magnetic field. Next, let us suppose that the iron block, or, as it is called, the armature core, is wound over with a circuit of wire indicated by the rectangle **A A** in Fig. 77. Let this core and armature wire be capable of rotating round an axis, and suppose that, by a couple of contact rings **B B**, we can always keep the ends of the armature circuit **A A** in conducting

connection with an external circuit **C**. If the rectangular core and its wire winding is made to revolve round its axis, it is easy to see that the direction of the magnetic current or induction through this secondary or armature circuit will be continually reversed, and hence that an induced secondary current will be created in it, which will also be continually reversed—that is, it will be an alternating current.

A dynamo, therefore, consists essentially of an iron core, or *field magnet*, as it is termed, which is wound over with a primary circuit or field wire circuit, and this circuit, when traversed by an electric current, creates a powerful magnetic field in the space between the poles or ends of the field magnet. In the interpolar space is placed an iron cylinder or ring, called the *armature core*, and this serves to complete the magnetic circuit. On this armature core is wound another coil or coils of wire called the *armature windings*, which correspond to the secondary circuit of the transformer. The armature core and its windings are rotated by being fixed on a spindle to which is usually attached a pulley. In Fig. 78 is shown a representation of an Edison-Hopkinson dynamo, in which the field magnets and armature are easily distinguished. In one large class of dynamos, called *alternate-current machines*, or *alternators*, the secondary current which is sent out from the secondary or armature circuit is, as observed above, an alternating current.

For many purposes, however, it is essential to produce a continuous current, or a current always in one direction. To obtain this, an additional organ has to be furnished to the dynamo. This is called a *commutator*, the function of which is to convert the alternating current produced in the armature coils into a continuous current in the external circuit. A little attentive consideration will, however, show that a dynamo and transformer are essentially

the same appliances in nature, only that in the latter the necessary reversal of the flow of magnetism through the secondary circuit is obtained by alternating or reversing the

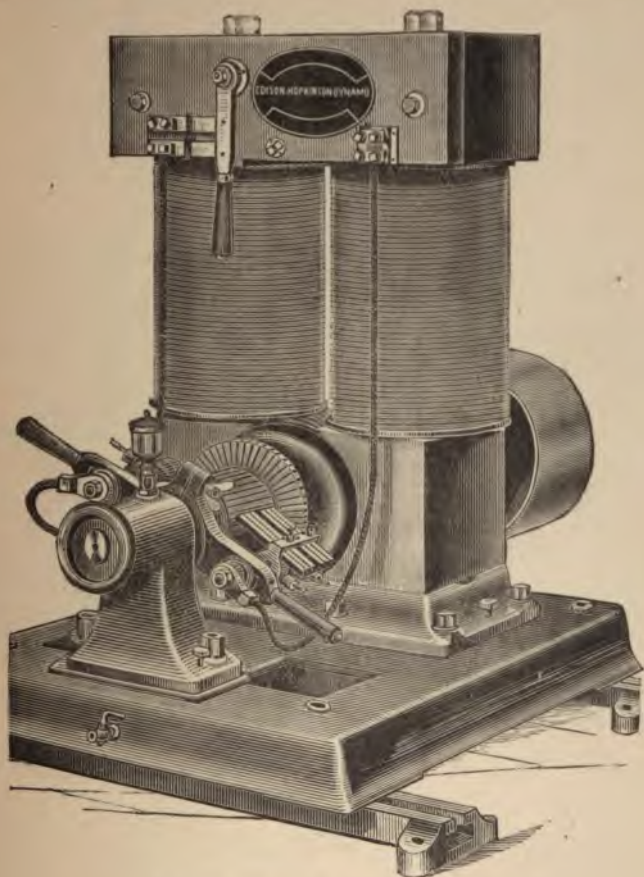


FIG. 78.—Edison-Hopkinson Dynamo.

direction of the primary current, whereas in the case of the dynamo the necessary reversal is obtained by rotating or turning round that part of the iron magnetic circuit on which

the secondary circuit is wound. This rotation of the armature of the dynamo may, in practice, be effected either by putting a pulley on the shaft and driving it by a belt or rope, or by coupling the dynamo shaft directly to the shaft of a high-speed steam engine.

One of the methods of direct driving, as this is called, most widely adopted in England is to employ a high-speed Willans engine, the shaft of which is continuous with that of the

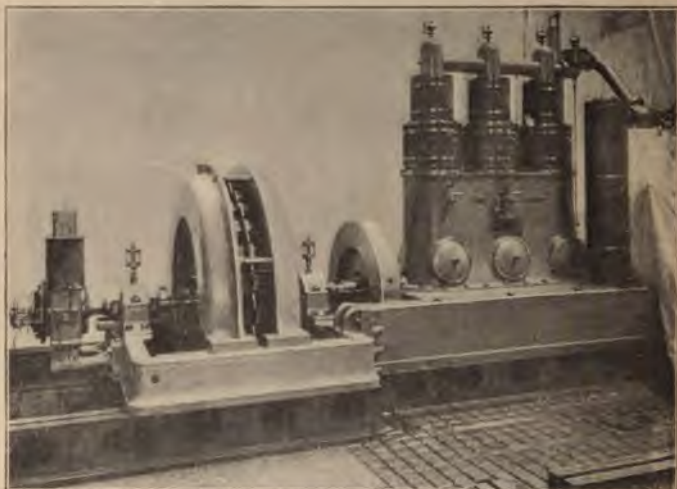


FIG. 79.—View of an Alternator direct driven by being coupled to a Willans Engine.

dynamo. In Fig. 79 is shown an illustration of such a combination. It is obvious that this arrangement is exceedingly economical of space, and hence has been widely adopted in the arrangement of electric generating stations, in which space is of importance.

/ It is impossible, as observed at the beginning of this Lecture, to do more than give here the briefest outline of the

different methods employed for distributing current for illuminating purposes; but the non-technical reader will probably be able to grasp most easily the nature of the various systems employed by my selecting one or two typical instances, and describing these somewhat in detail. At the present moment there are two principal systems of electric lighting, both of which have their advantages: these are called respectively the alternating current supply system and the continuous current supply system. In the first of these a generating station is established, which is provided with dynamo-electric machines called alternators, which are capable of producing an electric current alternating in direction a large number of times in a second. In the case of alternating currents the current flows in one direction in the circuit for a short fraction of a second, is then reversed, and then flows in the opposite direction for a short fraction of a second, and repeats the same cycle of operations continuously. The number of times per second which the current cycle is repeated—namely, a flow in one direction succeeded by a flow in the opposite direction, like the ebb and flow of the tide—is called the *frequency* of the electric current, and the frequencies most usually employed are either 40, 83, 100 or about 125. In England the practice generally is to use a frequency of 80 or 100, on the continent of Europe the lower frequency of 40, and in America the higher of 120. This alternating current is generated by machines called alternators at a high pressure, which is generally either 1,000, 2,000 or 2,400 volts, and in some cases 5,000 or 10,000. This high-pressure alternating current is then led out from the station by highly insulated conductors, which are called the primary mains, and these primary mains are generally made in what is called the concentric form. A stranded copper cable is covered over with insulating material, and then a circular strand of wire is plaited over the insulating material, and this again

due to the resistance of the line, which is about 800 ohms in the eighteen miles when working at full load. The cables, therefore, dissipate about 16 per cent. of the whole of the power capable of being transmitted at full load. The alternators have been so constructed that they can all work in parallel, sending their current into one or both pairs of cables as may be desired. On arriving at Rome the current enters a transformer house (*see* Fig. 88) placed just outside the Porta Pia of Rome. In this transformer house are placed a series of transformers, sixteen in number, each capable of transforming 80,000 watts from 5,000 to 2,000 volts. This transformer house is a substantially-built stone building two stories in height. The current, reduced from 5,000 to 2,000 volts, is then transmitted by underground cables through the streets of Rome to certain transformer centres, where it is again reduced in pressure to 100 volts. In addition to this, a portion of the current, at a pressure of 2,000 volts, is utilised for working a series of alternating current arc lamps. There are six circuits of these arc lamps, each circuit being arranged for 48 lamps. Each series of lamps takes a current of 14 amperes, and there is a special automatic pressure-regulator attached to the transformer supplying these arc light circuits, so that the current is kept perfectly constant, whatever may be the number of lamps upon the circuit. In addition to the arc light circuits, there are five or six incandescent lighting circuits consisting of Siemens concentric steel-armoured cables laid underground. The current supplied from Tivoli can be arranged to assist the Cerchi station, so that the two stations, 18 miles apart, one in the gasworks and the other at Tivoli, assist one another in furnishing current for incandescent and arc lighting in Rome as may be required. This Tivoli station is one of the finest examples of an alternating-current station operated by water-power, and it has been in perfectly successful operation since the year 1890. In order to protect the overhead line, going over 18 miles along the Campagna,

from damage from lightning there are lightning protectors of the kind described on page 145, placed at Tivoli, in the transformer house at Rome, and in the half-way house shown in Fig. 87. These protectors have proved sufficient to guard the dynamos and transformers from danger by lightning stroke.

The general arrangement of a low-pressure continuous current station on the three-wire system may be described as follows: In the station are placed a series of dynamos, each dynamo in most English practice being direct-coupled to a high-pressure compound steam engine. In Fig. 79 is shown

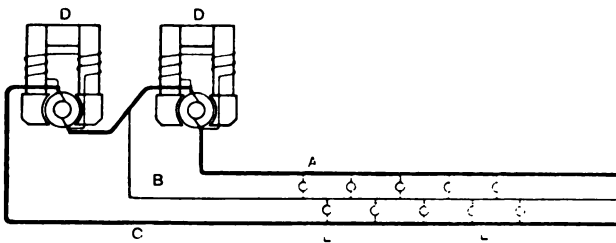


FIG. 89.—Diagram showing the arrangement of Dynamos *D D*, and Lamps *L* on the "Three-wire" System of Distribution.

such a steam dynamo, in which the armature shaft is coupled direct to the crank shaft of a Willans engine. This combination of engine and dynamo on the same bedplate is obviously very economical in floor space, and has therefore naturally become very popular in places where space is valuable. A series of these engine-dynamo sets is generally arranged in a central station, each set being either similar in size and appearance, or else in graduated sizes. In Fig. 91 is shown the interior of the St. Pancras electric lighting station, and in Fig. 92 the interior of the central station at Glasgow. In each of these the steam dynamos are arranged to generate current generally at either 110 volts or 220 volts, delivering this current to a switchboard. If the current is

supplied direct to the circuits, then the dynamos are joined in pairs between the three bars of the switchboard, as shown in Figs. 89 and 90. From the ends of these "omnibus bars," as they are called, are laid out a series of feeding mains or feeders, which feeders terminate in the distributing mains laid down in the streets of the district to be supplied. The circuits of the various houses are connected up to these triple distributing mains in such a manner that as nearly as possible one-half of the lamps are joined in between one pair of mains and the



FIG. 90.—Dynamoes in a Three-wire Low-Pressure Station.

other half between the other. In laying out such a district great discretion and knowledge have to be brought to bear in selecting points, which are called the feeder centres, so that at those points the pressure may be kept as constant as possible. Each dynamo in the central station may be regarded as a pump, pumping its current into the positive main omnibus bar, from which it flows out by the feeders to the distributing circuit and returns back again to the opposite main. During the hours of heavy demand for current there is, of course, a

large current flowing out by these feeders, and a correspondingly large fall of pressure down the feeder. Hence the pressure at the terminals of the dynamos has to be kept up in order to allow for this fall. It is usual to maintain the pressure at certain feeding points constant at 100 or 110 volts, whatever may be the nature of the supply, and to do this the pressure at the dynamo terminals has to be varied from 110 to 150 volts or thereabouts, as the demand varies at different hours of the day or night. It will thus be seen that in the feeders in a low-pressure station the loss of energy is greatest at the time of greatest load, because they are then traversed by the largest currents, and, therefore, the total supply of energy from the dynamos has to be equal to the amount required by the lamps in use plus that required to supply the corresponding loss in the feeders. A low-pressure continuous current station has, consequently, its least efficiency of distribution at the time of fullest load. Exactly the opposite is the case with the alternating current supply. In this case the efficiency of distribution is a maximum at the time of largest demand.

We may take as a good example of a low-pressure station the St. Pancras Vestry Electric Lighting Station, which has been carried out by the local authority of the parish of St. Pancras, in the north-west of London, for the supply of electrical energy for lighting and motive power purposes in that district. In this part of London, which is densely populated, it was considered advisable to adopt the continuous current low-pressure system, and a typical station of this kind was, therefore, designed for the district by Prof. Henry Robinson. The following is a description of the works as given by him: The station buildings consist of an engine-house 106ft. by 26ft., a boiler-house 53ft. by 35ft., a coal store 43ft. by 11ft., a battery room 40ft. by 14ft. 6in., as well as a testing-room, office, stores, and an underground

tank for condensing water capable of containing 170,000 gallons of water, and a chimney shaft 5ft. square inside and 90ft. high. The dynamo room contains eleven engines and dynamos, which are erected on a concrete foundation, surrounded by sand to prevent vibration being communicated to the walls. The floor is carried independently of the engine foundations by cantilevers from the walls. The engines employed are of the Willans compound



FIG. 91.—View of the Interior of St. Pancras Electric Lighting Station, London, showing the Dynamos.

central valve type, and the dynamos are of the 6-pole Kapp type made by Messrs. Johnson and Phillips. A view of the interior of this station, showing the long perspective of engines and dynamos, is given in Fig. 91. Each Willans engine is coupled direct to its corresponding dynamo. These dynamos furnish a continuous current, nine of them giving 680 amperes at a pressure varying from 112 to 130 volts, and three of them giving 145 volts, with a small current for charging secondary batteries at a distant sub-station. The

remaining two dynamos are wound for an output of 90 amperes at 540 to 575 volts. These are used for working the street arc lamps as well as for charging in series four sets of storage batteries at the central station which are capable of discharging at a rate of 60 to 75 amperes. The boiler plant consists of five Babcock-Wilcox boilers, each capable of evaporating over 5,000lb. of water per hour, the working pressure being 170lb. on the square inch. Along the top of these runs a steam main into which all the boilers deliver their steam and from which the engines separately take their steam. The steam, having done its work in the engines, is then either delivered into the atmosphere or passes into a jet condenser, which draws its water from the bottom of the underground tank. From the top of this tank the hot water is pumped by independent pumps to a cooling apparatus, which is capable of dealing with a minimum of 10,000 gallons of water per hour. The cooling arrangement consists of a large surface of corrugated sheet iron attached to a framing placed round the chimney. The water, after flowing along the corrugated surface, is collected underneath and returned to the bottom of the tank. There is also an air pump in the station for delivering dry air into the street main culverts at the rate of 5,000 cubic feet of air per hour.

The street lamps are Brockie-Pell arc lamps, placed on tall iron posts placed in the centre of the road where admissible. They are fixed at distances varying from 160ft. to 245ft. apart, the height of the lamp varying from 22ft. to 25ft. above the pavement. These lamps are worked 11 in series near the central station and 10 in series at a distance, and each lamp takes a current of 10 amperes, which is supplied by the two 500-volt dynamo machines. The main switchboard for the incandescent lighting is arranged on the "three-wire" system. Each dynamo is provided with a double-pole switch and copper coupling strips to connect the

machines to the third or middle wire and to the different bars on the feeder board on either side of the circuit. There is an amperemeter on each dynamo to indicate the current being given out. On the switchboard there are four positive omnibus bars and four negative bars, and at full load one machine can be run on to each bar and be worked at any pressure which may be necessary to serve the feeders which are switched on to it. Thus the dynamos may be worked at different pressures to suit the demand in the districts which they serve.

There are seven feeder mains going out of the station, in addition to a direct supply to the distributing mains. The mains are laid throughout the district on the three-wire system, the principal mains being sufficient to supply 25,000 incandescent lamps of 16-candle power in use simultaneously. These conductors are laid down under the streets, and are composed of copper strips $1\frac{1}{2}$ in. wide and $\frac{1}{8}$ in. thick, supported on edge in glazed porcelain insulators, which are carried on small cast-iron brackets built into the walls of the culverts. Some of the mains are cables laid in cast-iron pipes, and some of them are armoured cables. In addition to this, the current is supplied from the central station to a distant secondary battery station 1,140 yards away, which contains two batteries of 58 E.P.S. cells each. By varying the number of dynamos in use at the station and the number of feeders in connection with the omnibus bars, the engineers in charge of the station have it in their power to regulate the electric pressure between the distributing mains laid down in the streets, and the supply is so regulated as to keep the pressure between the middle main or wire and each of the outer wires to 110 volts.

A very similar station has been erected for the supply of electric current in Glasgow by the Glasgow Corporation. A view of part of the interior of this station is shown in Fig. 92.

The station buildings are substantially built, standing on a layer of concrete two feet thick laid throughout the entire block. These buildings consist of an engine and dynamo room, boiler room, workshop, stores, and offices. The boiler room contains six steel boilers of the marine type, measuring 12ft. by 10ft. in diameter, and working at a pressure of 160lb. Each contains two furnaces of the corrugated type, with an internal diameter of 3ft. These boilers deliver steam through a ring



FIG. 92.—View of Part of the Glasgow Corporation Electric Lighting Station.

main into the engine room. The engine room contains a series of Willans engines coupled to dynamos (*see* Fig. 92). The engines are Willans central valve compound engines, and the nine dynamos are two-pole shunt-wound machines with drum armatures. The boilers are fed with gas coke obtained from the Corporation gas works. Over the boiler house is placed an accumulator room containing 114 secondary battery cells. Each cell contains 61 lead plates, and has a capacity for storing up and discharging 1,000 ampere-hours

of electrical quantity. The current from the dynamos, which is generated at from 210 to 230 volts, is led to a distributing board in the station, and this supply has a three-wire system of distribution laid down in the streets. The three copper conductors consist of copper strip, which is laid on porcelain insulators contained in iron culverts. The current is brought to the two outer wires of the three-wire strip by conductors, which, as in the St. Pancras station before described, are called "feeders," and which consist of only two mains, a positive and a negative main. The lamps are all joined in parallel between the middle main of the distributing mains and one of the outer ones, the total of about 46,000 eight candle-power lamps being, as far as possible, connected so that one-half of the lamps are joined in between the middle main and one of the others, called the "positive," and the other half of the lamps being connected in between the middle main and the other conductor, called the "negative." The middle wire, therefore, serves to carry the balance of current from one part of the distributing system to the other at those times when the number of lamps joined in between the two sides of the three-wire system is not exactly equal.

These two stations may be taken as fairly typical examples of low-pressure stations as established in Great Britain, although many other equally good, such as the Kensington and Knightsbridge and Westminster stations in London, might be described. Such brief space as we are able to give to this portion of the subject in no way enables us to exhaust the description of the whole of the methods for a direct current low-pressure supply; but the general tendency, at any rate in England, is to adopt the direct-driven type of dynamo, on account of the considerable economy which is thereby effected in space. In some low-pressure stations, where space is not of much importance, it is found convenient

to drive the dynamos by means of belts from the flywheels of engines.

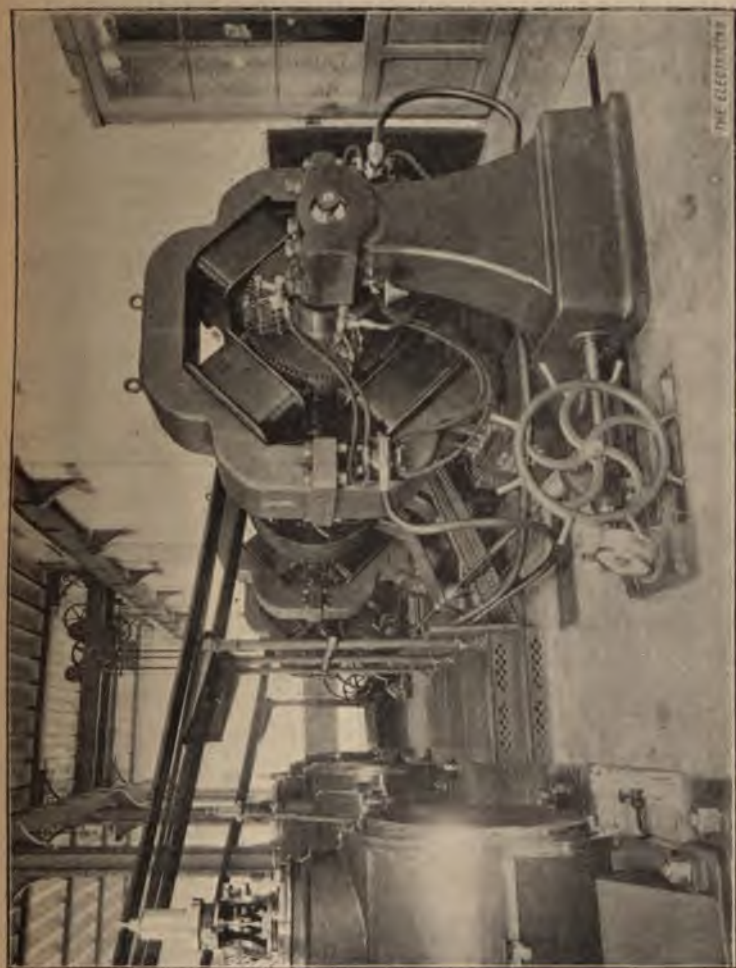


FIG. 93.—View of part of the interior of the Brussels Electric Station, showing the Dynamos.

In Fig. 93 is shown a portion of the interior of a Continental electric lighting station—that of the City of Brussels.

This station is constructed to contain plant of 3,000 indicated horse-power, the unit adopted being a 500 horse-power engine driving two 250 horse-power dynamos. Provision is made for placing six such units, one being for reserve. The present boiler plant consists of three Babcock-Wilcox water-tube boilers capable of evaporating 7,500lb. of water per hour. The engine room contains engines, at present two in number, of the horizontal compound condensing type. The low-pressure cylinders are 40 inches in diameter, and the high-pressure cylinders 26 inches in diameter, the length of the stroke being 4 ft. The speed is capable of variation between 62 and 75 revolutions per minute. Each engine is fitted with two grooved flywheels 20 feet in diameter, and weighing 20 tons. These drive the dynamos attached to each engine by means of ropes. The dynamos shown in the illustration are of the four-pole type, shunt-wound, with drum armatures, and with an output capacity of 145 kilowatts. These dynamos supply a three-wire distributing system, the cables being laid down underground in iron pipes.

In addition to the generating plant a storage plant of accumulators is provided, in two batteries of 70 cells each. Each battery is capable of giving a normal discharge of 350 amperes for ten hours. The underground cables consist of copper-stranded cables insulated with india-rubber, which are drawn into iron pipes. This work was designed and carried out by the engineers of the Silver-town Company.

The above descriptions of alternating current and continuous current generating stations will be sufficient to give the student a general view of the methods which are at present (1894) employed for the generation of electric current for illuminating purposes. It would lead us into matters too highly technical to attempt to discuss fully when, and

under what conditions, each system finds its best application. Suffice it to say that no one system can be described as the best system; both the low-pressure direct current and the high-pressure alternating current systems have certain peculiar advantages and disadvantages which have to be considered in the designs for a system of electric distribution. The engineer who is called upon at the present time to design and carry out a system of electric supply for public purposes has to consider a large number of facts which determine the choice that shall be made of the method of supply. Broadly speaking, the low-pressure continuous current system finds its best field in areas in which the density of the lighting is considerable.

Taking such cases as the central areas in large towns, it will be found that at the present time the demand for lighting will vary from one 8-c.p. lamp to three 8-c.p. lamps per yard of distributing main or of house frontage. In the outer portions of large towns and in smaller country ones the demand for light is much more scattered, and in these cases the only possible system of supply which meets the conditions is the alternating current system. It is possible to some extent to combine together the advantages of the two systems. The generating plant in the station can be made to furnish an alternating current at a low pressure. This can be used to supply the district within $\frac{3}{4}$ of a mile or a mile round the station with current furnished directly from the generating machines. Transformers can then be used to raise this pressure and transmit the current to the distant portions of the area of supply, and the pressure can then be reduced, either in sub-centres or in the houses, by means of transformers, to the normal pressure of 100 volts.

For a fuller discussion of the questions of electrical distribution we must refer those who desire to know more of these

matters to the many special treatises and text books in which the problems are handled in a thoroughly technical manner. The aim in these Lectures has been merely so to place the broad outlines of the subject before the general reader that the way may be prepared for more advanced instruction. No one at the present time can afford to be entirely ignorant of the principles which underlie the industrial uses of electricity. By the end of the nineteenth century all large towns in the world will be underlaid by a network of copper conductors distributing from house to house the electric current as an essential requisite of modern life. When in 1901 we celebrate the centenary of the birth of the galvanic battery, by which Volta in 1801 gave us the first practical means of producing this wonder-working agent, we shall witness the opening of an era in which the utilisation of electric current for the purposes of domestic life will have, to us, as yet uncomprehended consequences in serving to enhance the comfort and convenience of mankind.



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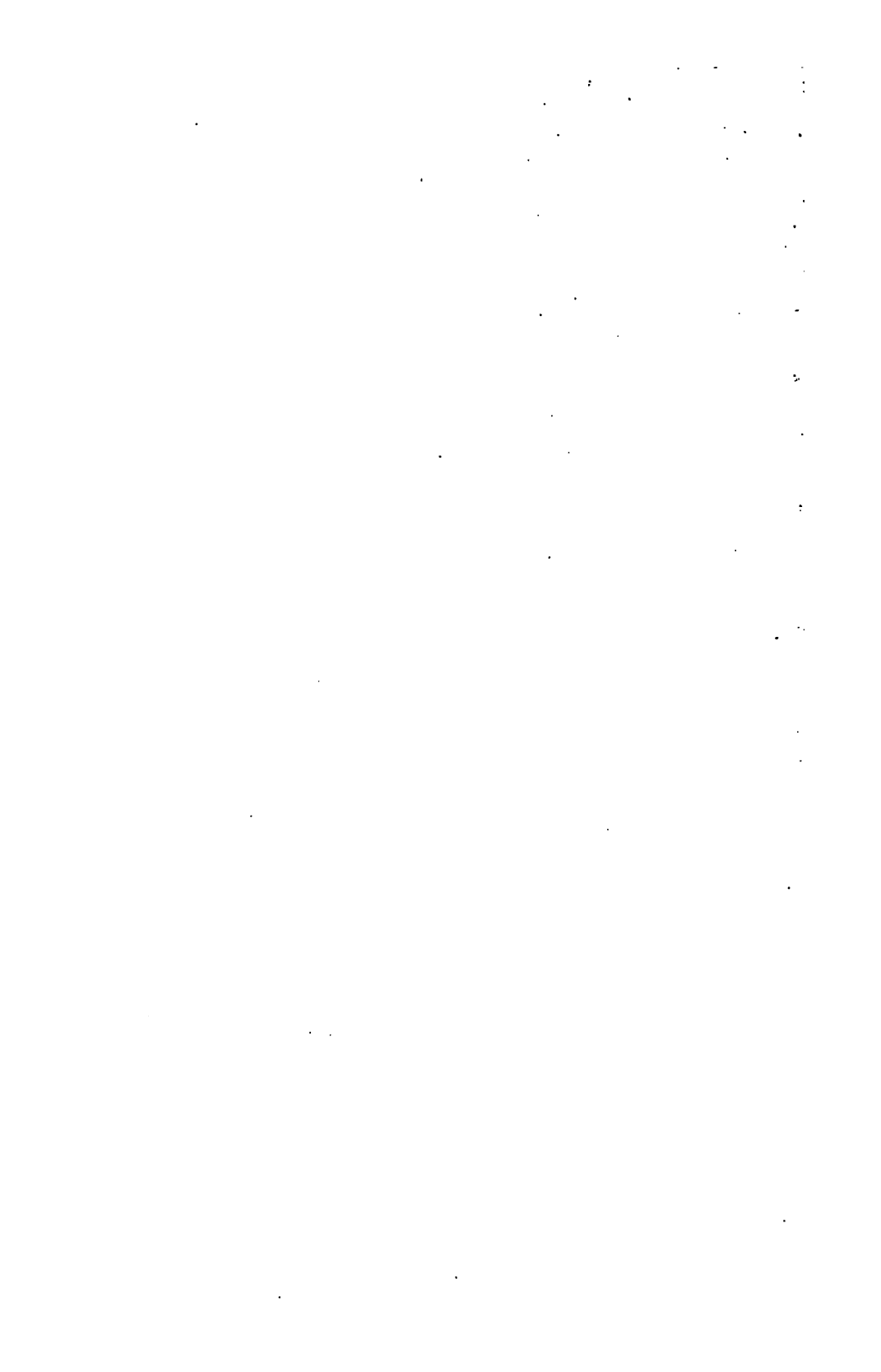
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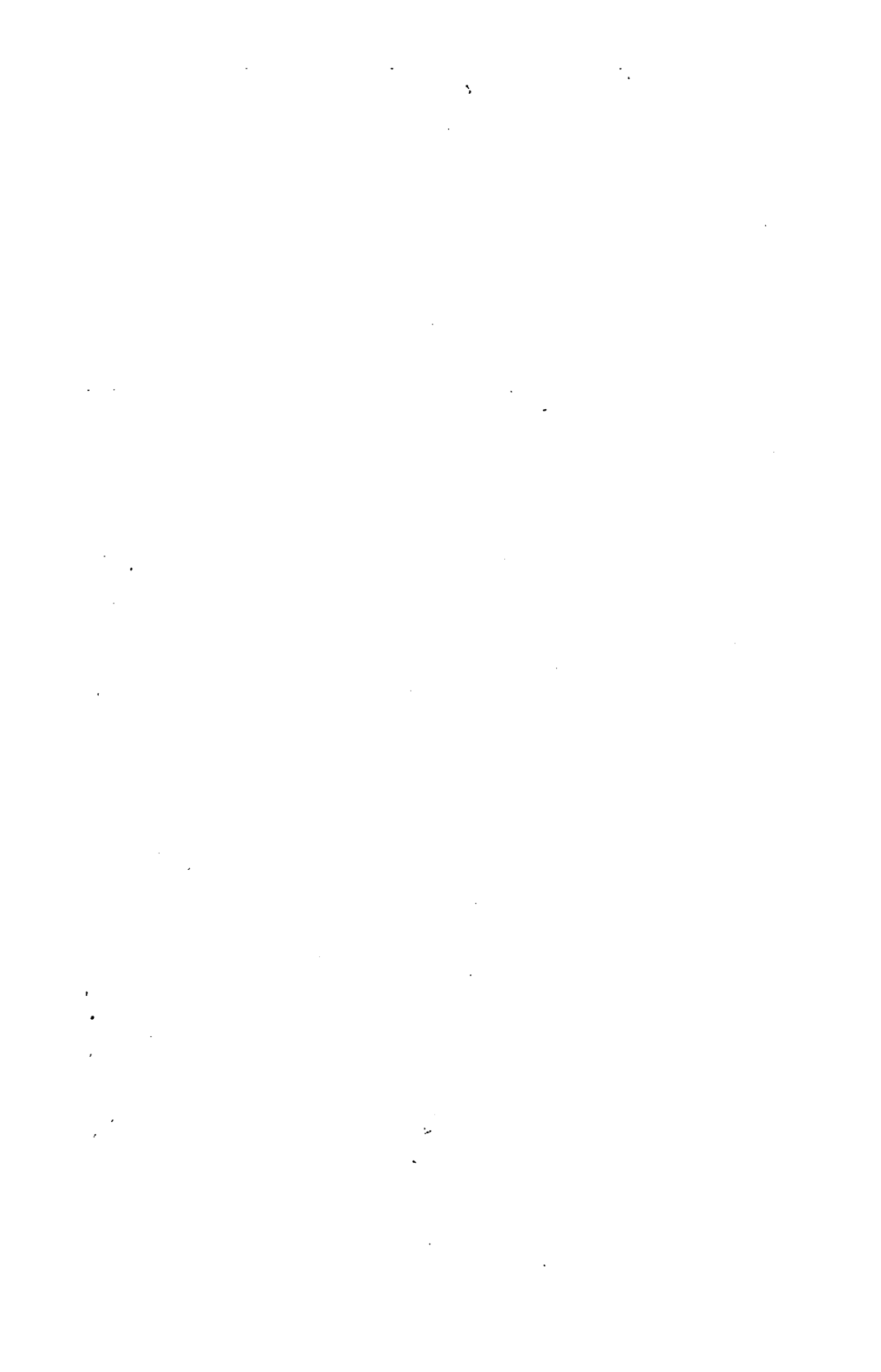
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